

# Emerging and Impactful Research Needs: Insights Regarding Sources

Why Indoor Chemistry Matters Workshop 2: Prioritizing Indoor Chemistry Research  
National Academy of Sciences

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# Motivation: major exposure location

- Time spent indoors
- Close proximity to source emissions we generate through activities
- Limited ventilation
- A complex and evolving mix of chemicals, from:

Unfortunately, one challenge is proprietary information about products

human activities  
consumer products  
building materials  
chemical reactions  
biological processes  
outdoor air

## Types of sources:

**Primary** - chemicals emitted directly (e.g., from cooking, volatile losses)

**Secondary** - emission of chemicals formed through indoor chemistry.

e.g., from oxidation of skin or surface films or gas phase

Sources to indoor air

Sources to outdoor air

# Chemicals with higher indoor concentrations have indoor sources

## VOCs:

TEAM Study

e.g., Wallace et al., *Atmos Environ.* 1985

NHANES

Su et al, *Atmos. Environ.*, 2012

RIOPA Study

Su et al., *Environ. Res.* 2013

# Chemicals with elevated indoor concentrations: WSOCg

## Water soluble organic gases

13 NC and NJ homes

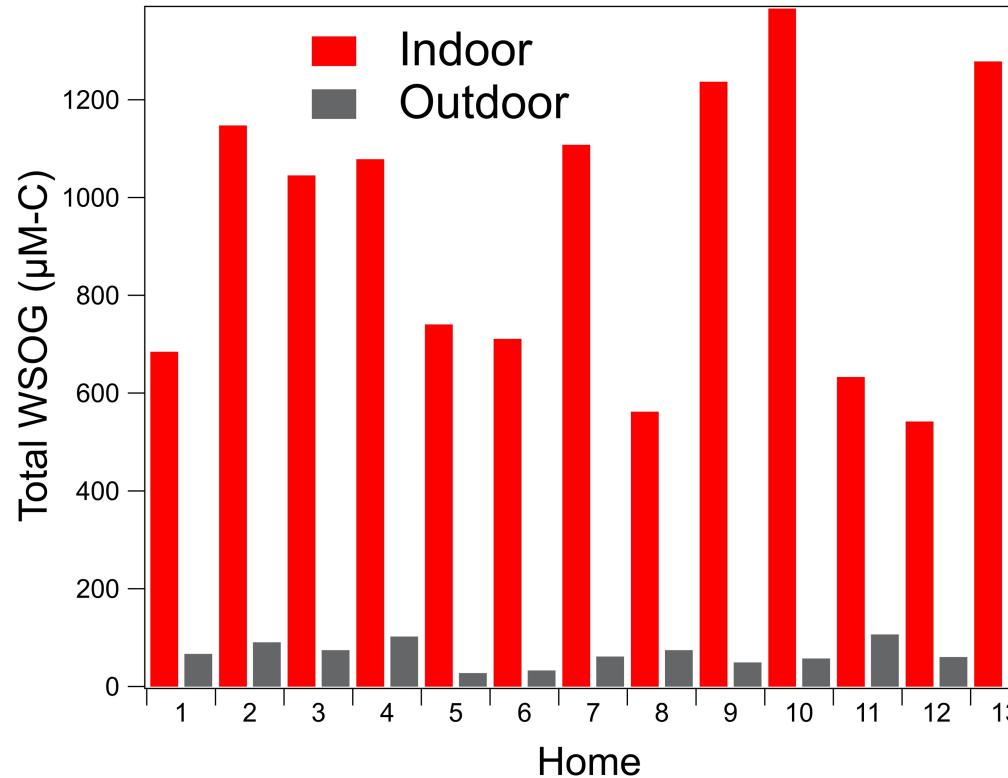
few measurements of this type. Value of mass balance

What chemicals?

Major sources?

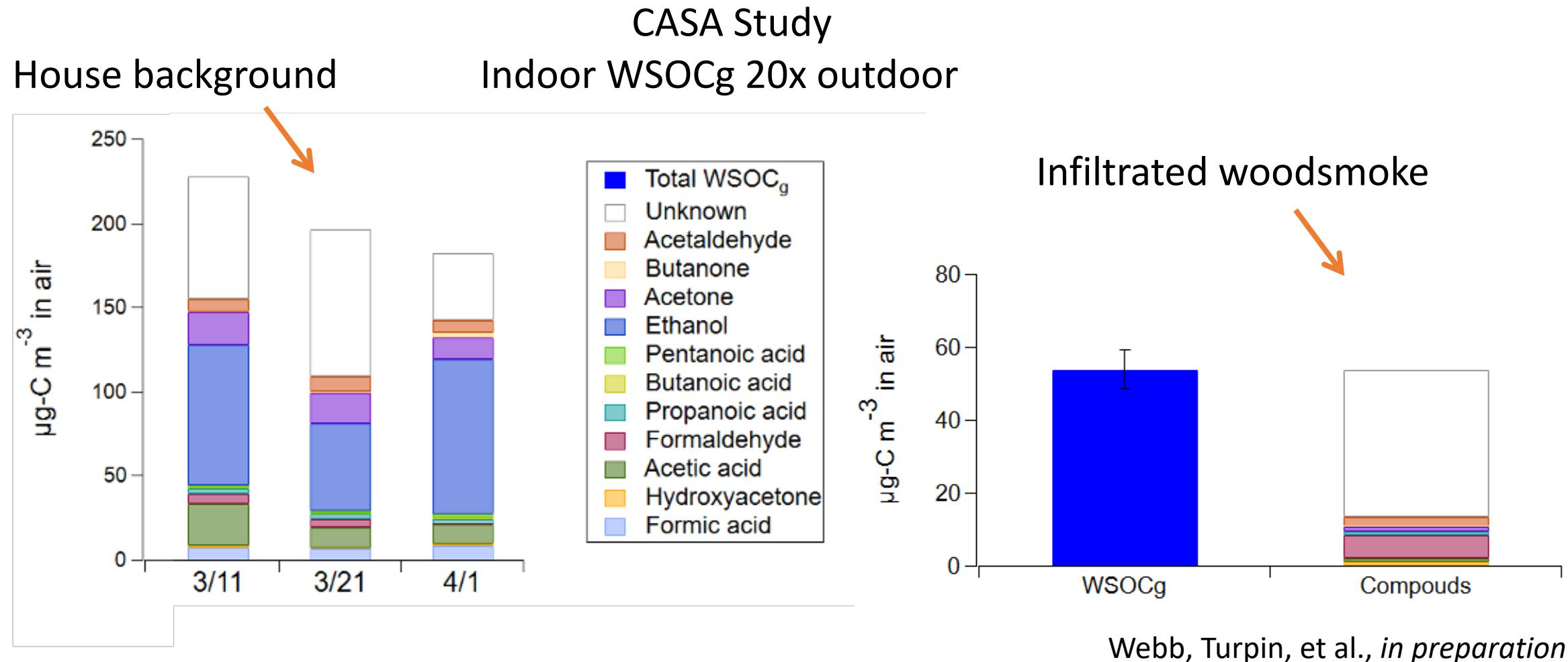
Primary or secondary (oxidation products)?

Total Water Soluble Organic Carbon (WSOCg)  
~13 times higher inside



Means: 145 vs 11  $\mu\text{g-C}/\text{m}^3$

# Value of species mass balance in identifying what is missing and do we understand the big picture

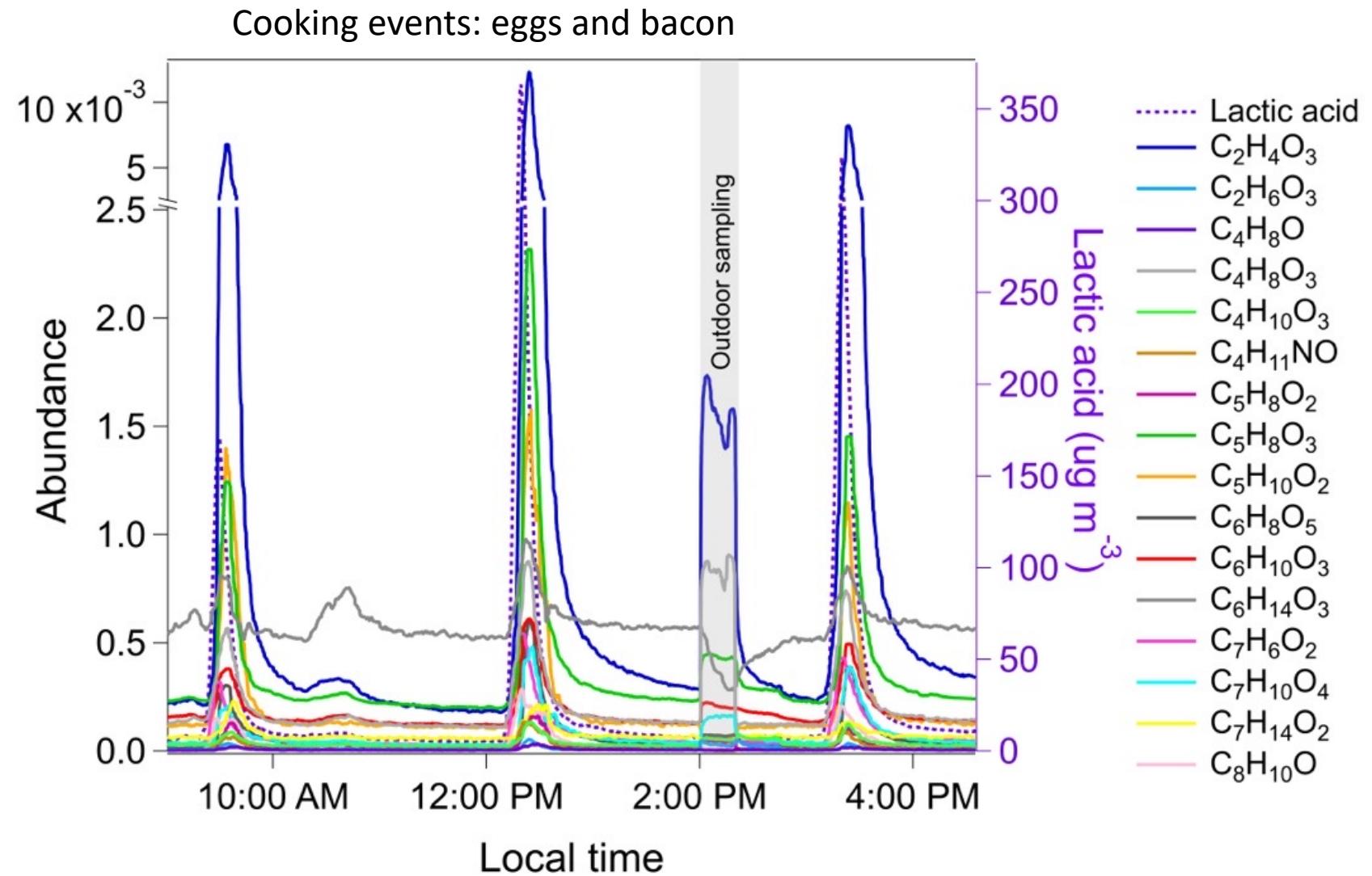


# Value of real time measurements in locating (identifying) sources

**Chemical  
Ionization  
Mass  
Spectrometry**

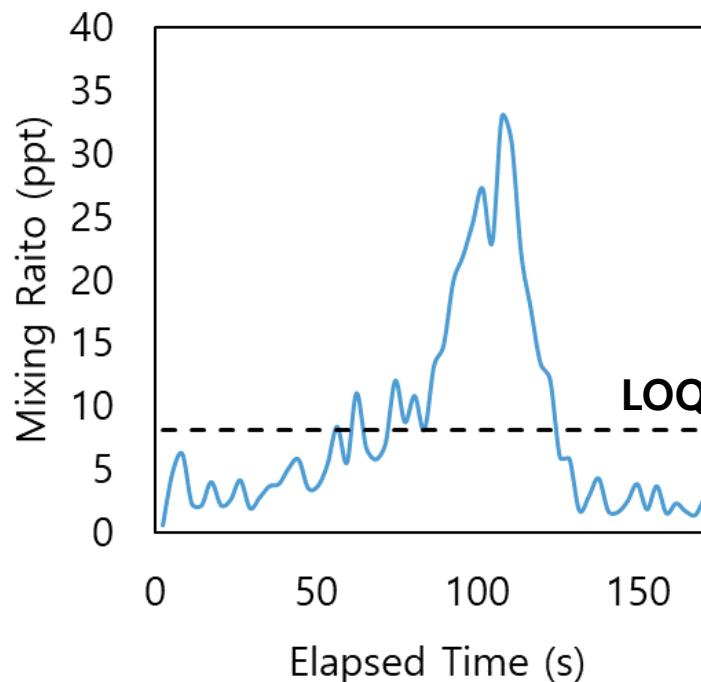
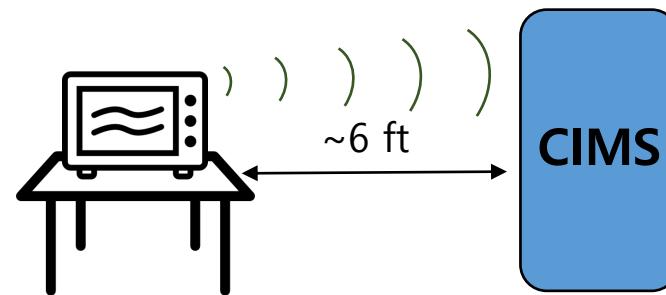
Small number  
of real-time  
indoor studies  
including:

- HomeChem
- CASA

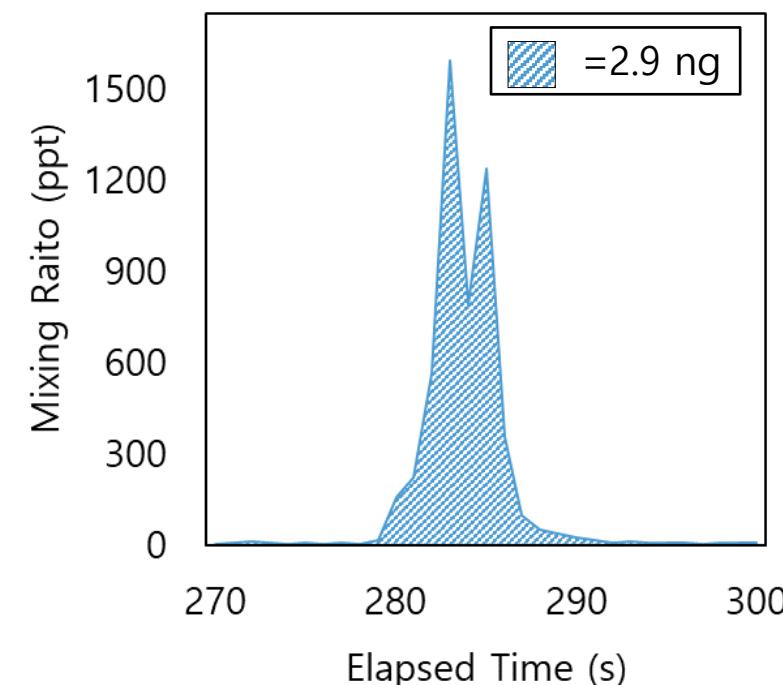
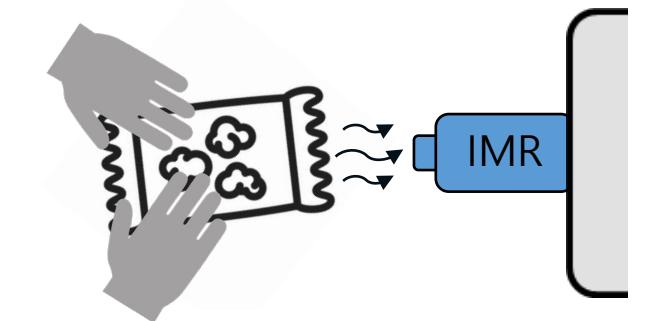


# Locating sources

Real time mass spectral methods promising for emissions measurements even for low concentration species



6:2 FTOH  
emitted from  
Microwaving  
Popcorn

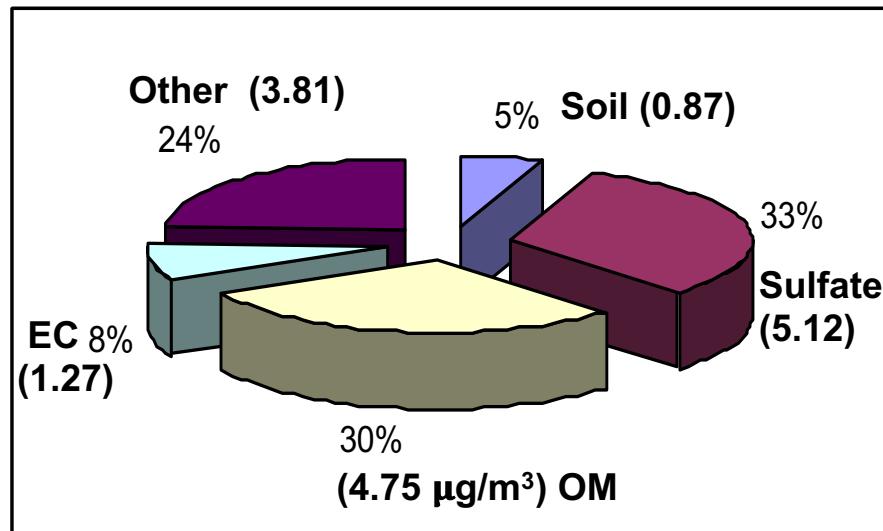


# Organic PM<sub>2.5</sub> twice as high indoors

RIOPA PM<sub>2.5</sub> Elizabeth, NJ, USA

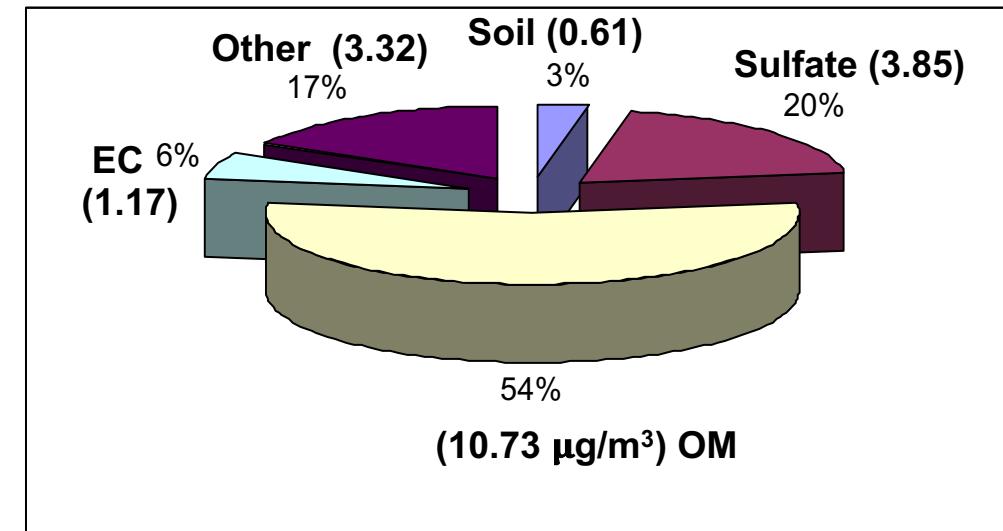
outdoor

Mass Conc. 15.82  $\mu\text{g}/\text{m}^3$



indoor

Mass Conc. 19.68  $\mu\text{g}/\text{m}^3$



Mass balance assumptions:

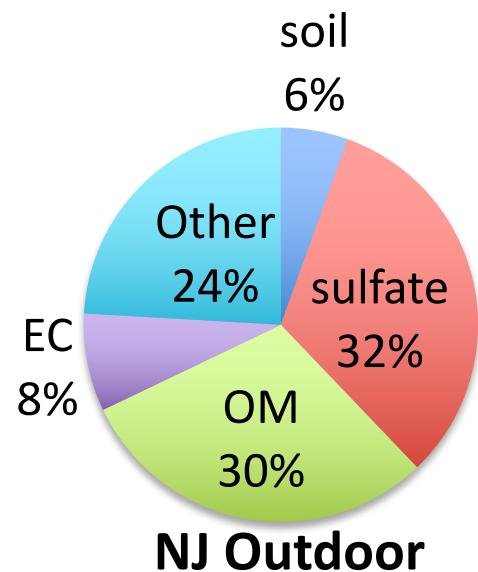
$\text{S} = (\text{NH}_4)_2\text{SO}_4$      $\text{OM} = 1.4 \times \text{OC}$

Soil = sum oxides

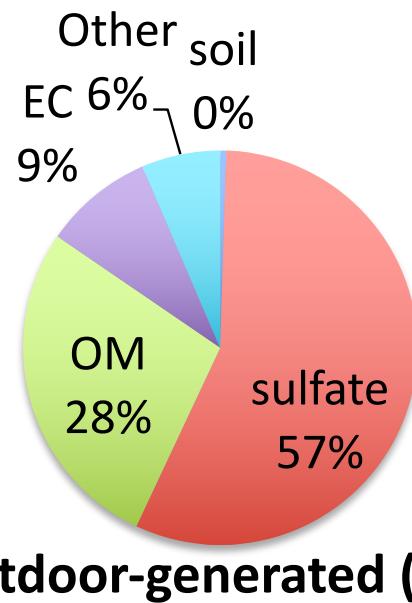
Polidori et al., *JESEE* 2006

# Indoor-generated PM<sub>2.5</sub> is largely organic

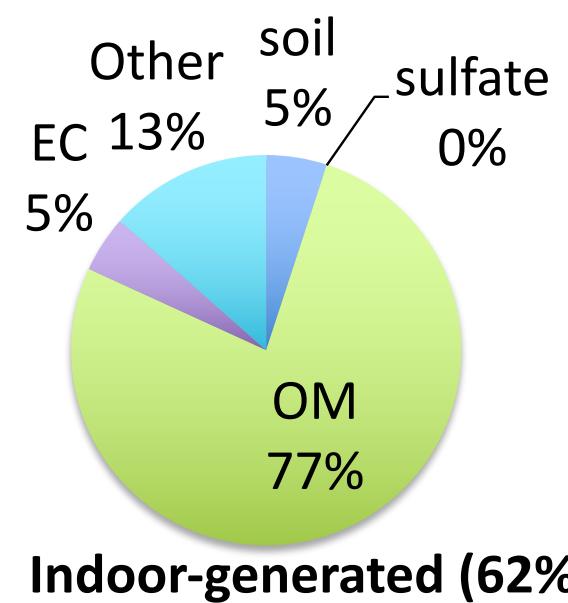
**NJ Outdoor (16  $\mu\text{g}/\text{m}^3$ )**



**NJ Indoor (20  $\mu\text{g}/\text{m}^3$ )**



Outdoor-generated (38%)



Indoor-generated (62%)

Derived from measurements of Polidori et al., *JESEE* 2006

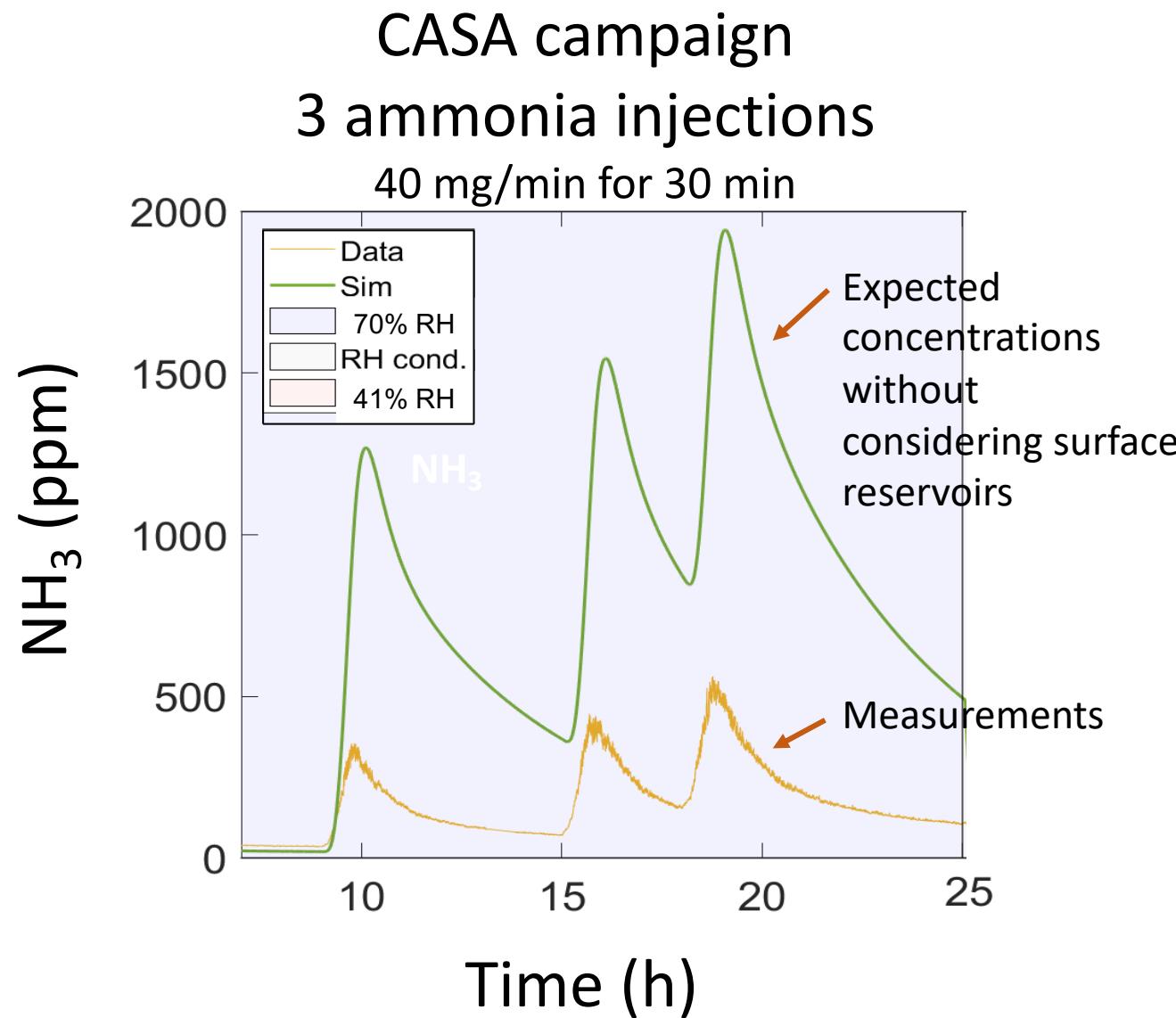
# Potential implications (example)

## Example:

Emitted organic PM could be a vehicle for transport of reactive or toxic chemicals, to the lower lung or to outdoor air.

- PAHs: Carbonaceous PM enhances partitioning of PAHs to particle phase  
Naumova et al., Atmos. Environ. 2003
- Does this effect exposure and dose of aerosol-associated ROS or toxic chemicals (e.g. PFAS) to the lungs or to outdoor air?

# Ammonia: indoor/outdoor > 10 in homes



Ampollini, DeCarlo et al (HomeChem) ES&T 2019

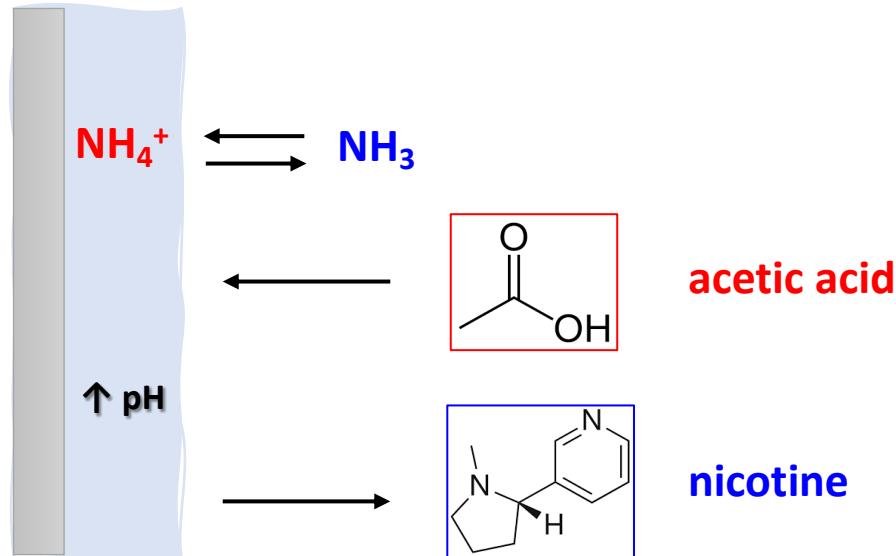
70-80% of injected  $\text{NH}_3$  was located in house reservoirs at the 24 h time point

Best fit model including surface reservoirs with reversible uptake:

Reservoir 50% larger at 70% RH than at 41% RH

Webb et al, in preparation

# Potential implications



- Indoors (30 to >1000 ppb); outdoors (1-5 ppb)
- Many known indoor sources (e.g. people, cleaning products, cooking)
- Water soluble base
- Increases pH in surface reservoirs
- Alters air concentrations of acids, bases

Ampollini, *ES&T* 2019,  
Ongwandee & Morrison, *ES&T* 2008; HealthyBuild 2006

Limited quantitative understanding of the impact of  $\text{NH}_3$  on indoor air

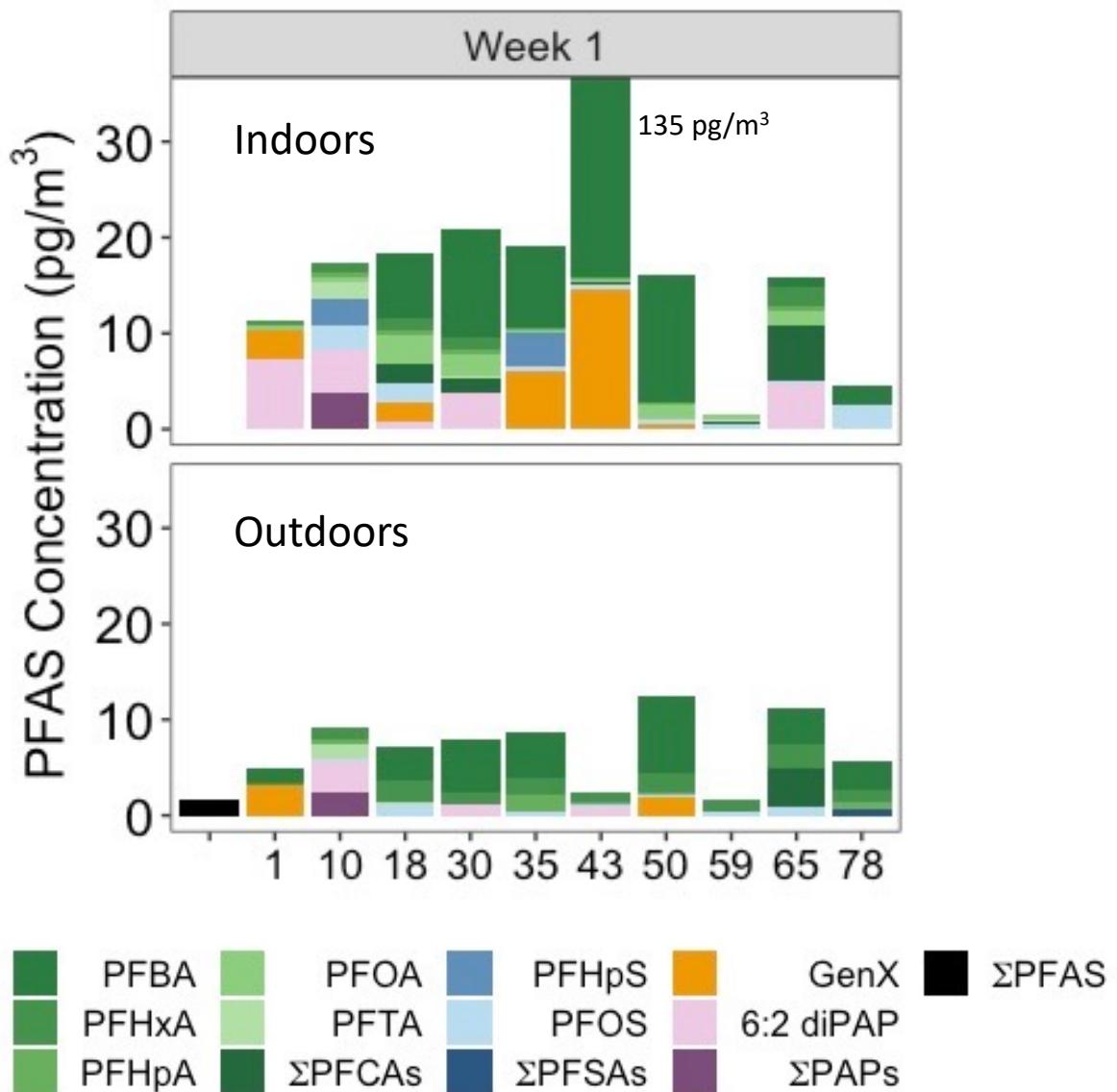
# Per- and polyfluoroalkyl substances are elevated indoors

## 26 Ionic PFAS:

- Residential indoor 2-3 times larger than residential outdoor
- order of magnitude higher than regional background

## 9 Neutral PFAS:

- Indoor concentrations 1000 times higher than for ionic PFAS



# PFAS: Many potential indoor sources, but which dominate?

Used because of water, stain, and heat resistant properties



CARPETS



CARPET CLEANING PRODUCTS



FOOD PACKAGING



FURNISHINGS



COSMETICS



OUTDOOR GEAR



CLOTHING



ADHESIVES AND SEALANTS



PROTECTIVE COATINGS



NON-STICK COOKWARE



CARSEATS



FIREFIGHTING FOAM

Photo Credit: <https://www.sixclasses.org/videos/pfas>

Challenges of evolving and proprietary product formulas,

Some of >10,000 PFAS species have been measured in many products

What are the major sources to indoor air?

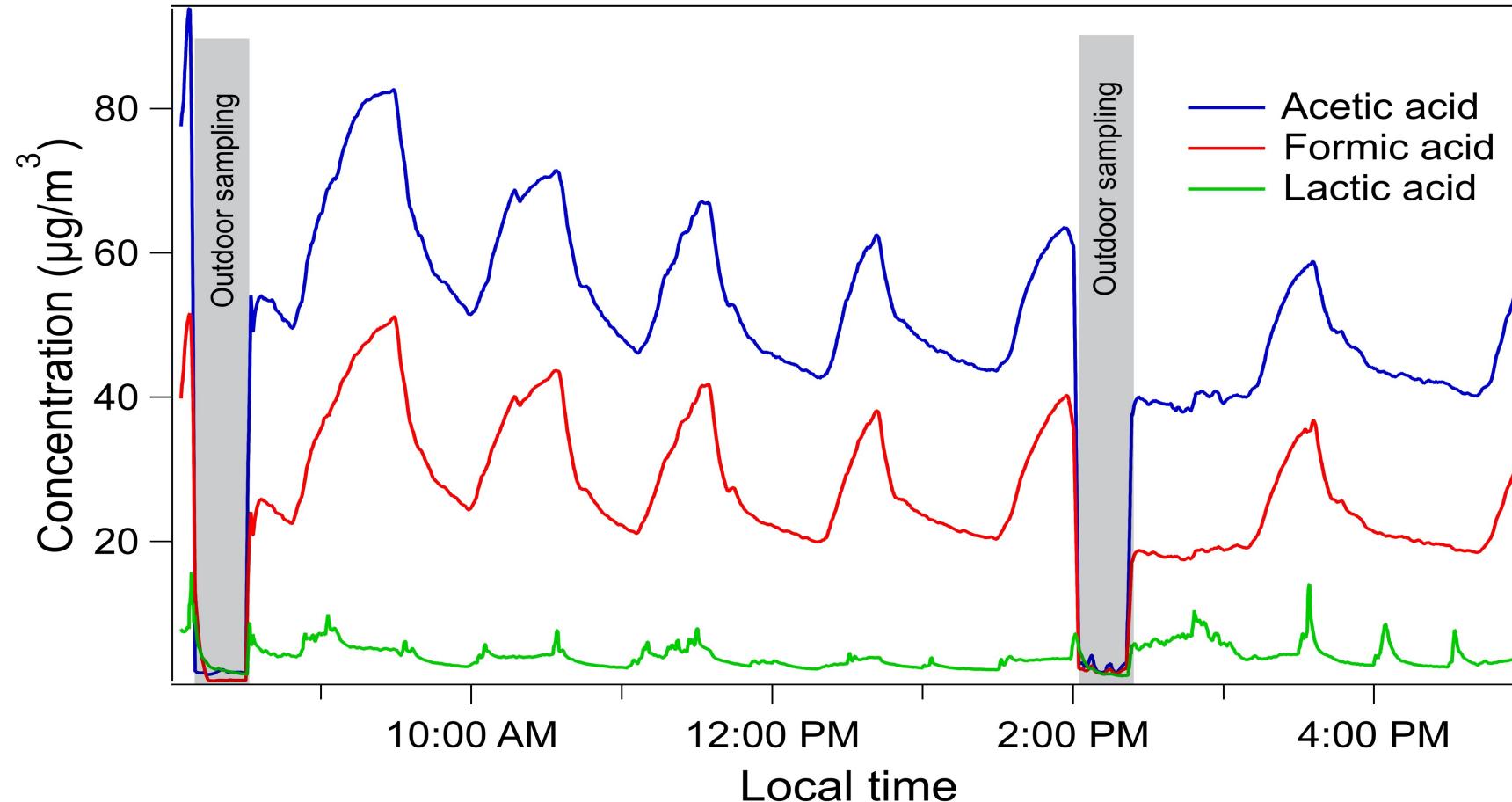
What influences partitioning?

What processes and pathways drive indoor concentrations, exposures and emissions to outdoor air?

Surfaces can have a large impact on air concentrations

Large loss of WS gases to AC system; Rapid rebound from building reservoirs

### Organic acids (CIMS)



Indoor surfaces, surface soiling and surface associated water provide sinks/reservoirs and also sources of primary/secondary

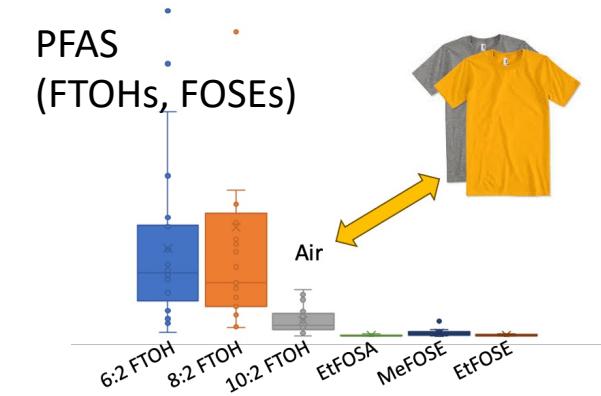
Higher surface-to-volume ratios indoors

Indoors:  $> 3 \text{ m}^2/\text{m}^3$

Outdoors:  $< 0.01 \text{ m}^2/\text{m}^3$

Singer et al. 2007

Morrison & Nazaroff, 2000



Eichler, et al., ES&T, 2023

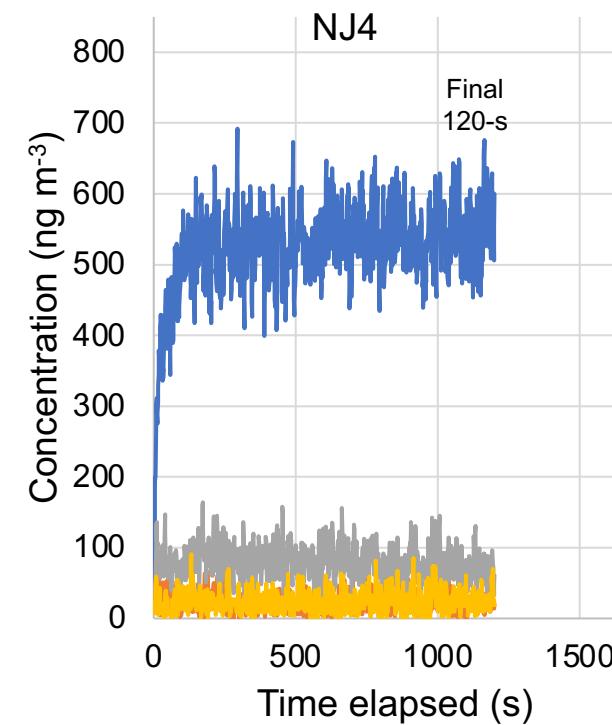
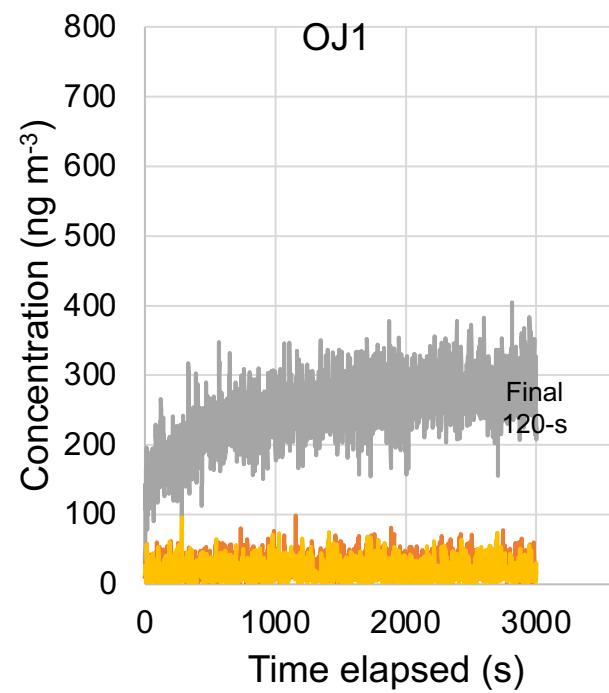
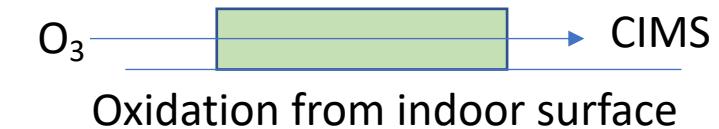
Surface reservoirs prolong residence time of chemicals indoors – time for reaction  
Strongest evidence for production (emission), is for ozone oxidation products

Why Indoor Chemistry Matters, NAS (2022)  
and references therein

Soiling/Water alter partitioning. Their role in chemistry is incompletely understood.  
Photosensitized reactions on soiled windows? Acid/base, hydrolysis, oxidation...  
**What are the secondary air emissions of volatile/semi-volatile products?**

# Potential for real-time mass spectral methods to measure emission rates with excellent sensitivity

Two Jackets  
FTOH emission rates in the 10s to 100s of  $\text{pg h}^{-1} \text{cm}^{-2}$



- 10:2 FTOH
- 8:2 FTOH
- 6:2 FTOH
- 4:2 FTOH

Emissions measurements feed models that test our understanding

# An iterative process

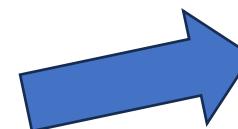


- Hypothesis-building field campaigns in real buildings
- Controlled experiments, perturbation experiments, emissions measurements with some level of authentic complexity
- Model development and model-driven measurements
- Field campaigns designed to test physical/chemical understanding in models

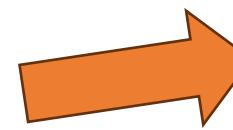
# Buildings are a source of chemicals to outdoor environment

Example: Volatile Chemical Products (VCP)

McDonald et al., *Science* 2018



Ventilation of 9 neutral PFAS



Ventilation of 26 ionic PFAS



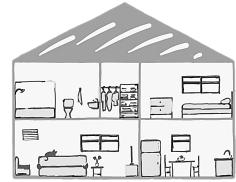
Dust removal (vacuuming)



Air exchange rates and cleaning/removal rates determined during the IPA Campaign

# PFAS emissions from homes by ventilation: neutral > ionic

One home



IPA Campaign: 10 NC homes

9 neutral PFAS

$\sim 39 \text{ mg yr}^{-1}$

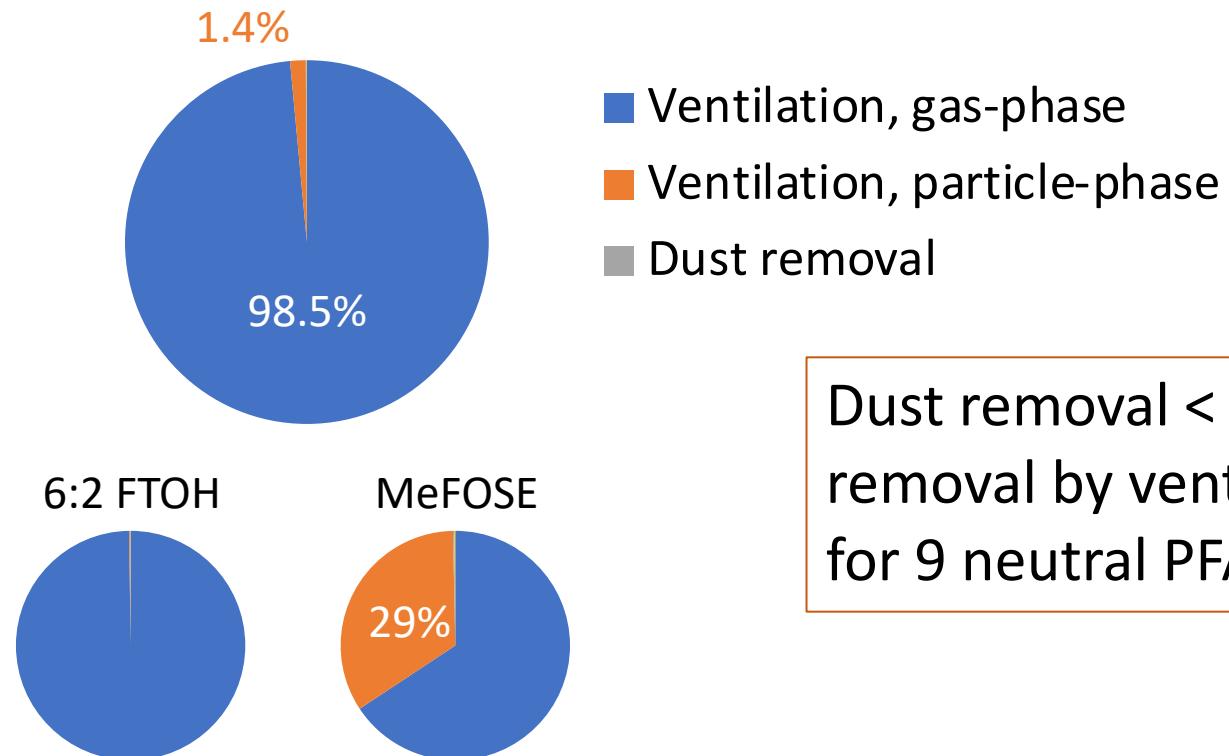
26 ionic PFAS

$\sim 0.02 \text{ mg yr}^{-1}$

Neutral PFAS are:

- byproducts of manufacturing
- rarely reported by manufacturing facilities.
- likely have emissions from consumer materials and products (based on above)
- thus, **could be a major largely unmeasured source of PFAS from manufacturing facilities**

## IPA Campaign: 10 NC homes



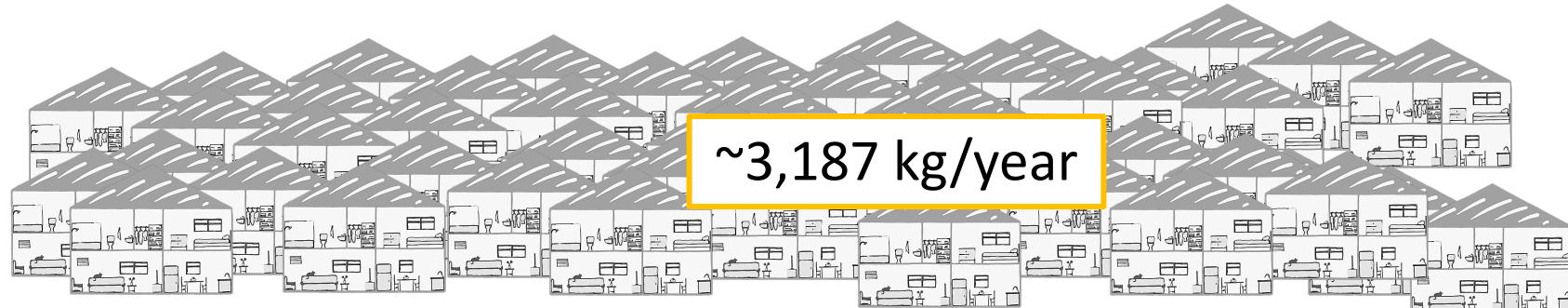
# PFAS emissions from single-family, detached homes

- In Raleigh metropolitan, NC: **~0.33 M** homes



**~13 kg/year**

- In the US: **~82 M**



**~3,187 kg/year**

- Only 9 out of >10,000 PFAS

- PFAS manufacturing: generally does not report these compounds

total reported : ~109,393 kg/year total (mostly short chain ionic PFAS);  
~19,700 kg/year long chain

a mass balance on total orgF would help understand the true impact

# Summarizing Opportunities and Needs

- Progress in understanding drivers of chemical dynamics
- Progress developing tools, methods
- Developing a quantitative, actionable understanding of indoor environment benefits from an iterative process
- Test understanding with models in realistic settings
- Identify of sources, emission rates to test our understanding with models under realistically-complex environments
- What are we missing? Mass balance? Surface-associated oxidants, Secondary sources?
- Impacts on outdoor air?

# Summarizing Opportunities and Needs

- An indoor challenge: complexity, heterogeneity
- An indoor advantage: can manipulate the system, while embracing complexity
- A challenge: proprietary and evolving chemical mixtures in products

# Extra Slides

# Indoor Challenges (homes)

**Every indoor environment is different**

And many buildings are exposure environments for only  
a few people

# Challenges: every home is different

Population-based sampling (sensors, passive, other)

Convenience sample – with purposeful variations

to understand determinants (RIOPA example)

Intensive sampling in single home, **benchmarking**

Purposeful manipulation, controlled experiments

Sampling for model testing

Medians and extremes

# Summarizing Opportunities

## **Every indoor environment is different**

Population-based sampling; benchmarking; identifying critical factors; median and outliers

## **Loss processes of comparable timescale**

Building manipulation to isolate a process  
Real time measurements

## **Chemistry**

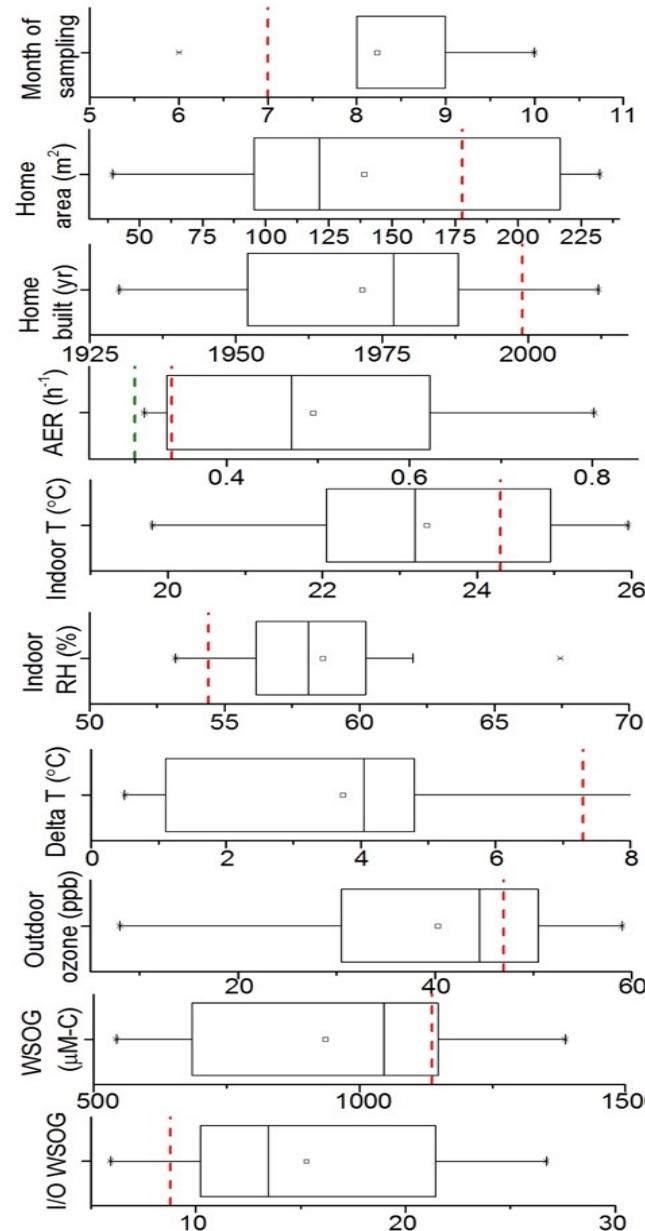
observations in authentic buildings  
hypotheses and model testing  
Controlled experiments (e.g. with authentic surfaces)

## **Characterization of indoor surfaces**

Continued characterization; thermodynamic modeling; surface/multiphase chemistry

# Benchmarking

N=13 NC/NJ  
homes



Sampling month

Floor area

Year built

AER ( $\text{h}^{-1}$ )

Indoor temp. ( $^{\circ}\text{C}$ )

Indoor RH (%)

Indoor-outdoor temp. difference

Outdoor ozone (ppb)

WSOG ( $\mu\text{M-C}$ )

Indoor/outdoor WSOG