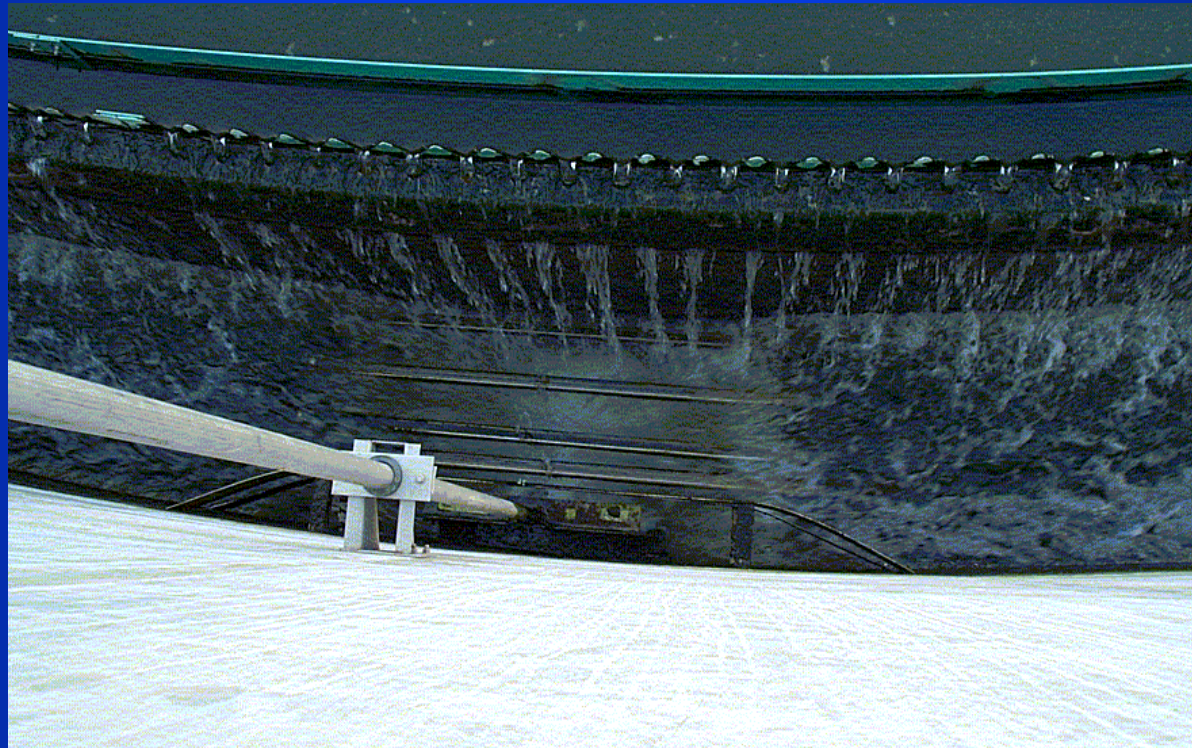


Occurrence and Fate of PPCPs in Wastewater Treatment Processes

Joan Oppenheimer, M.S.P.H., B.C.E.S. (Stantec)



ACKNOWLEDGMENTS

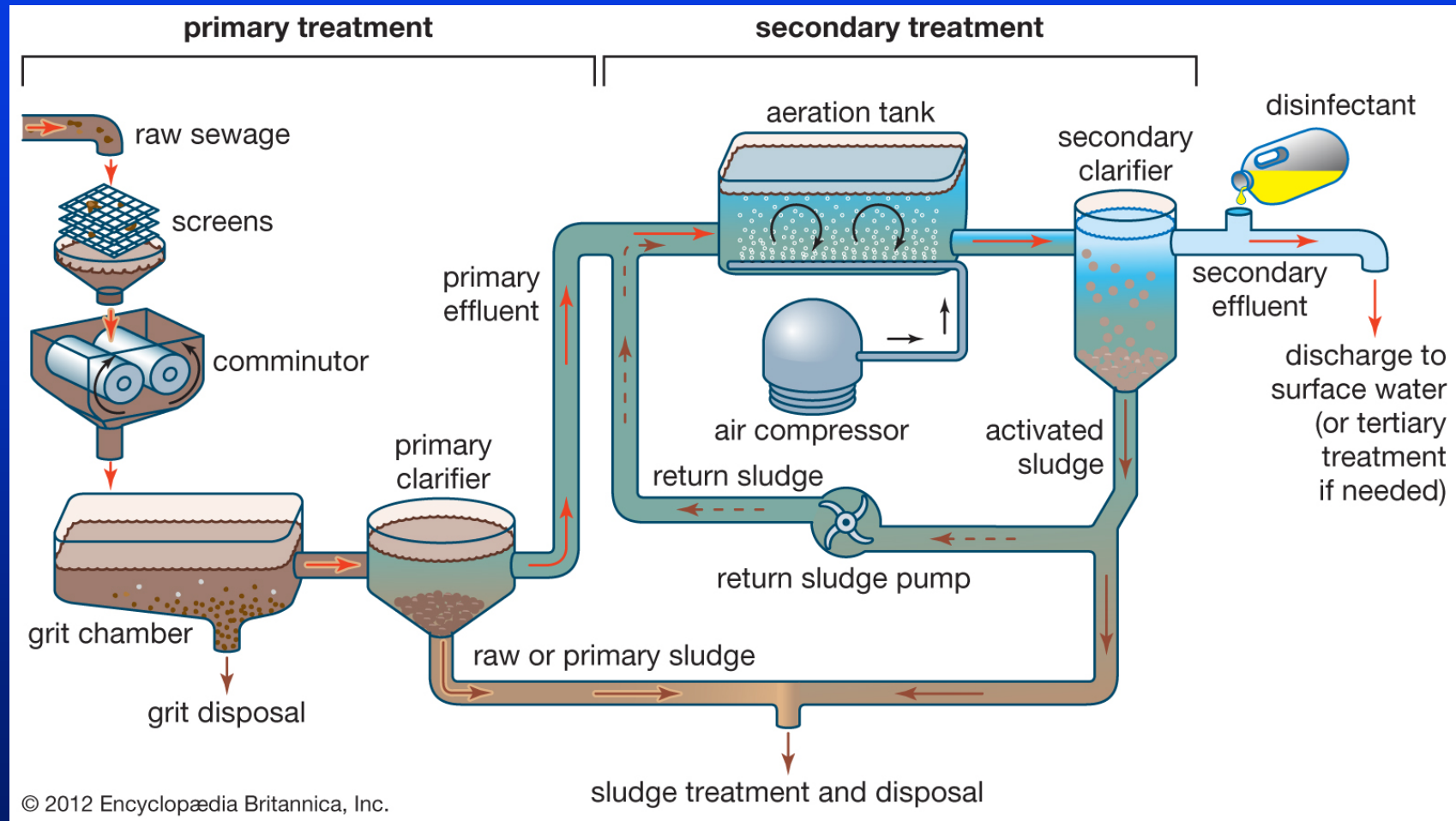
- **Water Environment Research Foundation (WERF 03-CTS-22UR)**
“Fate of PPCPs through Municipal Wastewater Treatment Processes”
(Co-PI : Dr. Roger Stephenson, Stantec)
- **WaterReuse Foundation (WRF-06-007)**
“Investigation of Membrane Bioreactor Effluent Water Quality and Technology”
(Co-Sponsors – U.S. Bureau of Reclamation & Global Water Research Coalition)
(Co-PI : Dr. Bruce Rittmann, Arizona State University)

WERF Study

Goals and Objectives

- Survey occurrence/passage of PPCPs through soluble phase of WWTP facilities
 - 3 discrete sampling events @ 6 facilities
- Compare PPCP passage through activated sludge plant and MBR pilot process

Aqueous Wastewater Treatment



Process Descriptions of Participating Treatment Facilities

Facility	Primary Treatment	Secondary Treatment	Secondary Aeration	MLSS (mg/L)	SRT (days)	Filters	Disinfection
1	Polymer Ferric	Activated Sludge	Pure Oxygen	1,300 to 2,600	0.5 - 1.5	None	None
2	No Chemicals	MLE Nit/Denit	Diffused Air	1,800 to 2,000	3 - 5	Granular Deep Bed	Chlorine
3	No Chemicals	Activated Sludge	Diffused Air	2,000 to 3,000	4 - 6	Granular Deep Bed	UV
4	No Chemicals	MLE Nit/Denit	Diffused Air	2,500 to 3,000	7 - 20	Granular Deep Bed + MF/RO	Chlorine
5	No Chemicals	Nit/Denit and Pilot MBR	Diffused Air	2,100	11 - 16	None	UV
6	None	Extended Aeration Nit/Denit	Surface Air	4,000	20 - 30	Granular Deep Bed	UV

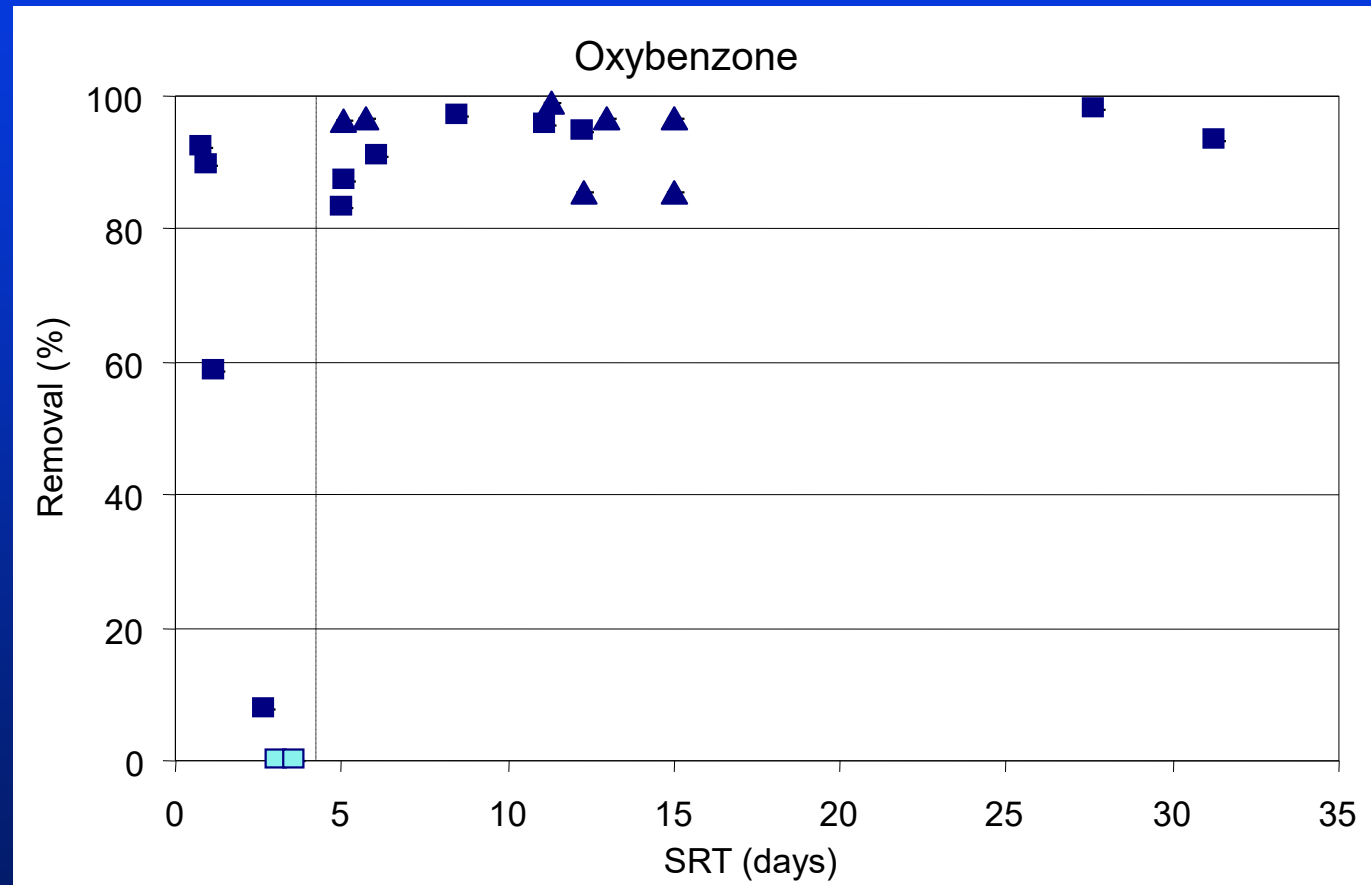
PPCP Occurrence & Treatment Bins: Activated Sludge Process

Treatment Occurrence	Bin T1 Good Removal	Bin T2 Moderate Removal	Bin T3 Poor Removal
Bin O1 Infrequent	(Methyl-3-phenyl-propionate)	Octylphenol	TCEP (Triphenyl-phosphate)
Bin O2 Intermediate		(Ethyl-3-phenyl-propionate)	BHA DEET Musk Ketone
Bin O3 Frequent	9 compounds	Triclosan Benzophenone	Galoxolide

PPCP Occurrence & Treatment Bins – Activated Sludge Process

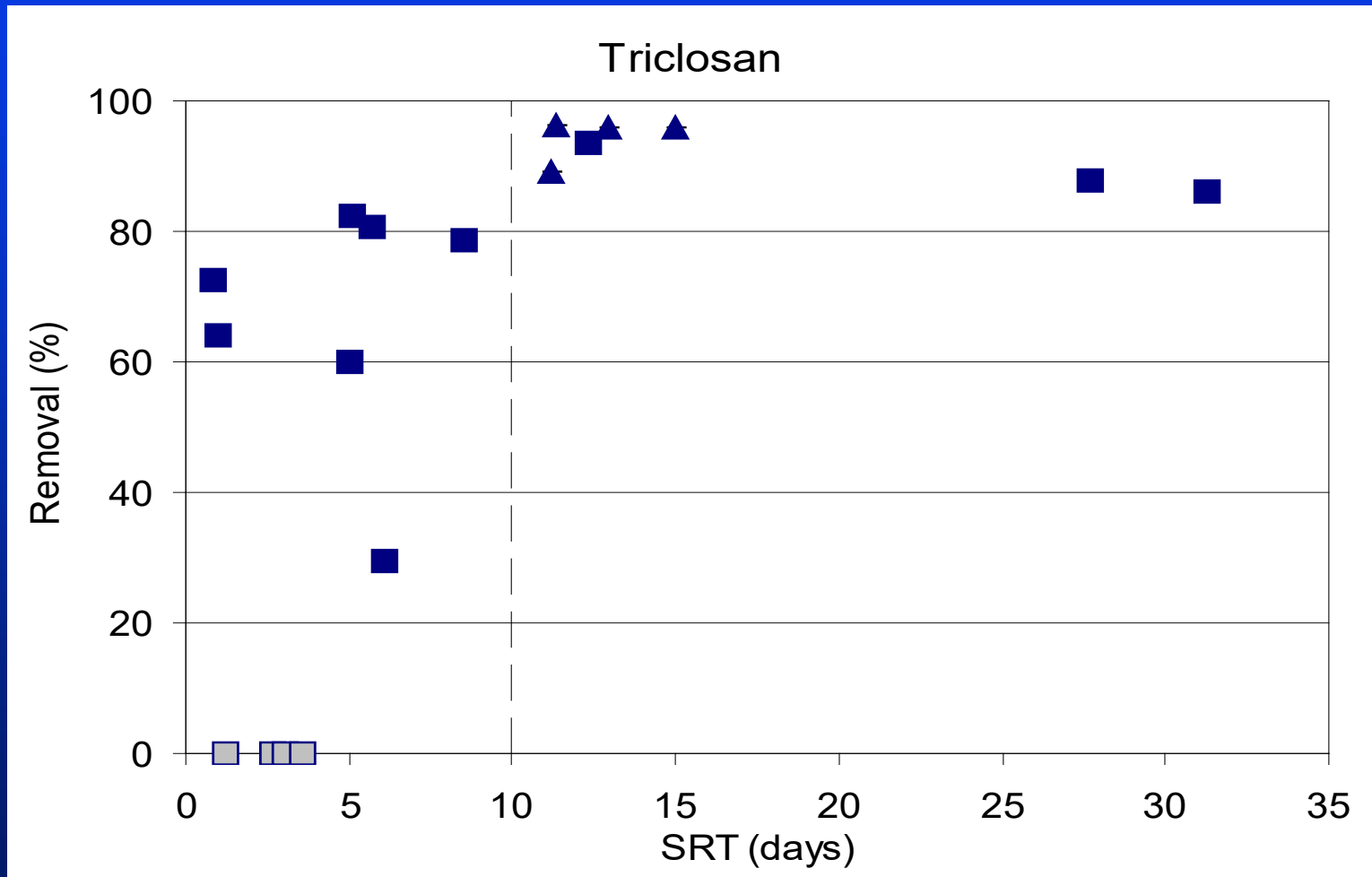
Treatment Occurrence	Bin T1 Good Removal	Bin T2 Moderate Removal	Bin T3 Poor Removal
Bin O1 Infrequent	$SRT_{80} \leq 5$ Days •Caffeine •Ibuprofen •Oxybenzone •Chloroxylenol •Methylparaben •Benzyl salicylate •3-phenylpropionate •Butylbenzylphthalate •Octylmethoxycinnamate		
Bin O2 Intermediate			
Bin O3 Frequent			

SRT for Oxybenzone (Bin T1)



■ = Actual Removal ▲ = Removal greater than percentage value □ = Effluent > Influent

SRT for Triclosan (Bin T2)

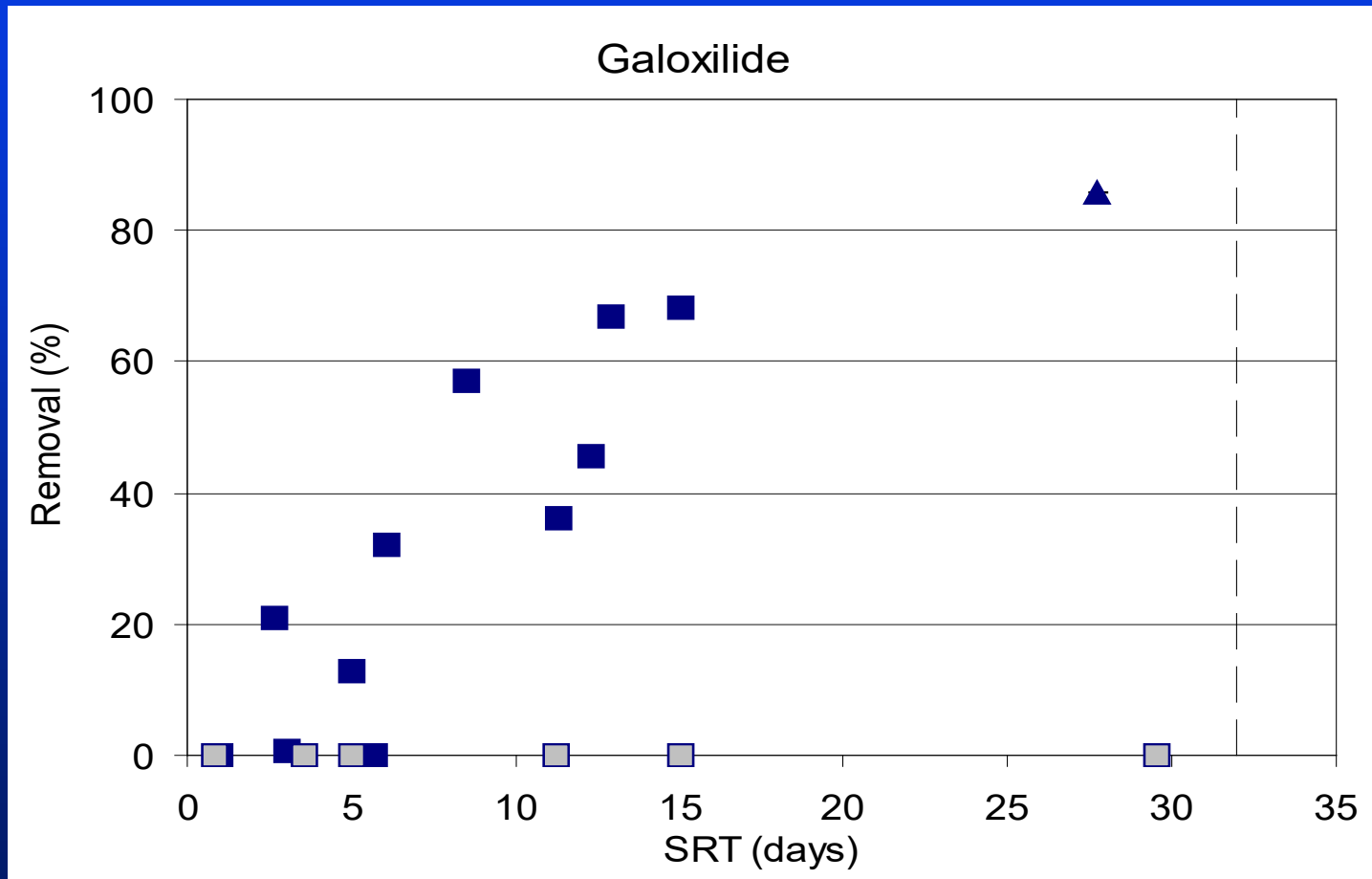


■ = Actual Removal

▲ = Removal greater than percentage value

□ = Effluent > Influent

SRT for Galoxolide (Bin T3)



■ = Actual Removal

▲ = Removal greater than percentage value

□ = Effluent > Influent

Longer SRT₈₀ Compounds

- SRT >8 Days: BHA
- SRT 10 Days: Triclosan
- SRT 13 Days: Benzophenone
- SRT >13 Days: DEET
- SRT > 15 Days: $\left[\begin{array}{l} \text{Galaxolide} \\ \text{Musk Ketone} \\ \text{TCEP} \end{array} \right]$

Media Filtration / RO

	Plant F	Plant D
Chloroxylenol	-27%	ND
Methylparaben	>97%	ND
DEET	26%	0.050
Benzophenone	-88%	ND
TCEP	-17%	ND
Galoxilide	-18%	ND
Musk Ketone	12%	ND
Oxybenzone	0%	ND
Triclosan	15%	ND

WERF Project Conclusions

- SRT dependence demonstrated for PPCPs

$SRT_{80} \leq 5$ days for Bin T1 compounds (10)

$5 \leq SRT_{80} \leq 15$ days for Bin T2 compounds (2)

$SRT_{80} \geq 15$ days for Bin T3 compounds (6)

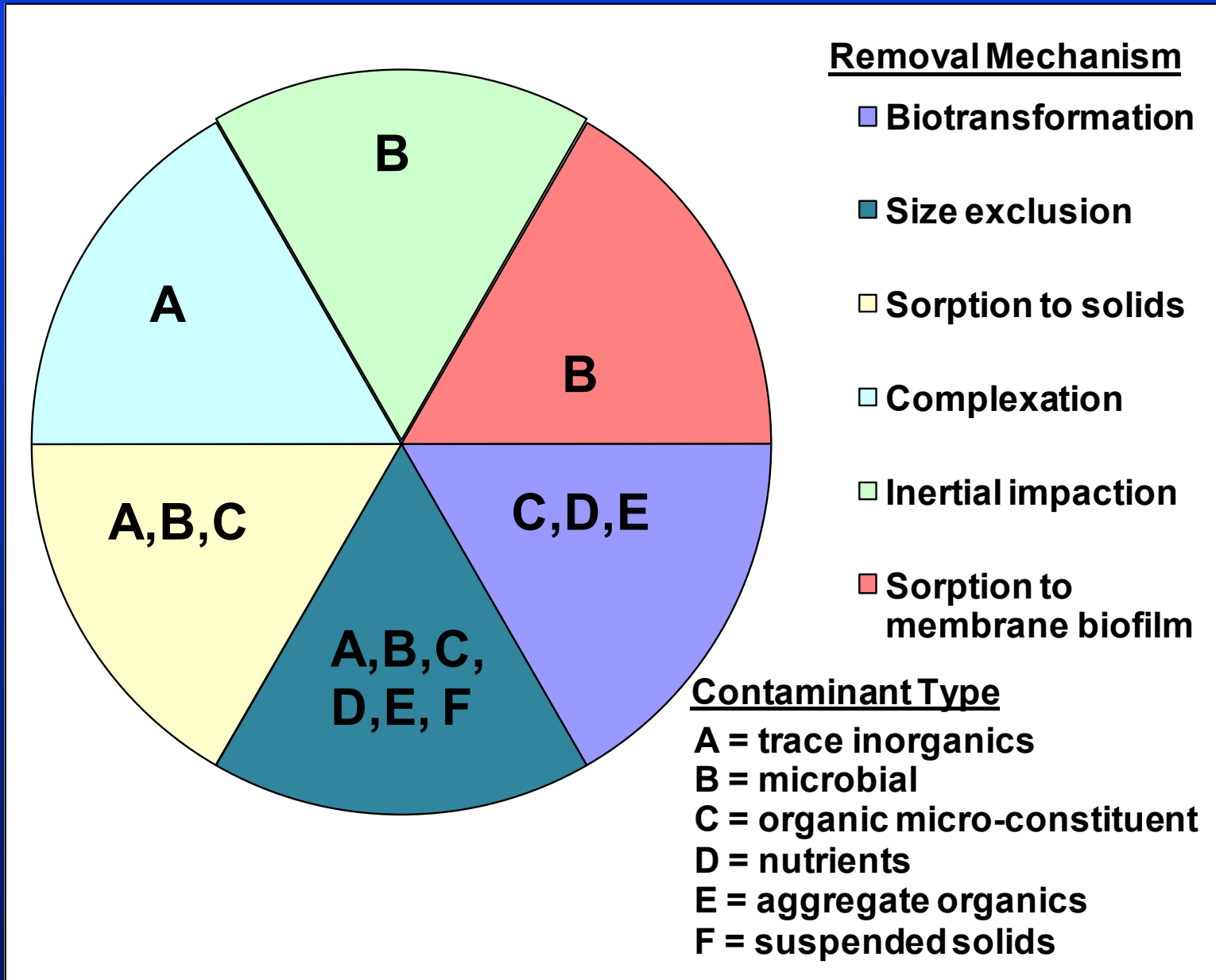
- Media filtration shows little evidence of removal while RO is very effective
- MBR performance equivalent to activated sludge operated above 10 days

WaterReuse Foundation

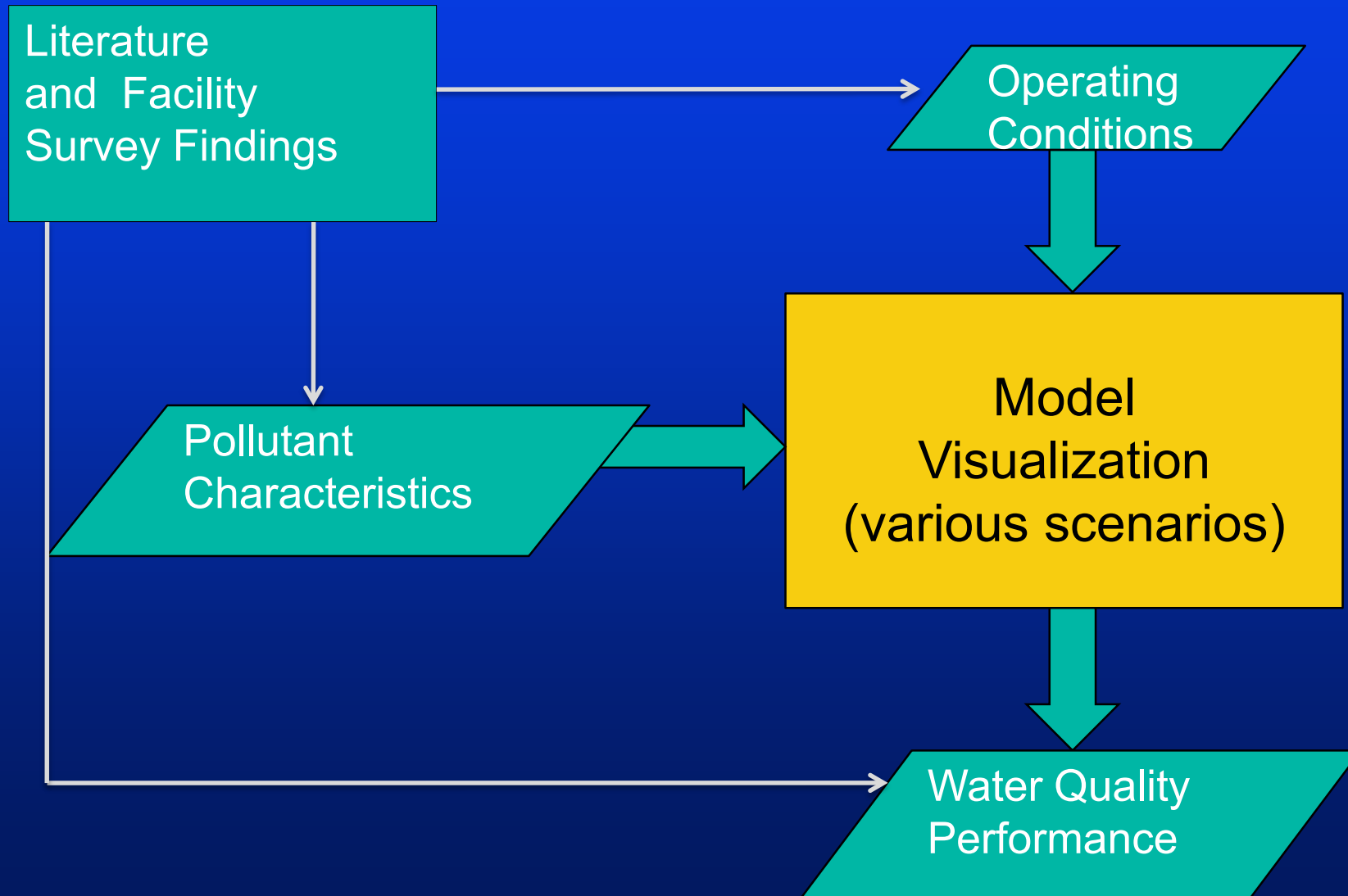
MBR Study Project Objectives

- **Consolidate Global Water Quality Performance Data**
- **Understand Water Quality Performance Boundaries**
- **Understand Operational Impacts on Water Quality Performance**
- **Optimize Water Quality and Energy Performance**

MBR REMOVAL MECHANISMS



PROJECT APPROACH



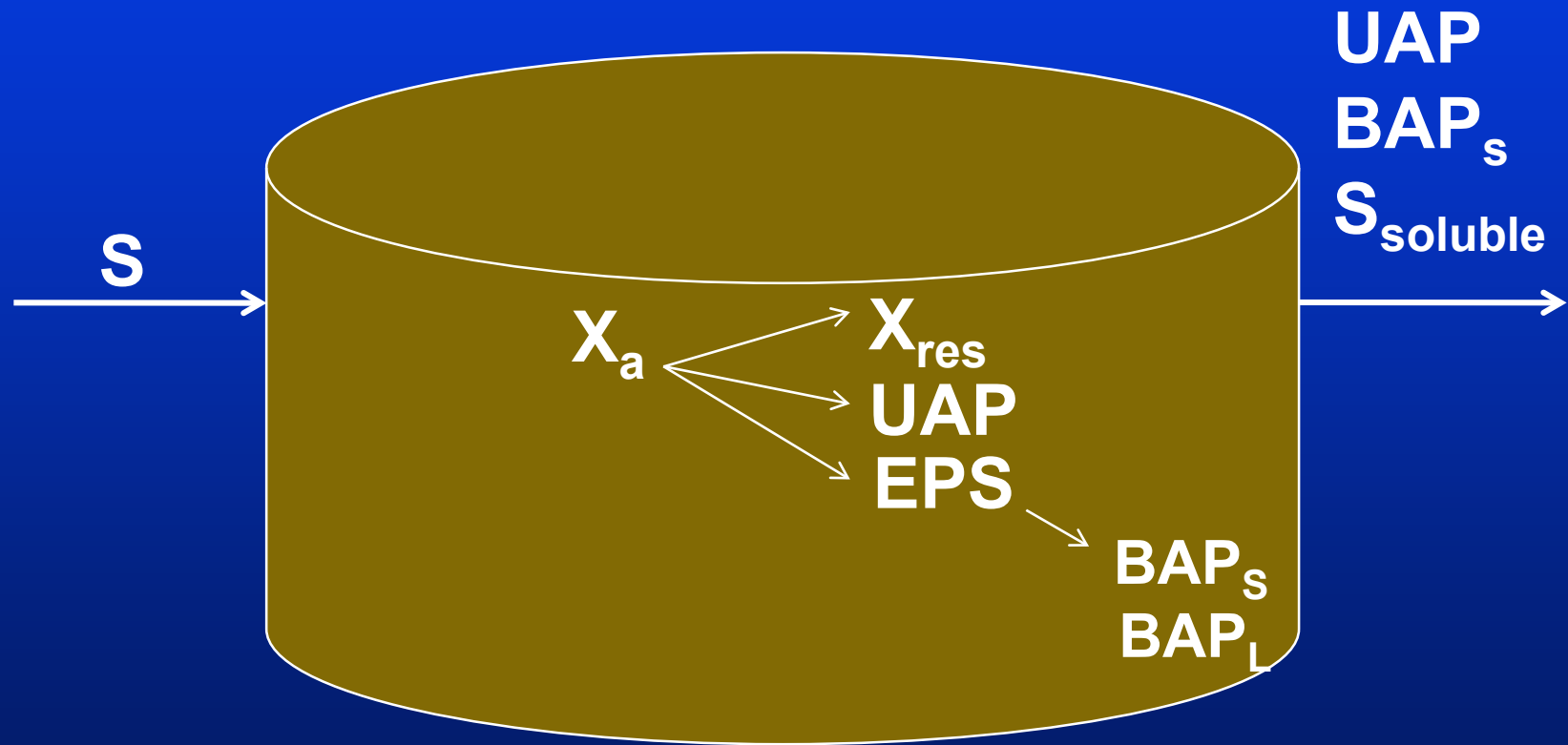
Modeling Approach

- Mechanistic Principles - Merging of 2 Models:
 - Rittmann & McCarty 2001 Activated Sludge Model
 - Laspidou & Rittmann's 2002 Unified Model for active biomass, inert biomass, EPS, and SMP
- Incorporation of Organic Micropollutant Fate Model
 - Lee, Rittmann, Shi, and McAvoy (1998) Fate Model
- Inclusion of Autotrophs for Cyclic Nitrification and Facultative Heterotrophs for Denitrification (*not yet*)

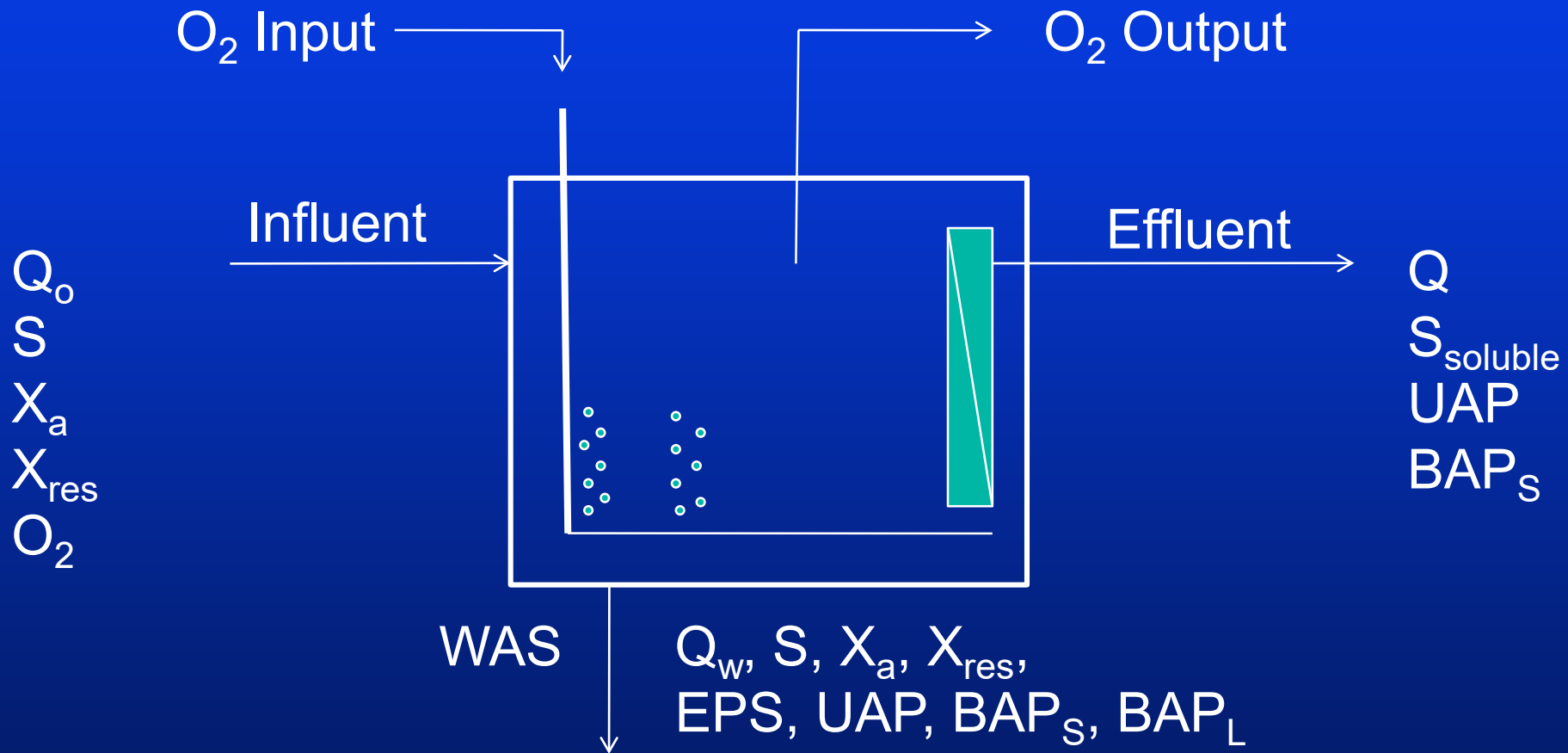
NOMENCLATURE OF COMPONENTS

- S = substrate (COD)
- X_a = active biomass
- X_{res} = residual inert biomass
- **BAP_s = small biomass-associated products**
- BAP_L = large biomass-associated products
- **EPS = extracellular polymeric substances**
- **UAP = utilization-associated products**

ACTIVATED SLUDGE BIOMASS FRACTIONS

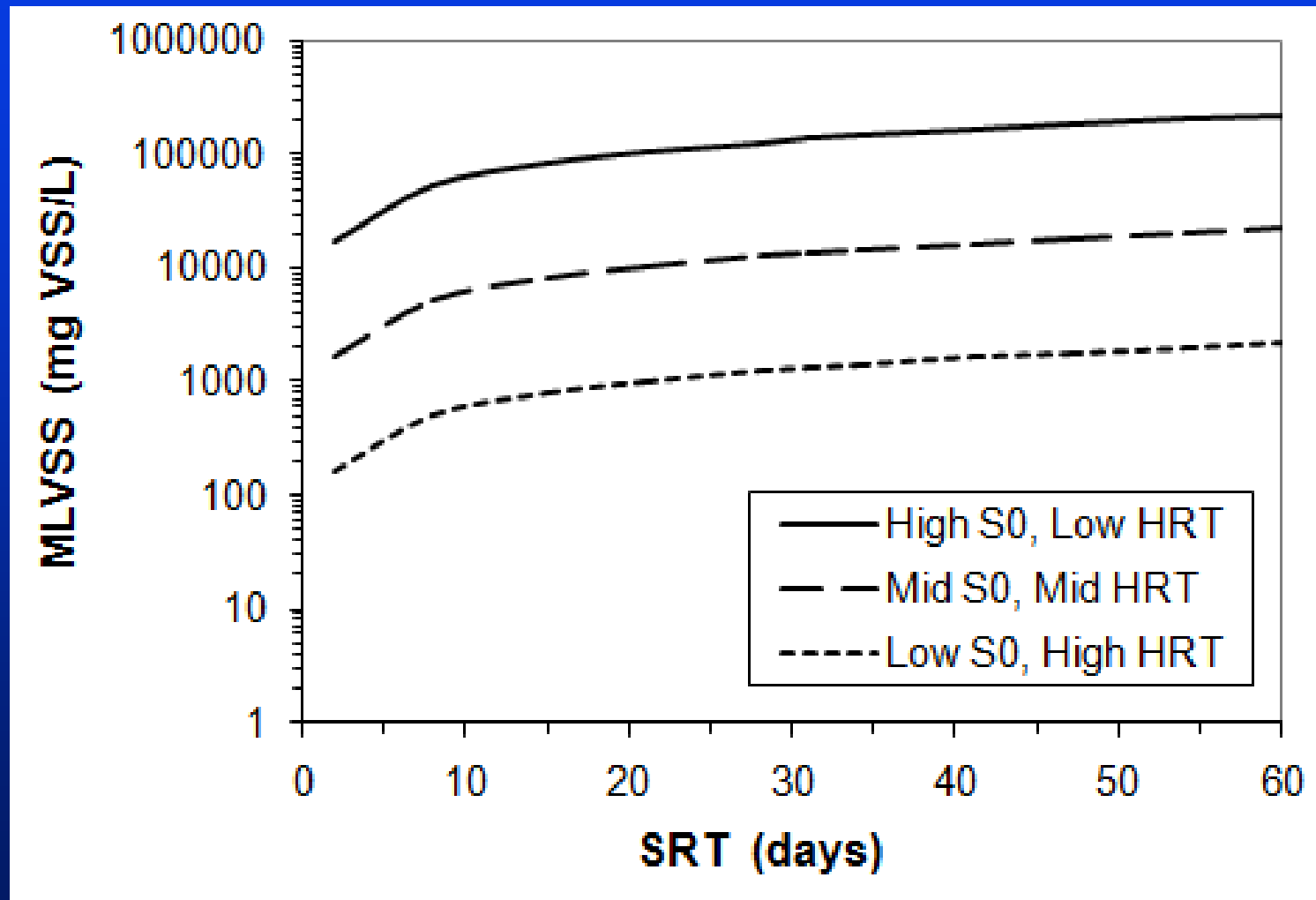


MBR MODEL INPUTS and OUTPUTS



V , SRT , Q_w = Key Operating Parameters

Impact of SRT on MLVSS



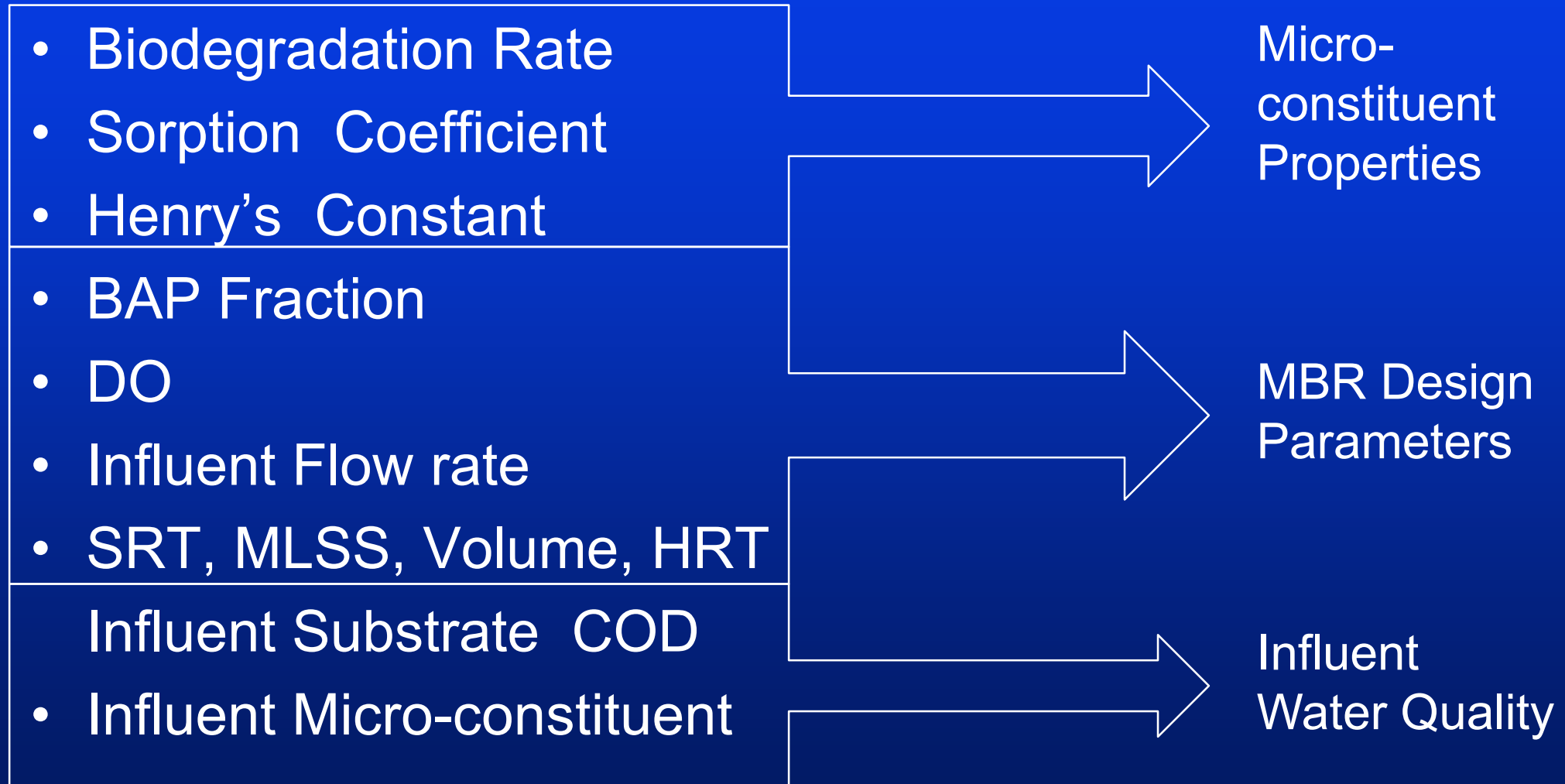
MICROPOLLUTANT FATE EQUATIONS

- Steady-state mass balance

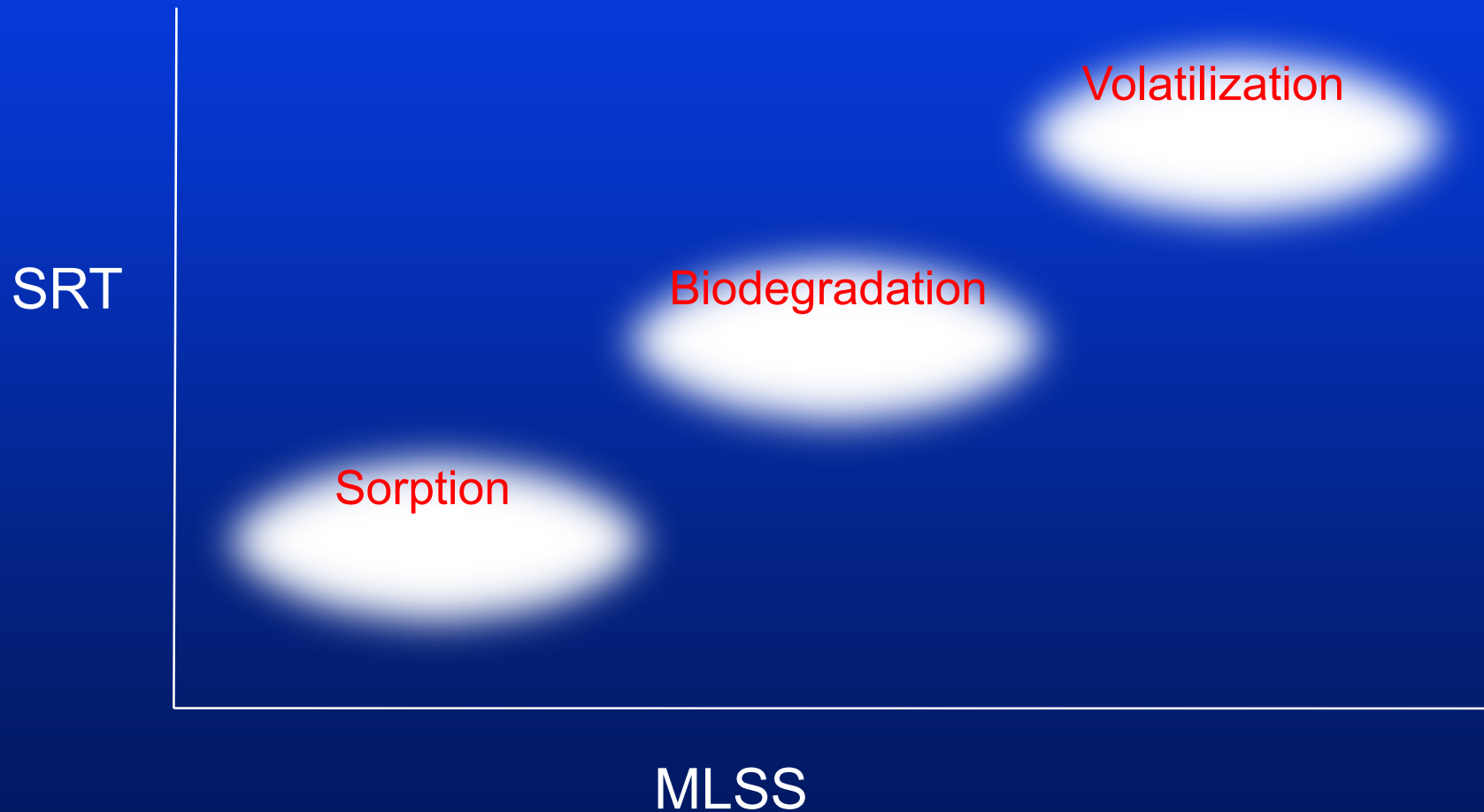
Advection In = Advection Out-Volatilization-Sorption-Biodegradation

- Volatilization at the water surface and bubble aeration
- Sorption assumes equilibrium conditions
- Biodegradation assumes first-order in C and X_a

MODEL SCENARIO INPUTS



Biodegradation Controlling Variable for Many Organic Micropollutants

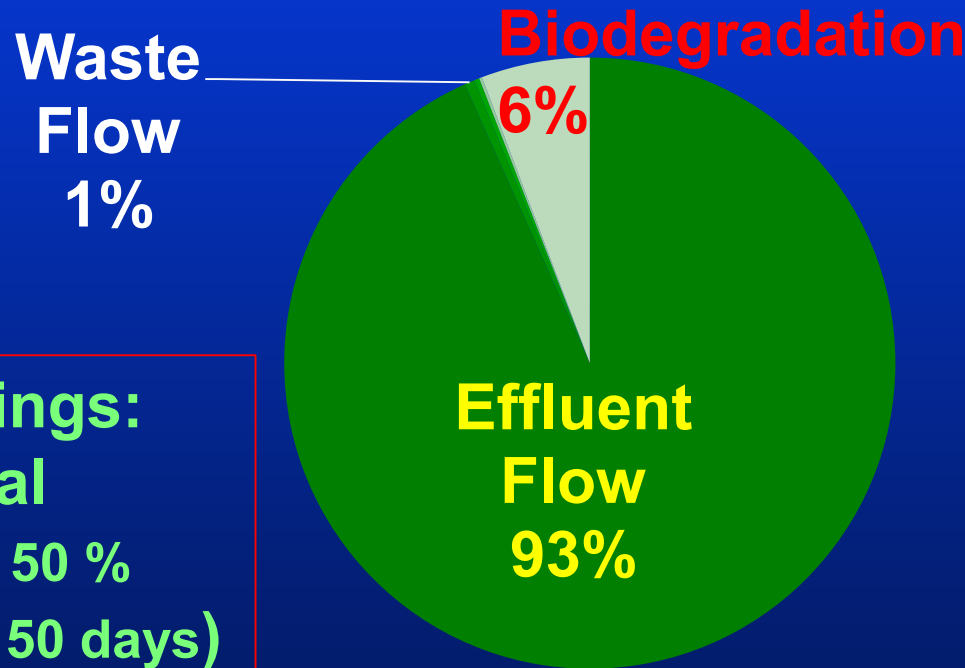


MODELING FOUR PHARMACEUTICALS

	K_b <u>(L/gSS-d)</u>	K_D <u>(L/kgSS)</u>
• Diclofenac (NSAID)	0.1	16
• Clofibric Acid (hyperlipidemia)	0.165	5
• Naproxen ((NSAID)	0.6	13
• Ibuprofen (NSAID)	15.5	7

LOW K_b and LOW K_D

Diclofenac
15-day SRT
20,000 mg/L MLSS

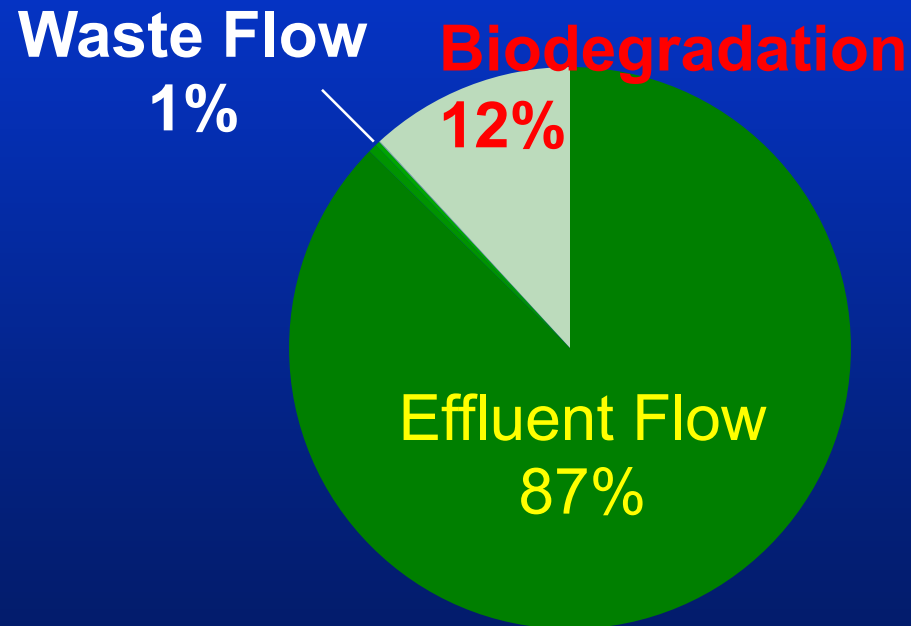


Literature Findings:
0 -35 % Removal
(Two papers cited 50 %
removal at SRT > 50 days)

SLIGHTLY HIGHER K_b

Literature Findings:
54 - 88 % Removal

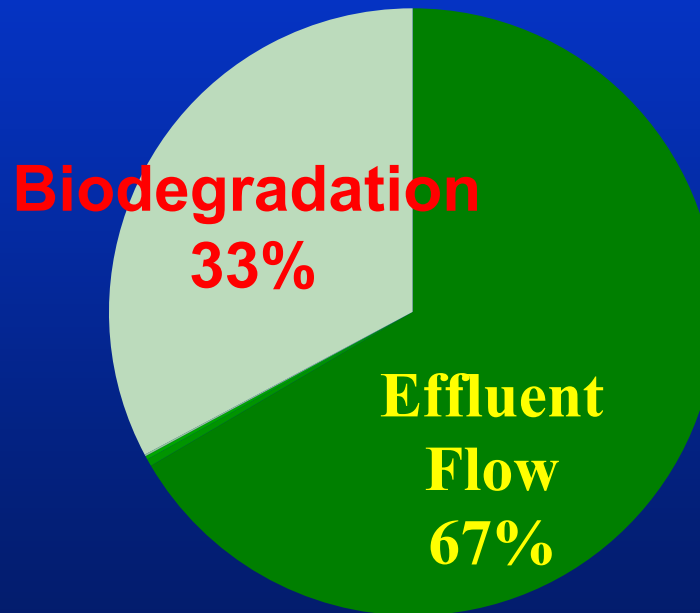
Clofibric Acid
15-day SRT
20,000 mg/L MLSS



STILL HIGHER K_b

Literature Findings:
69 - 99 % Removal

Naproxen
15-day SRT
20,000 mg/L MLSS

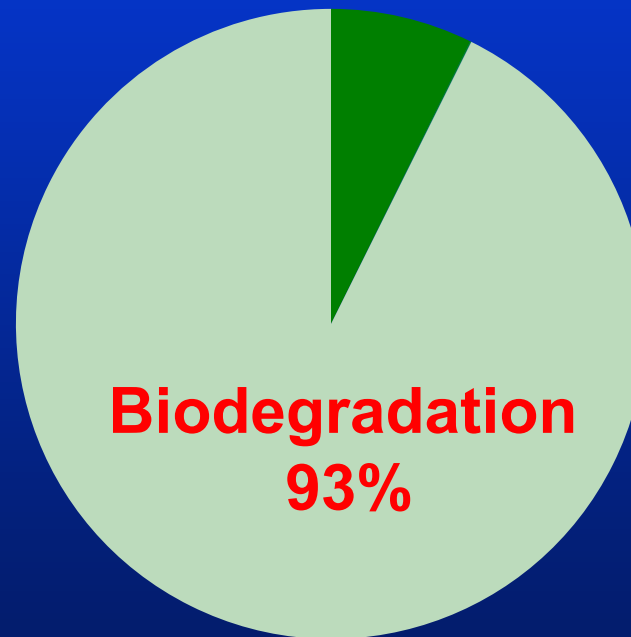


SIGNIFICANT K_b

Literature Findings:
88 - 100 % Removal

Ibuprofen
15-day SRT
20,000 mg/L MLSS

Effluent
Flow
7%



BIN CATEGORIZATION of MICROPOLLUTANTS from LITERATURE

Low Removal (<20%)	Moderate Removal	High Removal (> 80%)
Atrazine EDTA TCEP TCPP Carbamezapine Diclofenac	Atenolol Iopromide Clofibric Acid Gemfibrozil Sulfamethoxazole DEET Ethinylestradiol Naproxen	Androstenedione Caffeine Estradiol Estriol Acetaminophen Nonylphenol Benzfibrate Ibuprofen

BIN CATEGORIZATION of MICROPOLLUTANTS from MODELING

Low Removal	Moderate Removal	High Removal
$K_b < 0.5 \text{ L/g}_{SS}\text{-day}$ OR $K_D < 4 \text{ L/g}_{SS}$	$K_b \sim 1 \text{ L/g}_{SS}\text{-day}$	$K_b > 2.5 \text{ L/g}_{SS}\text{-day}$ OR $K_D > 40 \text{ L/g}_{SS}$

RELEVANT CONCLUSIONS

- MBR micropollutant removal mainly biological rather than membrane barrier
- Heterotroph model predicts approximately 50% reduction for K_b value of $1 \text{ L} \cdot \text{g}_{\text{SS}}^{-1} \text{ d}^{-1}$
- Literature values may be higher than model predictions because of presence of autotrophs
- Biodegradation enhanced at higher MLVSS, but only up to SRT of ~ 20 days when Xa predominates



THANK YOU

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