

Next-generation rechargeable batteries enabled by 3D zinc anodes

Team Zinc

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NRL's "Advanced Electrochemical Materials" Section: Addressing Navy & Marine Corps Power & Energy Needs through Focused R&D

NRL 6171's motivations & themes for power source development:

- ❑ Scalable energy-storage:
micropower → personal power → grid-scale
- ❑ Battery chemistries inherently safer
than Li-ion → *target aqueous systems*
- ❑ Low-cost, domestically sourced
components; manufacture-friendly
fabrication processes
- ❑ Variable form factors for ready integration into
established architectures
→ *opportunities for "structural" energy storage*
- ❑ Hybrid configurations that integrate various
energy storage/conversion components
→ *energy-dense battery + high-power capacitor*

Navy/Marine Corps applications that we target:

- Portable electronics
- Communications
- Wearable power
- Autonomous sensors
- UUV & UAV
- Hybrid vehicles
- Wireless recharging
- Directed-energy weapons
- Space-based platforms

Looking towards the next energy- & power-dense (& safe) battery

Zn-based batteries

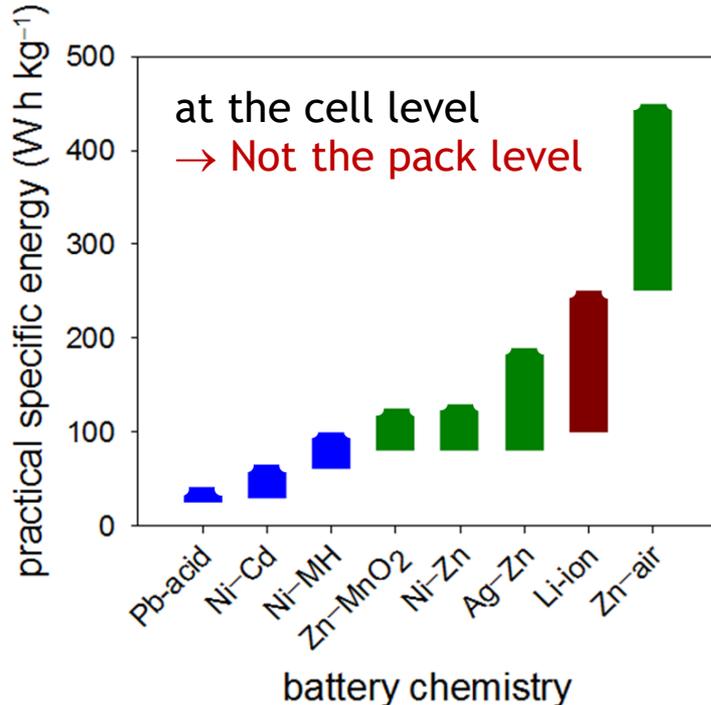
- ✓ High practical specific energy
- ✓ High specific power
- ✓ Safe, aqueous electrolyte
- ✓ Globally available materials
- ⊗ Limited rechargeability
- ⊗ Subpar material utilization



*Once Zinc is made
rechargeable*

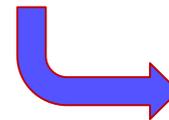


Rechargeable Zn-air, Ag-Zn,
Ni-Zn, MnO₂-Zn & Zn-ion batteries



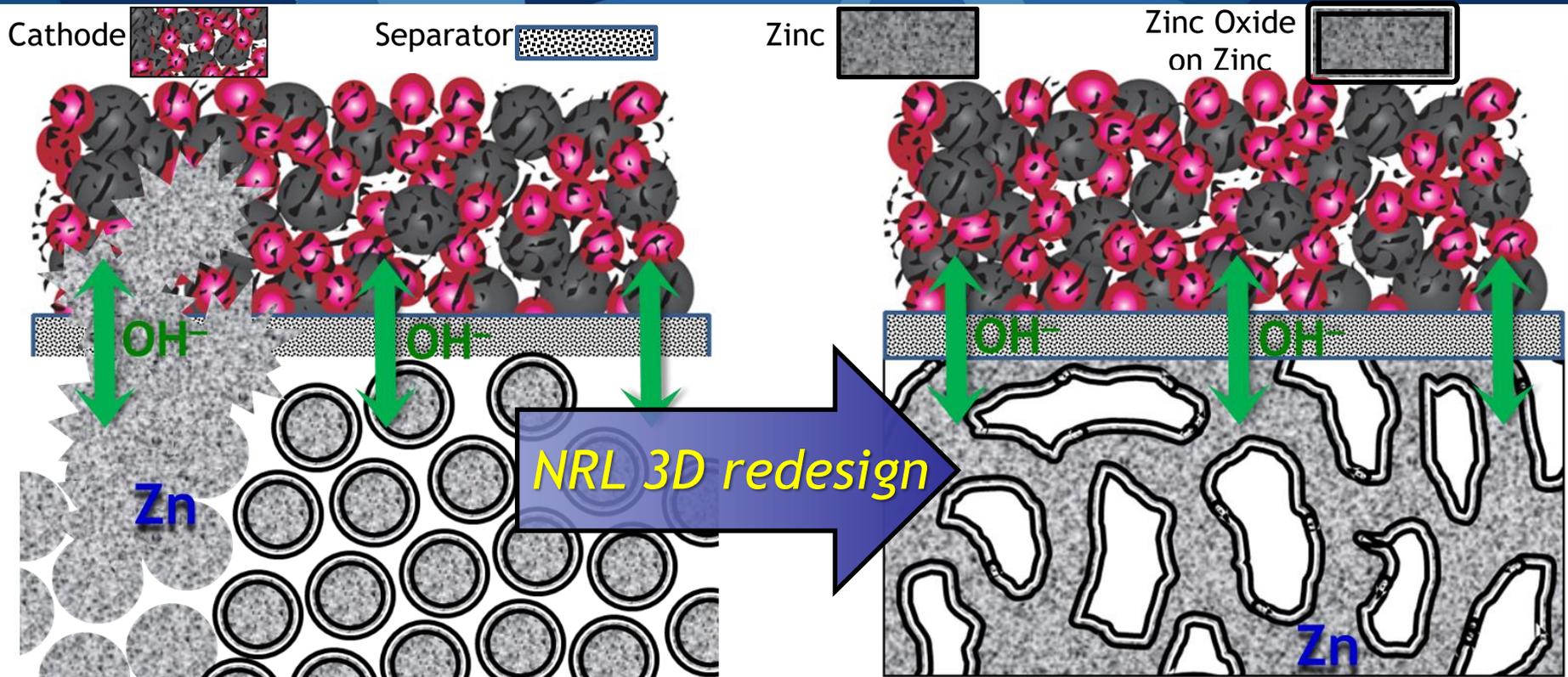
What will be the impact?

- ☐ Replace Pb-acid, Ni-Cd & Ni-MH batteries with higher energy density, rechargeable, Zn-X batteries
- ☐ Provide alternative to Li-ion batteries in niche applications or extreme environments or where safety is high priority



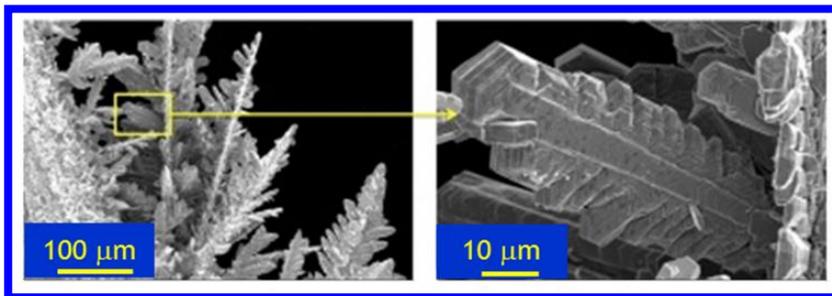
*Disruptive technology for the
military & consumer battery
market*

NRL Breakthrough: 3D redesign of the Zn anode

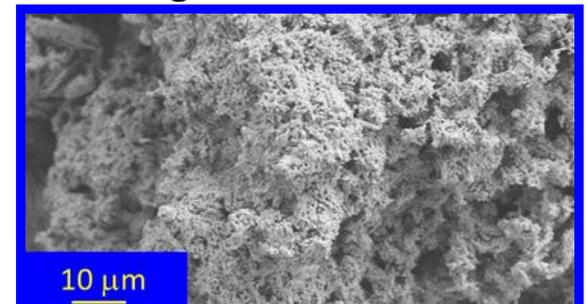


Traditional powder-composite Zn anode

Next-gen 3D Zn

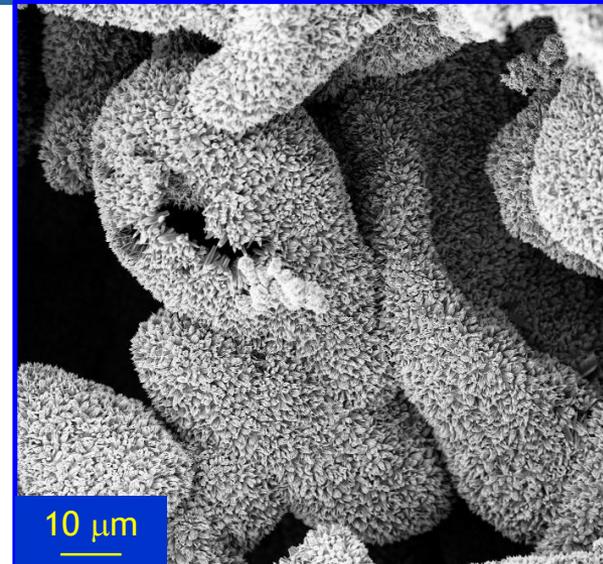
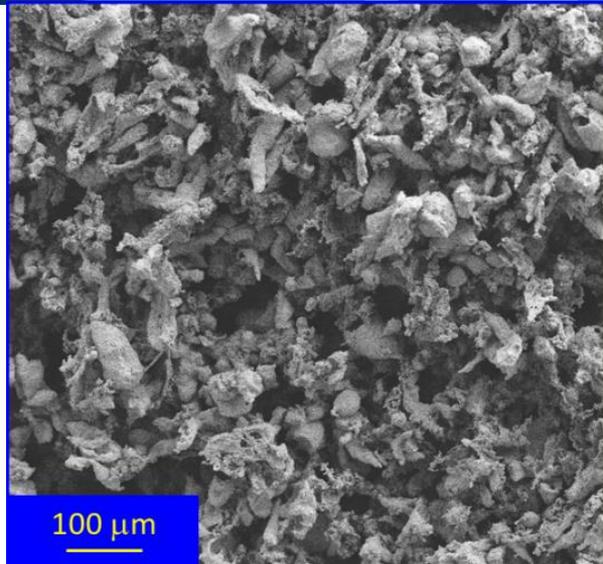


Current density
required to
Initiate dendrites:
 $\sim 10 \text{ mA cm}^{-2}$
Oxley, Fleishmann,
NSWC Quarterly
Report (1966)



NRL Breakthrough

3D redesign of the Zn anode



✓ *Scalable, monolithic form factor*

- tailor size to desired cell dimensions
- mechanically robust

✓ *Fully metallic, highly conductive pathways in 3D*

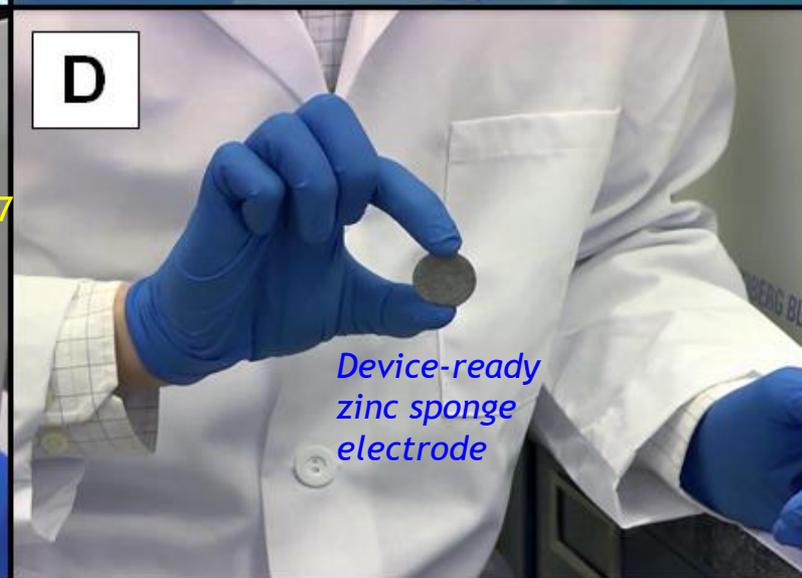
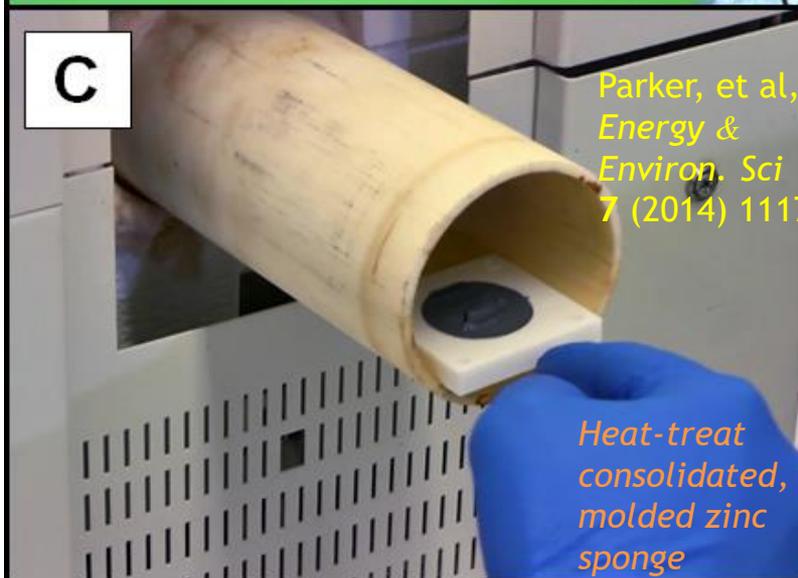
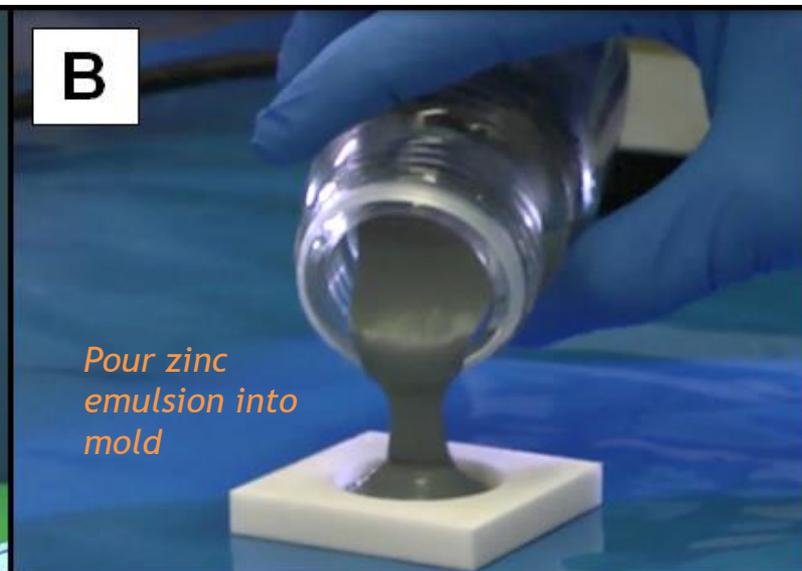
- improve current distribution
- suppress dendrite formation

✓ *Confined volume elements*

- ensure more uniform ZnO precipitation inside & out on discharge
- ZnO@Zn core-shell

Fabrication of 3D Zn (“sponge”) anode

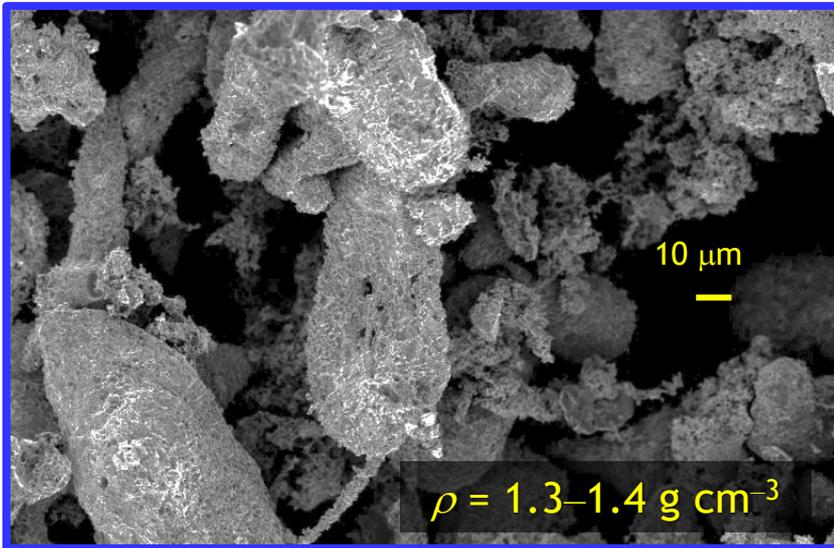
<https://www.youtube.com/watch?v=YnRp49d4FMA>



Increasing density of Zn sponge: Gen-1 (Zn_{20}) to Gen-2 (Zn_{30})

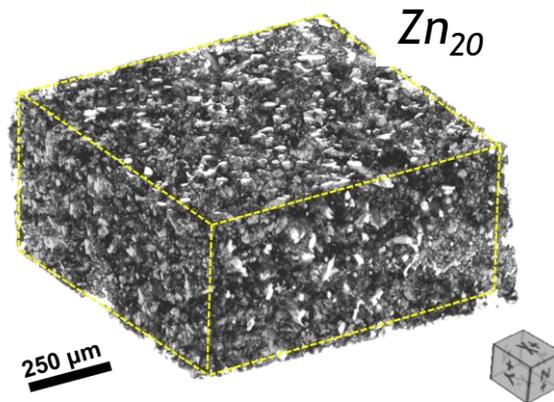
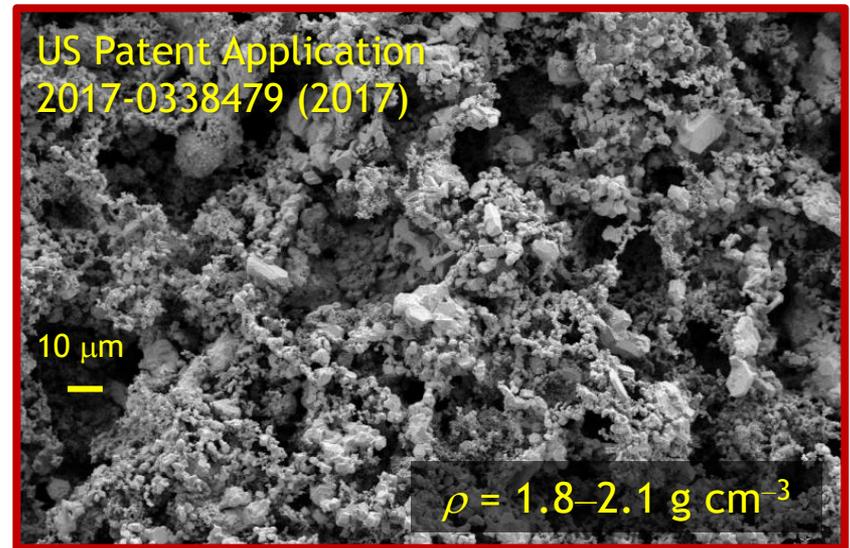
Gen-1 (Zn_{20})

→ 20% volume fill of Zn

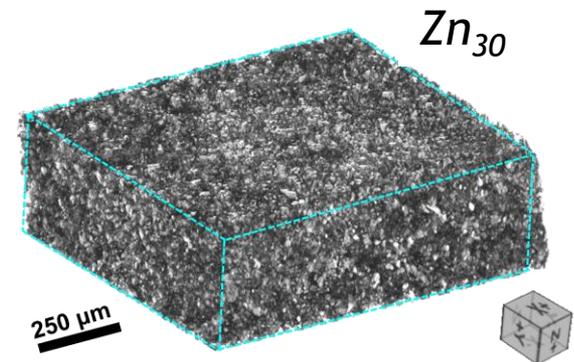


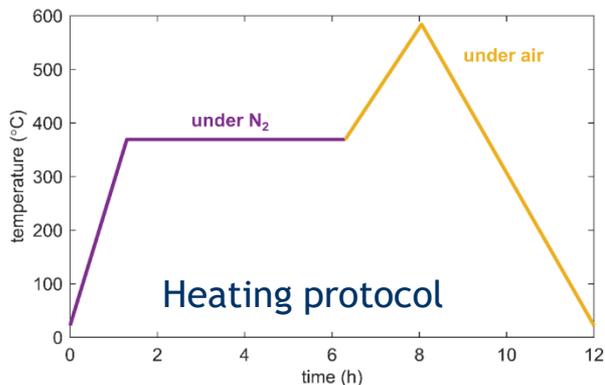
Gen-2 (Zn_{30})

→ Denser, more mechanically robust

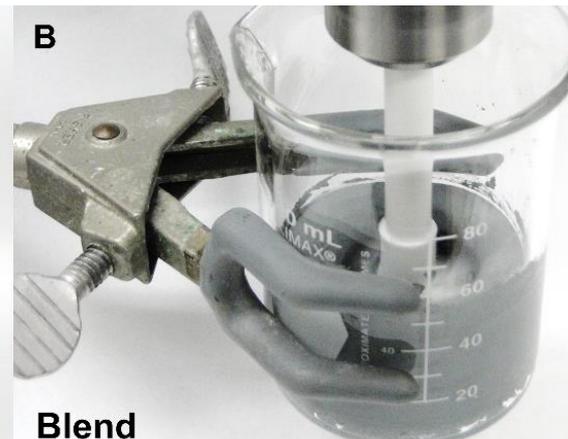


X-ray tomography verifies
3D interconnectivity of
monolithic Zn sponges





**Pour Zn
into emulsion**



Blend

C

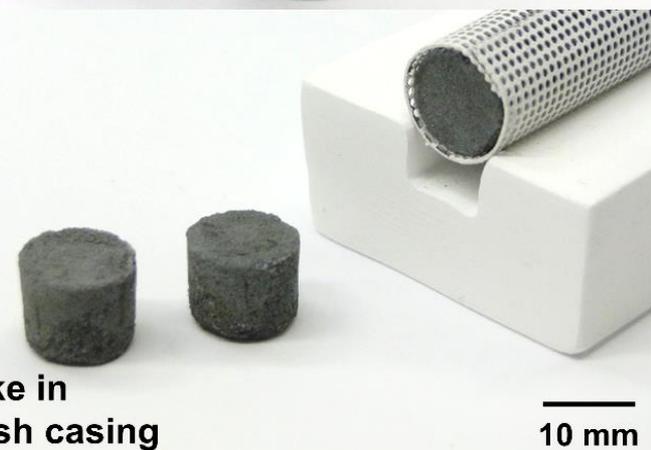
No
electroreduction
required before
cell assembly

**Portion into
drying molds**

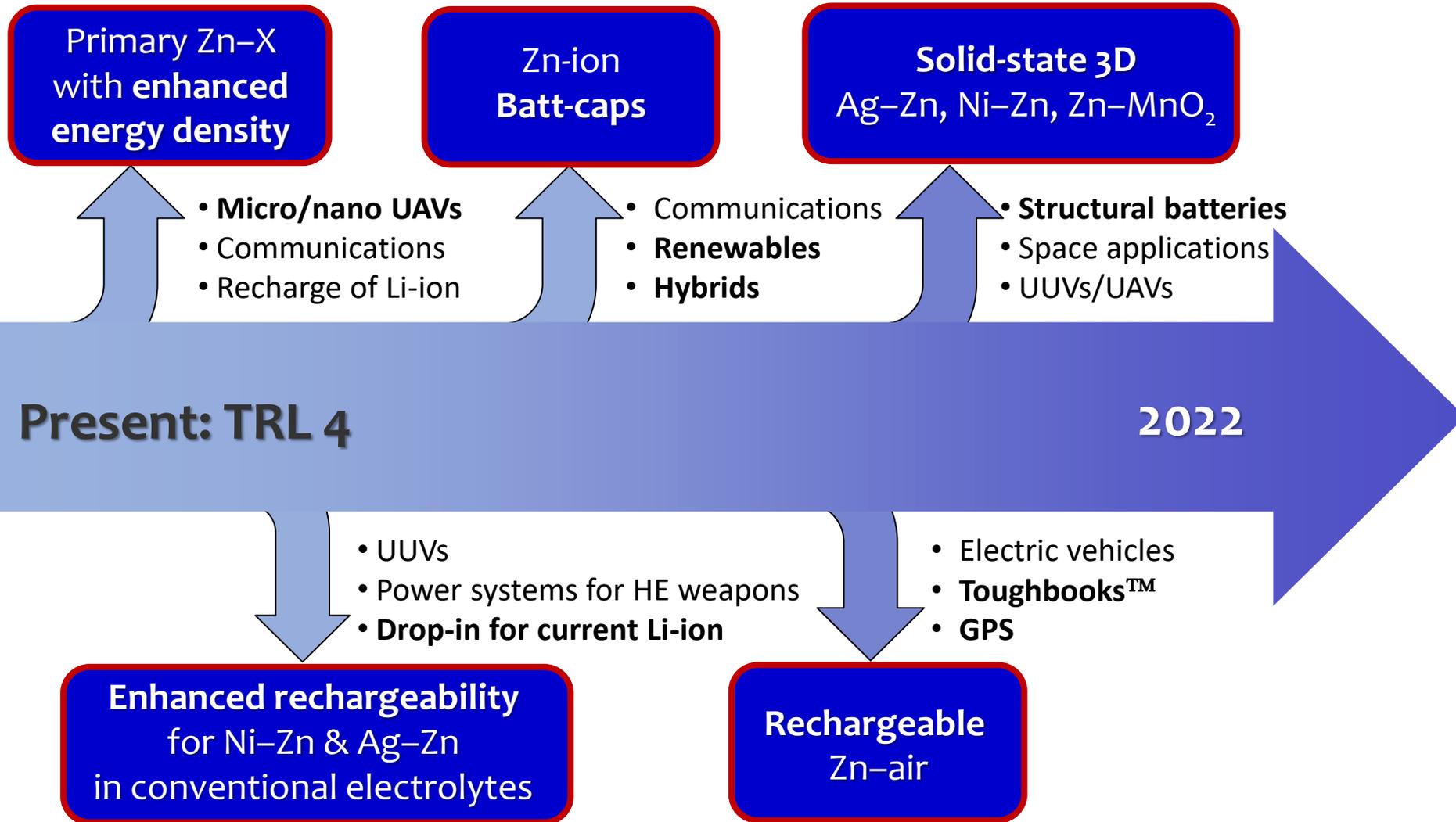


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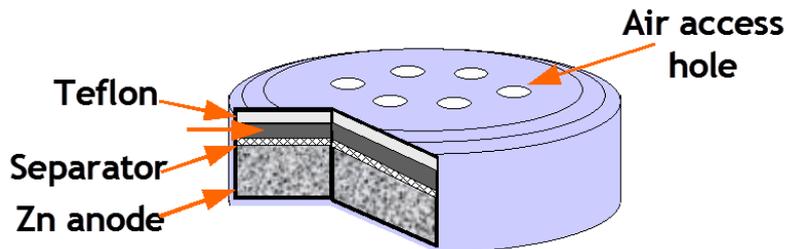
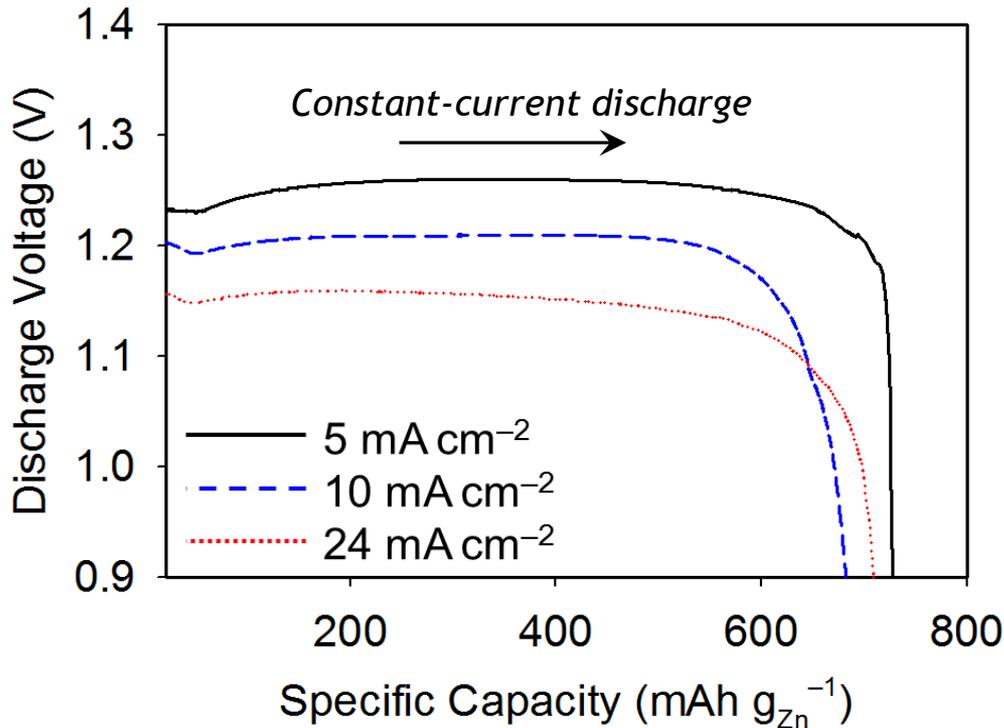
**Bake in
mesh casing**



Hopkins, Sassin, Chervin, DeSario, Parker, Long, Rolison, "Low-cost green synthesis of zinc sponge for rechargeable, sustainable batteries," *Sustainable Energy & Fuels* (2020)



Evaluation in prototype primary Zn-air cells

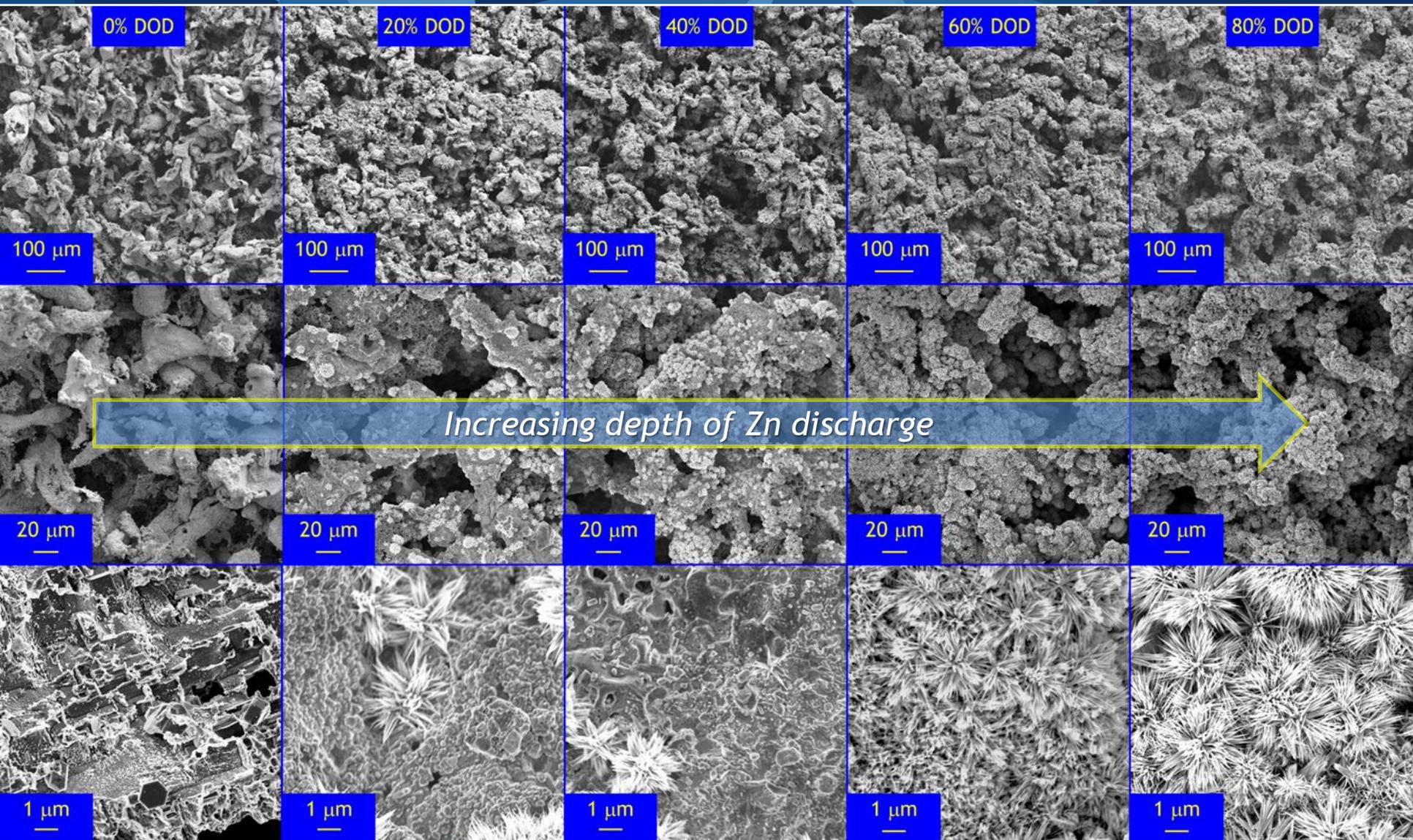


http://www.duracell.com/oem/primary/Zinc/zinc_air_tech.asp

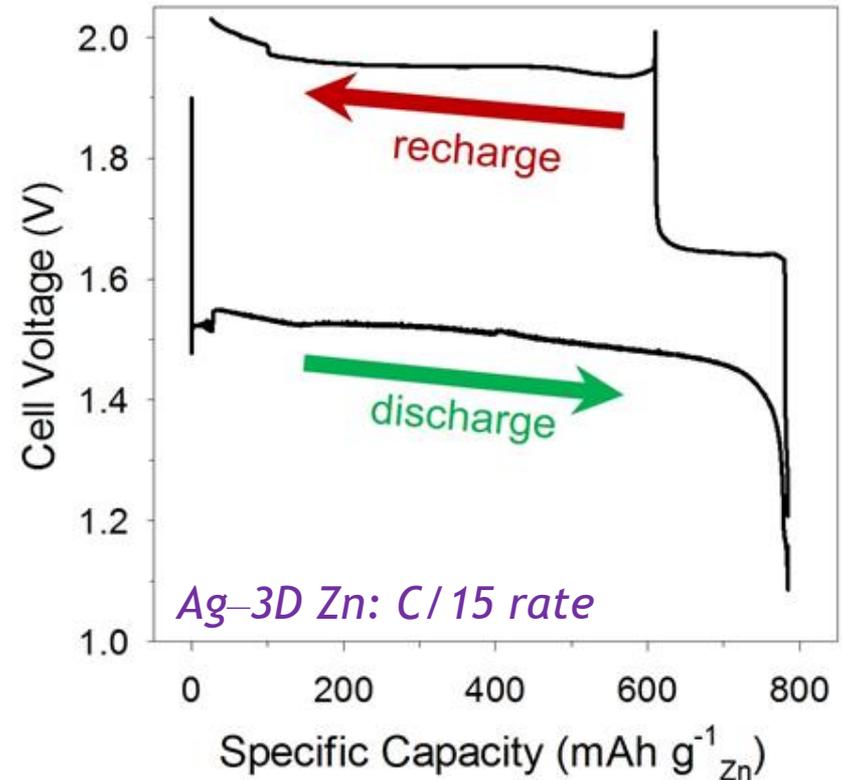
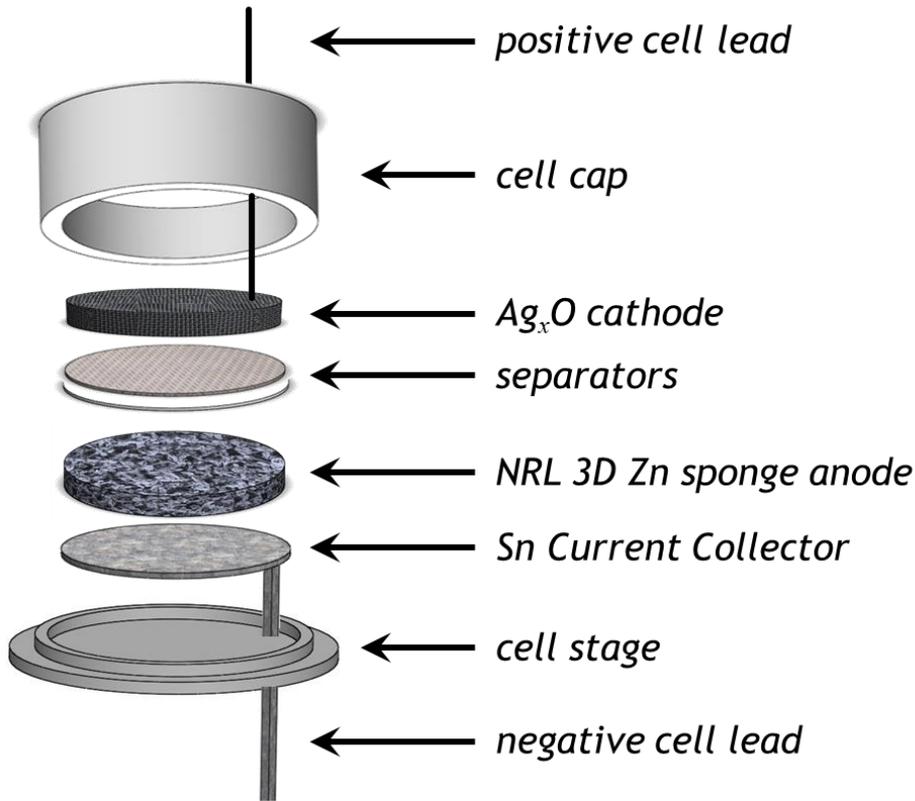
- ✓ NRL's 3D Zn sponge increases battery lifetime by 50% relative to commercial batteries
- ✓ Allows for plug-&-play into existing Zn-based battery technologies
 - Ag-Zn
 - Ni-Zn
 - Zn-air

	Percentage utilization of Zn
Commercial Zn-Air (Zn bed)*	60
NRL Zn-Air (3D Zn sponge)	95

Maintaining 3D Zn structure to deep discharge



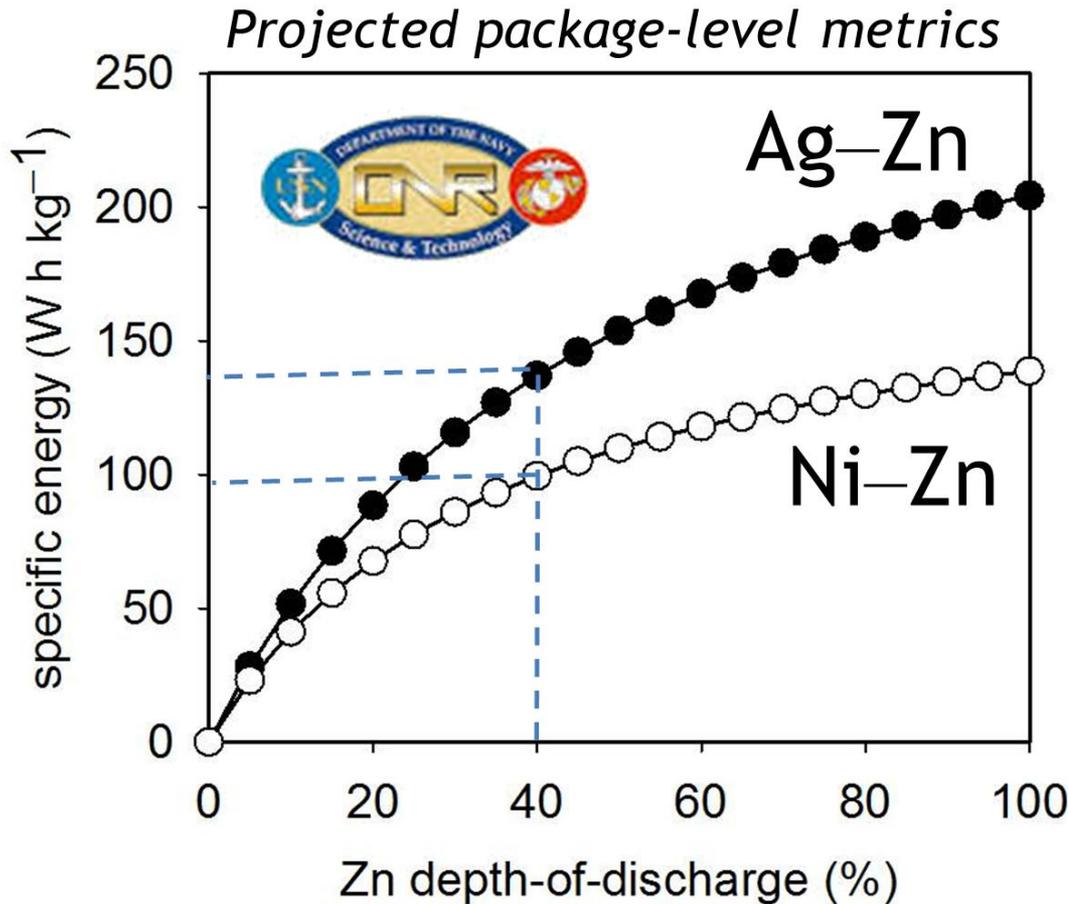
Accessing unprecedented levels of discharge (and recharge) with 3D Zn anodes



✓ 95% of theoretical
Zn specific capacity
($820 \text{ mA h g}^{-1}_{\text{Zn}}$)

✓ 97% recovery of
exhaustive discharge

→ Only 40–60% DOD_{Zn}
achieved with
commercial Zn-
based batteries



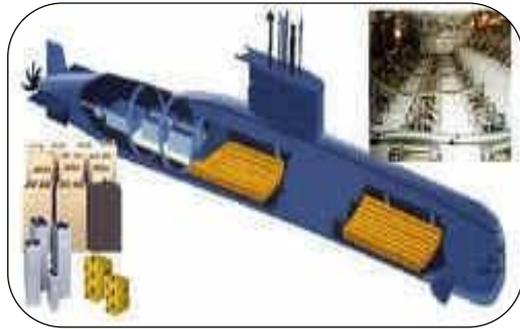
“Depth-of-discharge” metrics for commercial batteries typically refer to “nominal” capacity:

→ Commercial rechargeable Ni-Zn cells typically tap ~20% theoretical DOD_{Zn} (up to 66 Wh kg⁻¹)

→ NRL 3D Zn supports $\geq 40\%$ theoretical DOD_{Zn} with extended cycling

Pushing DoD_{Zn} beyond 60%? Not necessary!

USN applications for Ni-3D Zn & Ag-3D Zn batteries



Submarines

Ni-3D Zn: energy density + lifetime

Ag-3D Zn: ✓ ✓ + power

SEAL Delivery Vehicles

Ni-3D Zn: safety + cost

Ag-3D Zn: safety + power



Ship Backup-Power

Ni-3D Zn: reliability + lifetime

Ag-3D Zn: ✓ ✓ + power

Unmanned Vehicles

Ni-3D Zn: cost

Ag-3D Zn: power



Replace Pb-acid batteries

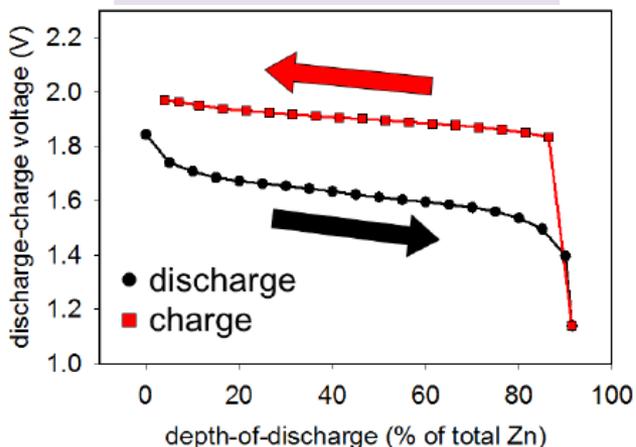
Opportunities
for next-generation
Ni-3D Zn & Ag-3D Zn
batteries

Replace Li-ion batteries

Ni-3D Zn batteries →

High performance in three fields of use

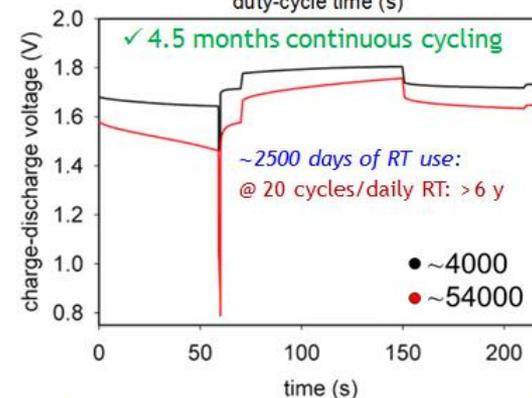
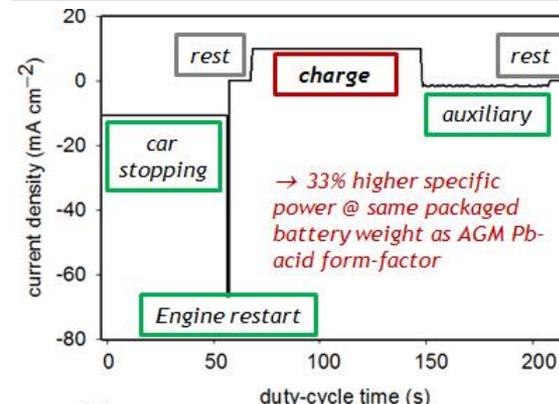
Single discharge to deep utilization



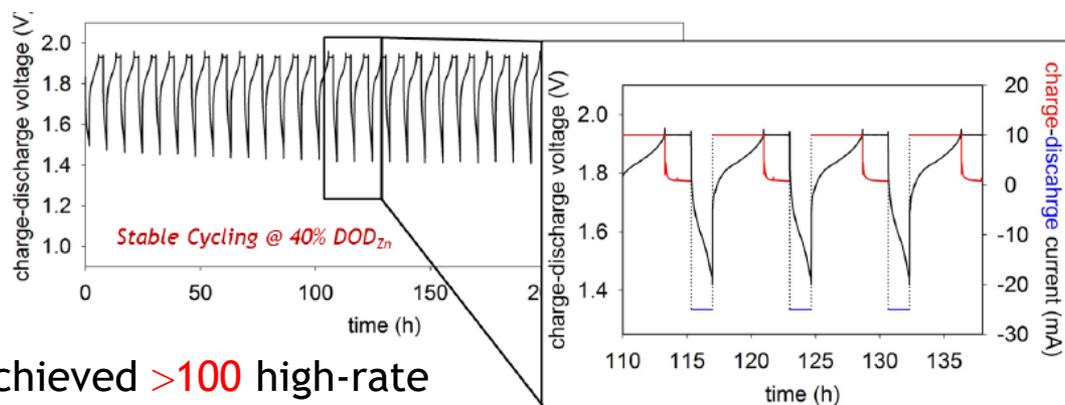
“Rechargeable nickel-3D zinc batteries: An energy-dense, safer alternative to lithium-ion.”
Parker, Chervin, Pala, Machler, Burz, Long, Rolison
Science **356** (2017) 415

Achieved **>90% DOD** with **95%** capacity recovery on charge

High-power duty cycles relevant to start-stop EVs

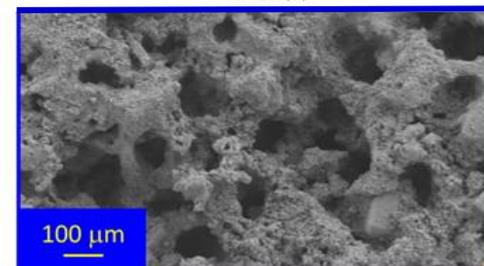


Rechargeability at Li-ion competitive specific energy



Achieved **>100** high-rate cycles (C/1.5 discharge)

328 mA h g⁻¹_{Zn} cycling



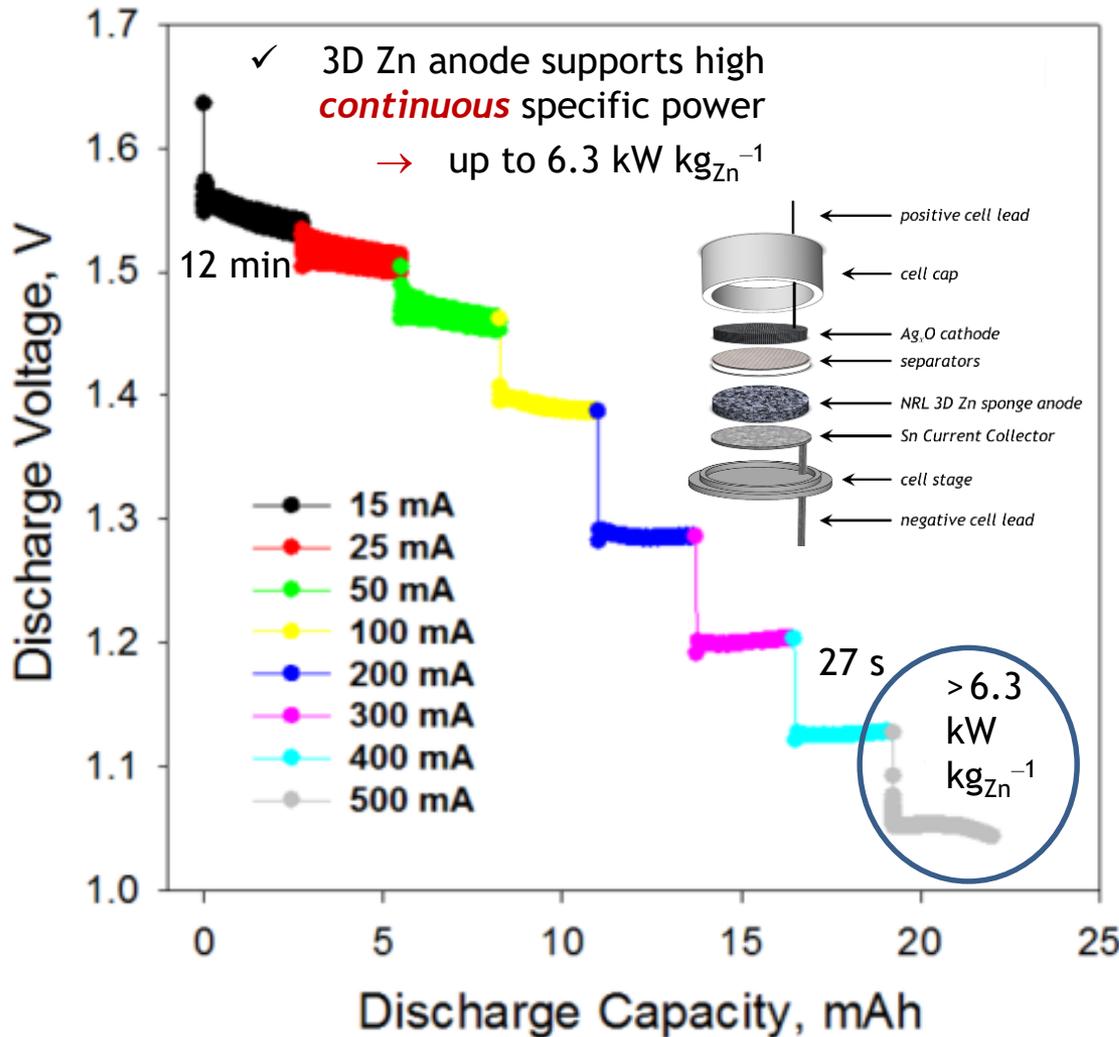
Zn sponge anode delivers performance in Ni-3D Zn cells to power EVs



Parker, Chervin, Pala, Machler, Burz, Long, Rotison,
Science 356 (2017) 415

	E-bike		Start-stop / Microhybrid		Battery Electric Vehicle	
	SLA ^[1]	Ni-3D Zn	AGM ^[2]	Ni-3D Zn	Li-ion ^[3]	Ni-3D Zn
Energy Capacity (Wh)	540	540	1720	1720	24000	24000
Weight (kg)	12.2	5.9	45.0	21.7	339	220
Effective ^[4] Specific Energy (Wh kg ⁻¹)	44.3	91.8	38.2	79.2	71	109
Effective Energy Density (Wh L ⁻¹)	140	225	126	164	96	216
Range (miles; years for start-stop)	10-18	10-18	1-2 y	4-5 y (est.)	84	90

Extreme power of Ag-3D Zn₃₀ cells

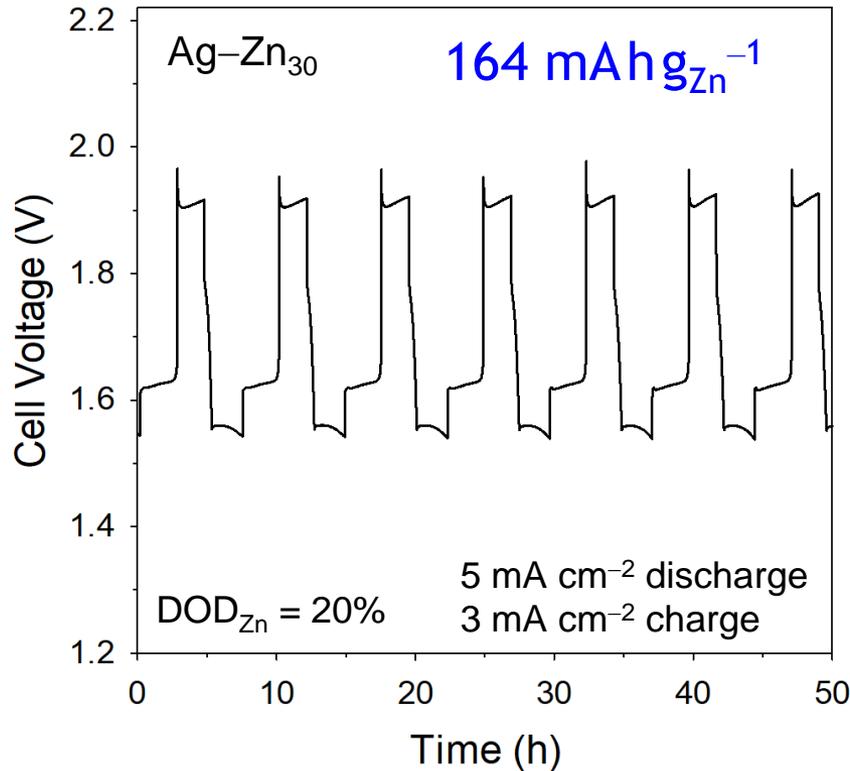


- Denser Zn sponges enable routine use of thinner anodes
- Thinner electrodes confer higher power

Power performance:

- ✓ Peak rate of 500 mA cm⁻² (~17C rate)
- ✓ Peak power of 6,300 W kg_{Zn}⁻¹
- ✓ 1,200 W kg⁻¹ in fully packaged Ag-3D Zn cell

Excellent cycling performance of Ag-3D Zn₃₀ cells



- 100 cycles for Zn₃₀ sponges vs. <50 cycles for Zn₂₀
 - even with using unoptimized Ag cathode + unoptimized separator system)
- 100% coulombic efficiency
- *still* no dendrites ...

More uniformly reactive Zn → more uniformly reactive Ag

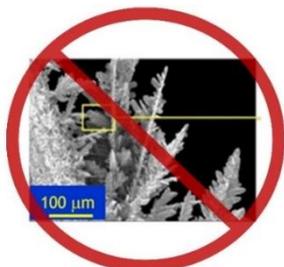
Rechargeable Zn-air: The Holy Grail of zinc batteries

❑ Integrate multiple functions to achieve next-gen rechargeable Zn-air battery:

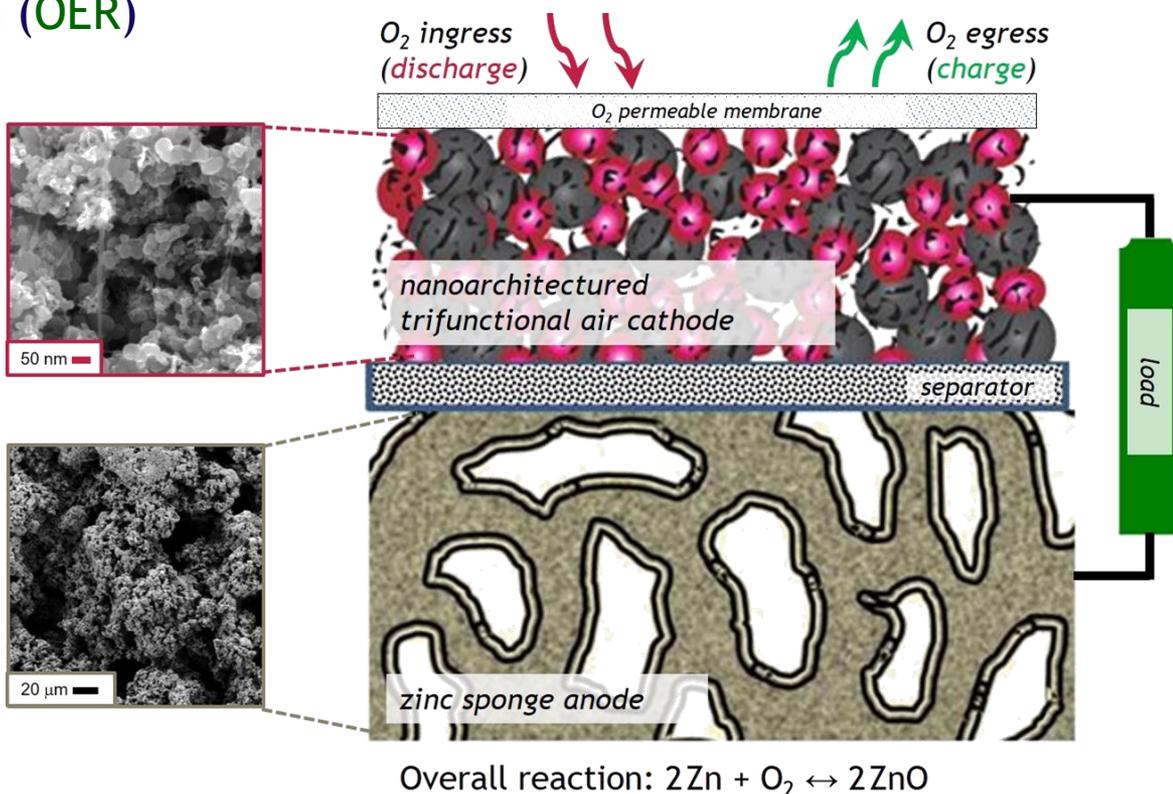
- (1) "Fuel cell" → O₂ reduction (ORR)
- (2) "Electrolyzer" → O₂ evolution (OER)
- (3) "Electrochemical capacitor"
→ Pulse power

↳ "Trifunctional"
air cathodes

- (4) "Battery" → Reversible Zn



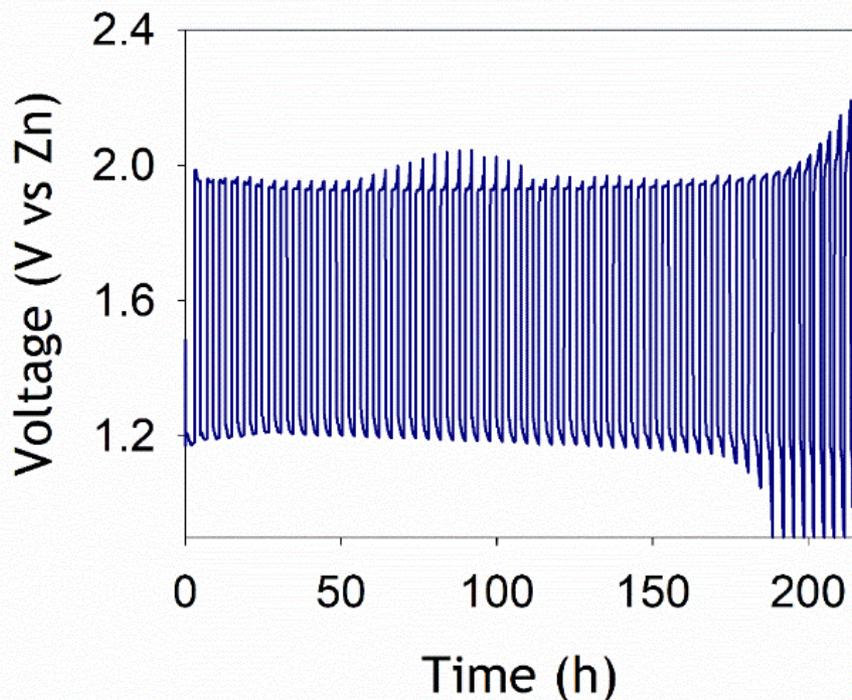
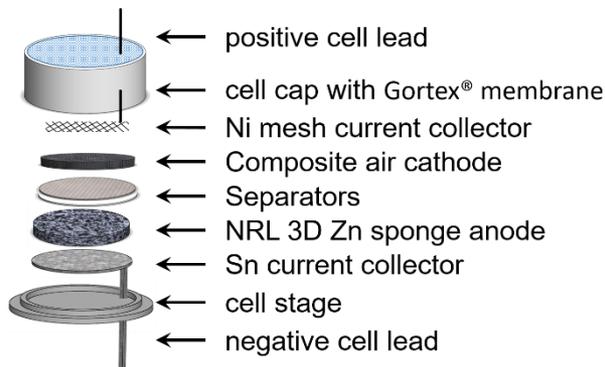
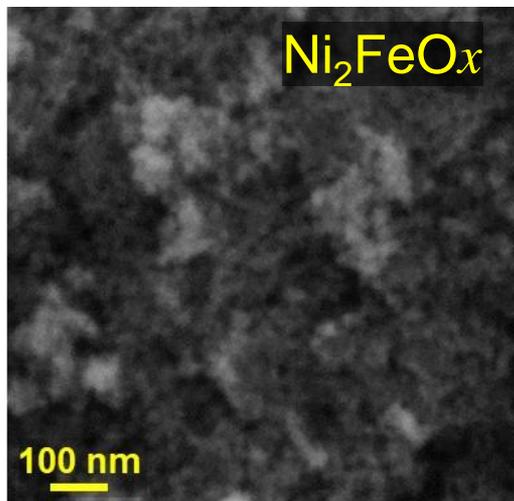
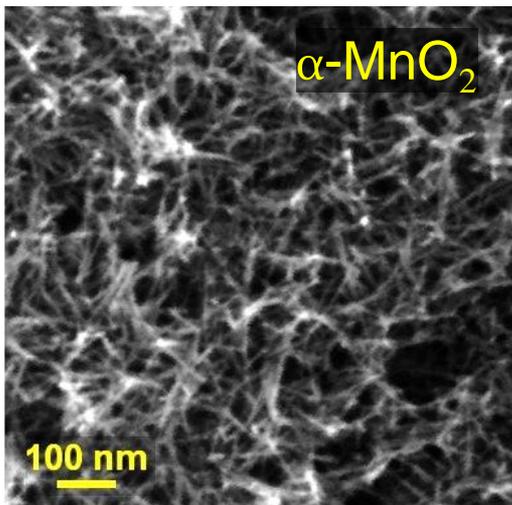
No Dendrites!



✓ Target specific energy of 300 Whkg⁻¹ with >500 cycles

Rechargeable Zn-air: 3D Zn + Nanoarchitected e-catalysts

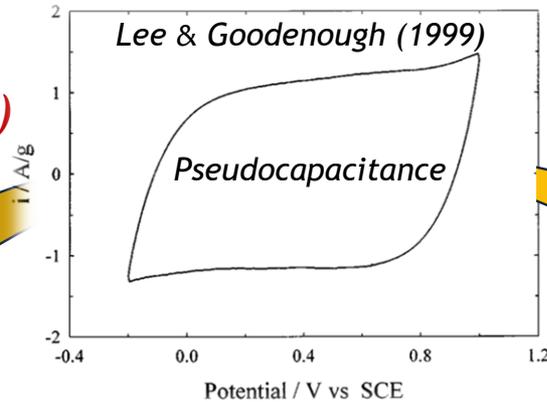
Air cathode comprising two
aerogel-type electrocatalysts



*Promising initial
cycling, but
additional
optimization of
air cathode
needed to reach
cycle-life goals*

Revisiting Zn–MnO₂ chemistry: Rechargeable “Zn-ion” batteries

- **Asymmetric ECs**
(e.g., MnO₂/carbon)



- Mild-pH electrolyte
- Cation-compensated redox
- Highly reversible
- Moderate capacity
- Fast response

Toupin, Brousse Belanger,
Chem. Mater. (2004)

- Up to 1 e⁻ per Mn for
thin-film MnO₂ @ Pt

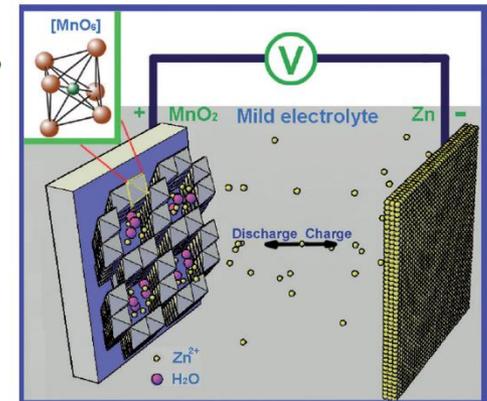
- **Primary alkaline Zn/MnO₂**



- Strong alkaline electrolyte
- H⁺-based redox @ MnO₂
- Irreversible (mostly)

- Mild-pH Zn²⁺ electrolyte
- H⁺ or Zn²⁺ redox?
- Reversible (mostly)
- High capacity

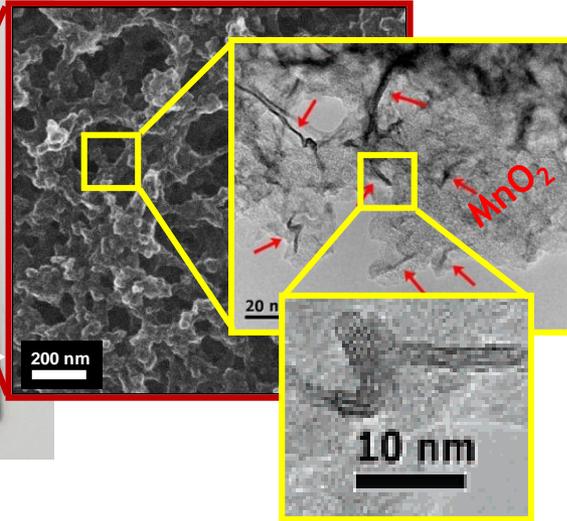
? Rechargeable “zinc-ion”



Rechargeable Zn-air: The Holy Grail of zinc batteries

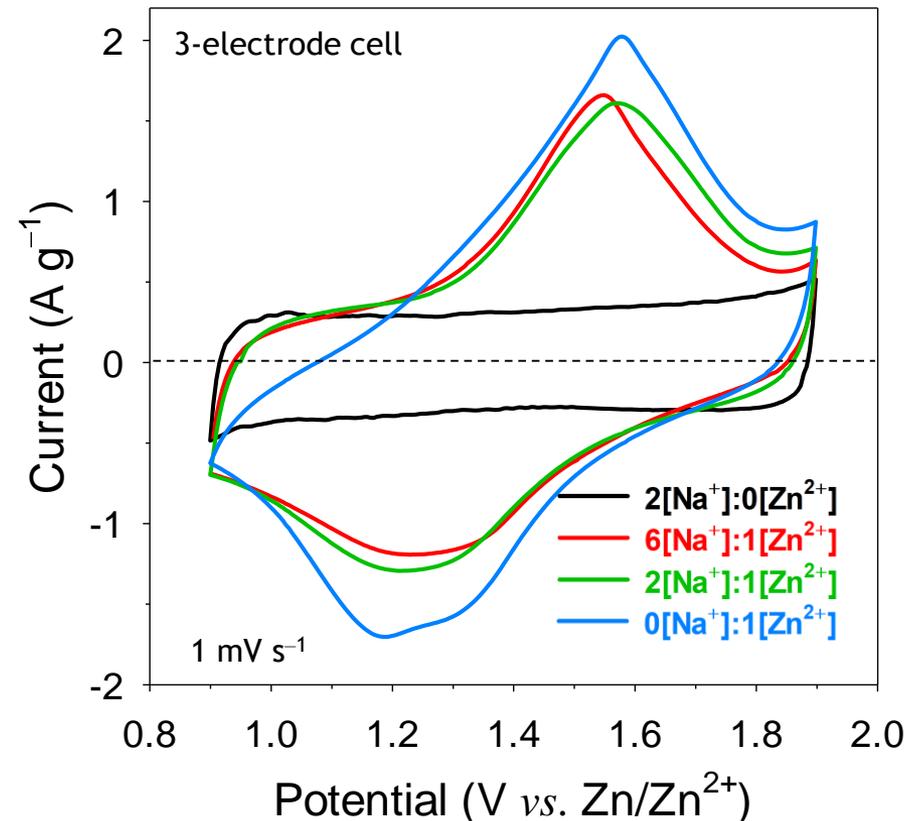
MnO_x-carbon
nanofoam

2.5 cm



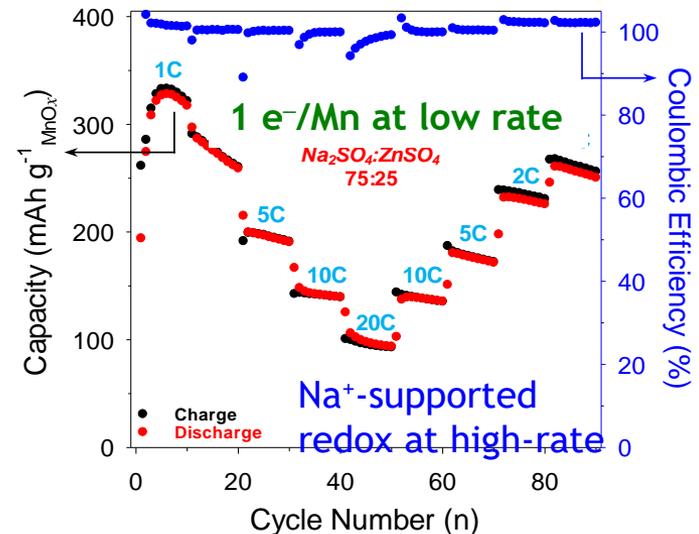
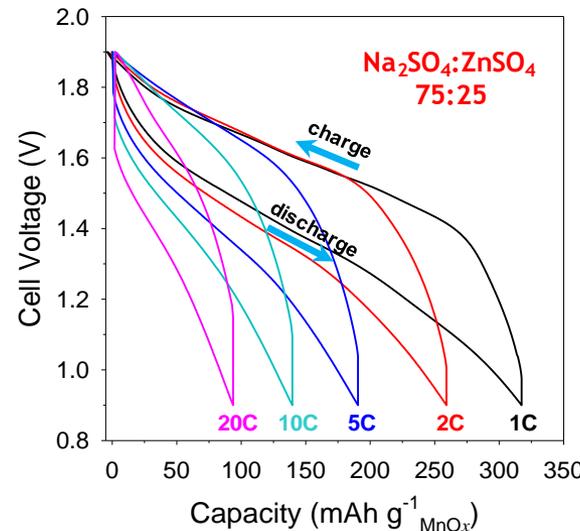
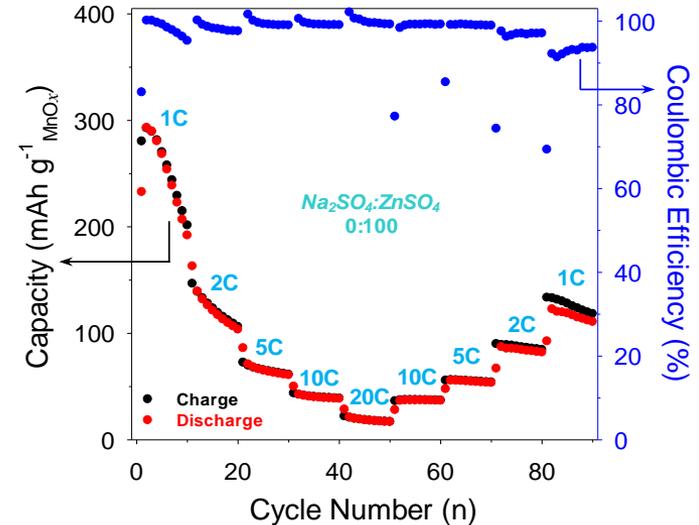
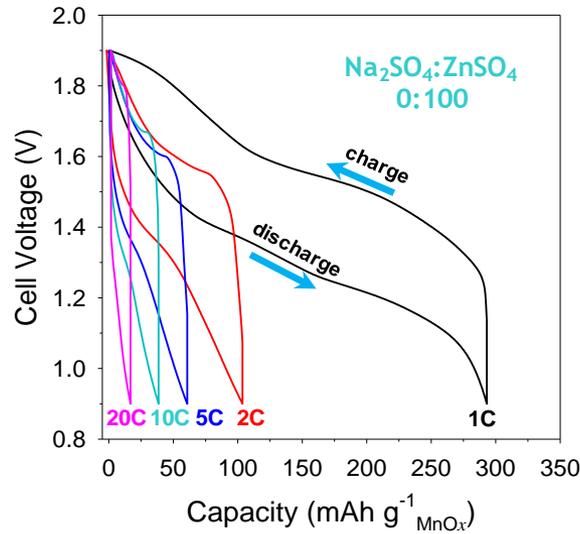
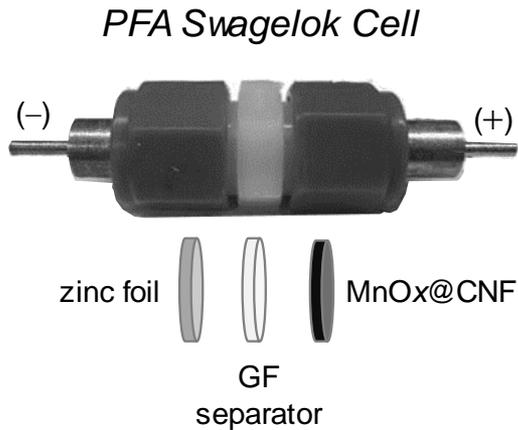
Molar concentration of electrolyte	Electrolyte notation (atom:atom ratio)
1 M ZnSO ₄	0[Na ⁺]:1[Zn ⁺]
0.5 M Na ₂ SO ₄ + 0.5 M ZnSO ₄	2[Na ⁺]:1[Zn ⁺]
0.75 M Na ₂ SO ₄ + 0.25 M ZnSO ₄	6[Na ⁺]:1[Zn ⁺]
1 M Na ₂ SO ₄	2[Na ⁺]:0[Zn ⁺]

- Partially exchange ZnSO₄ for Na₂SO₄ in aqueous electrolyte to tap MnO_x pseudocapacitance mechanisms



Rechargeable Zn-ion + power!

(power requires a cathode architecture)



Mixed Zn²⁺/Na⁺ electrolyte yields:

- ✓ Greater capacity retention with increasing rate
- ✓ Greater recovery in capacity upon return to slow rates

