

Mitigation of fine particulate matter exposures in schools

Indoor Exposure to Fine Particulate Matter and Practical Mitigation Approaches

Workshop on Mitigation of Indoor Exposure to Fine Particulate Matter
National Academies of Science and Engineering

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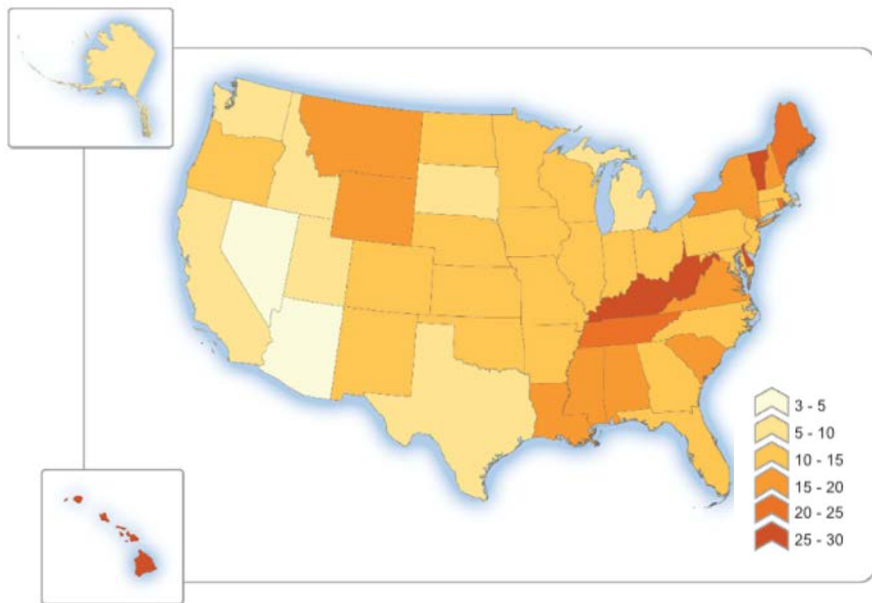
Air pollution and near-roadway schools

Schools are critical environment for susceptible population:

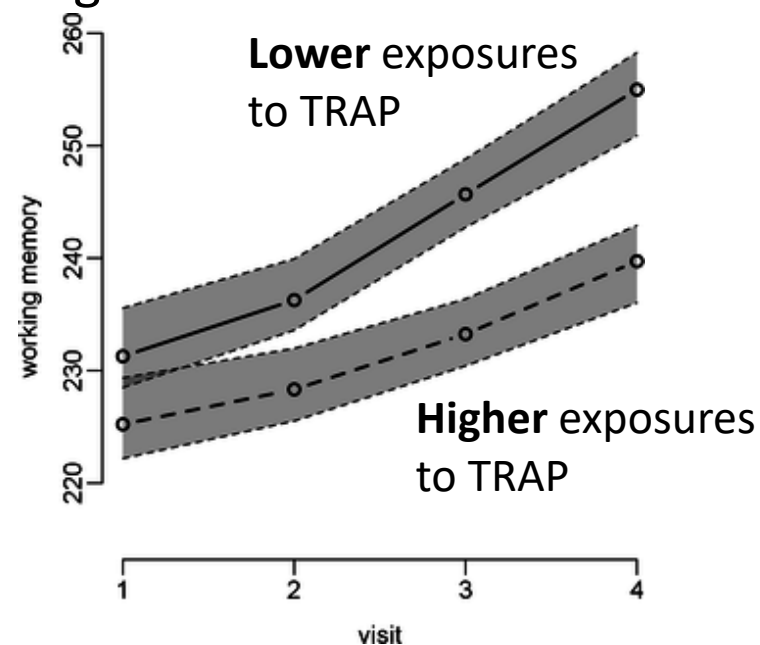
- 15% of schools (6.4 million children) < 250 m distance from major roadway¹
- Schools with higher % Hispanic, Black, Asian students have disparate exposure^{2,3}

Traffic-related air pollution (TRAP) affects student health and cognition:

- e.g., increased asthma diagnosis⁴
- Decreased working memory scores, other cognitive markers⁵



% of students attending a school within 250 m of major roadway¹



BREATHE study in Spain: 2,715 children and 10,112 tests from 39 schools.

¹Kingsley et al. 2014, JESEE 24:253-259; ²Grineski and Collins, 2018 Env. Res. 161:580-587, ³Gaffron and Niemeier 2015 Int. J. Environ. Res. Public Health (2) 2009-2021. ⁴HEI, 2010 Traffic-related air pollution: A critical review of the Literature on Emissions, Exposure, and Health Effects; ⁵Sunyer et al., 2015. PLOSone, March 2015: 1-24.

Air pollution in near-roadway schools

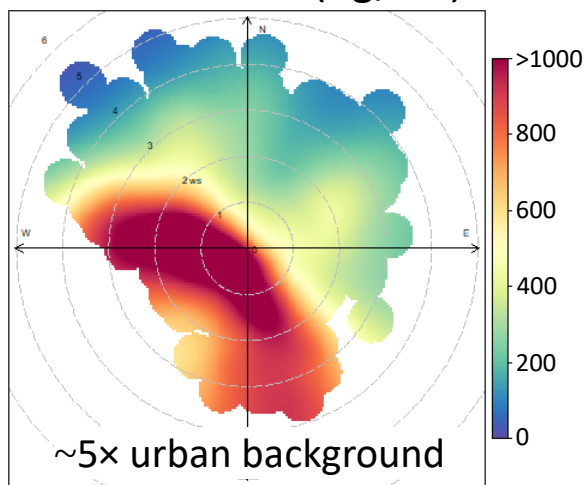


Near-roadway middle school **Harriet Tubman Middle School (HTMS)** in Portland, OR, USA; site of field study.

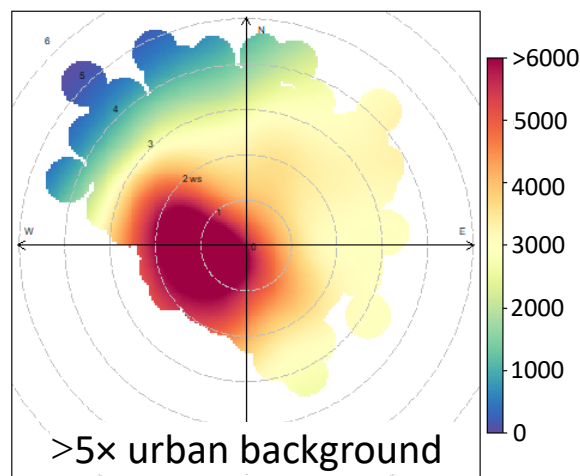
Challenges for near-roadway schools¹:

- o Elevated air pollution levels
 - Temporally and spatially variant
 - Meteorology is important
- o TRAP constituents elevated, PM_{2.5} less so
 - Health impacts greater for TRAP than PM_{2.5}²
 - Standards (ASHRAE 62.1) ↑ filtration if NAAQS exceeded
- o Indoor + site outdoors contribute to student exposure

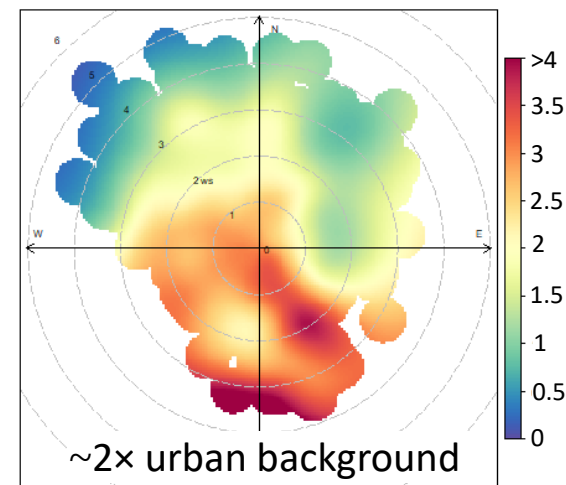
Black carbon (ng/m³)



Ultrafine particles (#/cm³)



PM_{2.5} (μg/m³)*



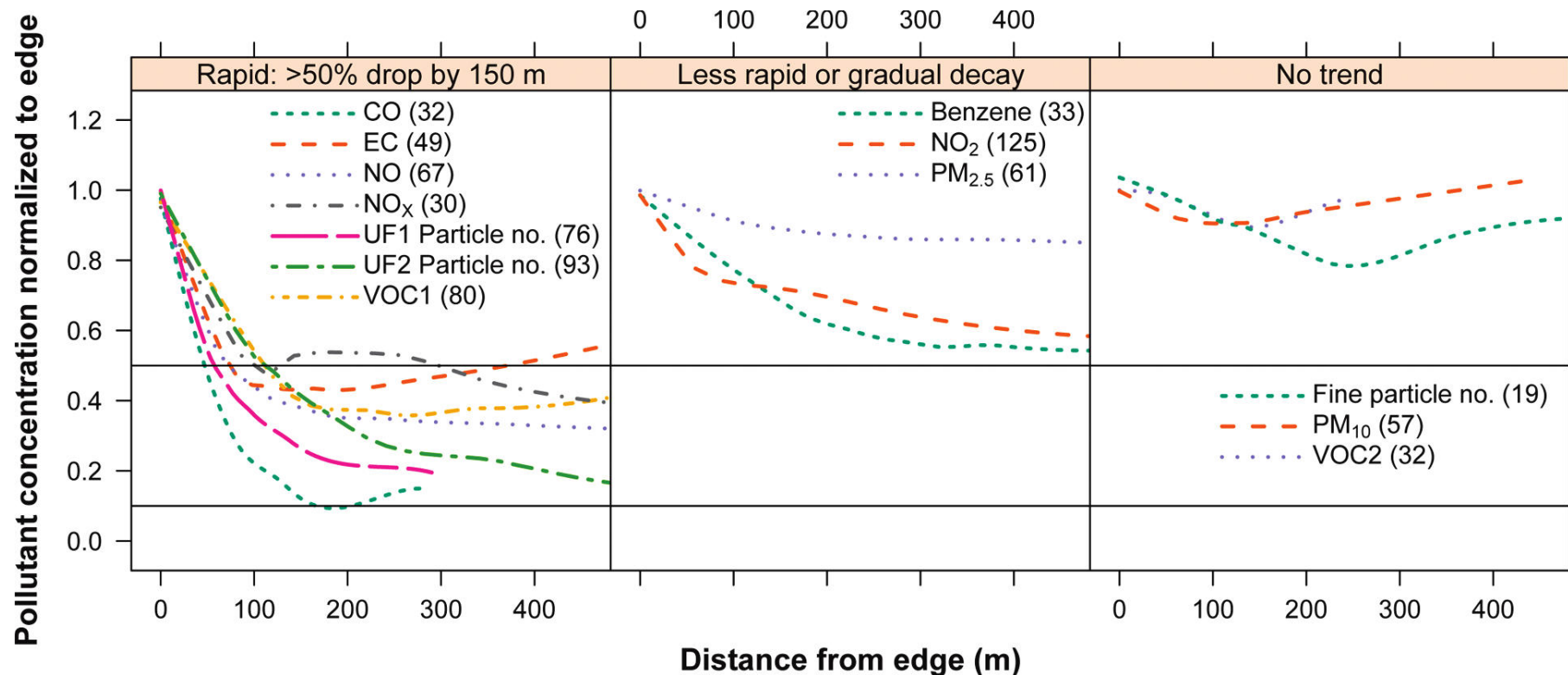
Polar plots from Spring 2019 campaign at HTMS

*w/ $\rho = 1 \text{ g/cm}^3$

¹ Gall and George, 2018, ASHRAE Journal: IEQ applications, 80-83; ² Janssen et al. 2011 EHP 119(12):1003369;

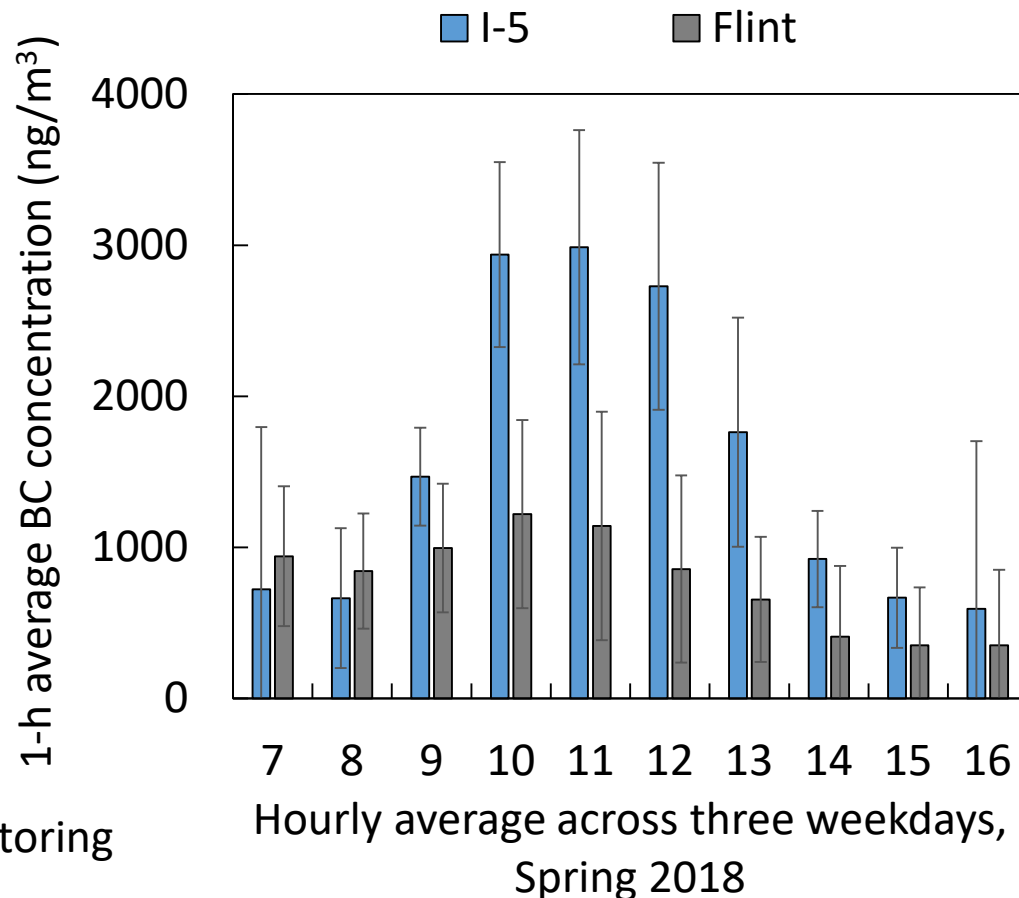
Strong spatial gradients for TRAP

- TRAP elevated above urban background within zone of $\sim 200\text{-}500\text{ m}^1$
- Zone may be 1000s of meters during nighttime²



Opportunity: leverage spatial gradient to reduce exposure

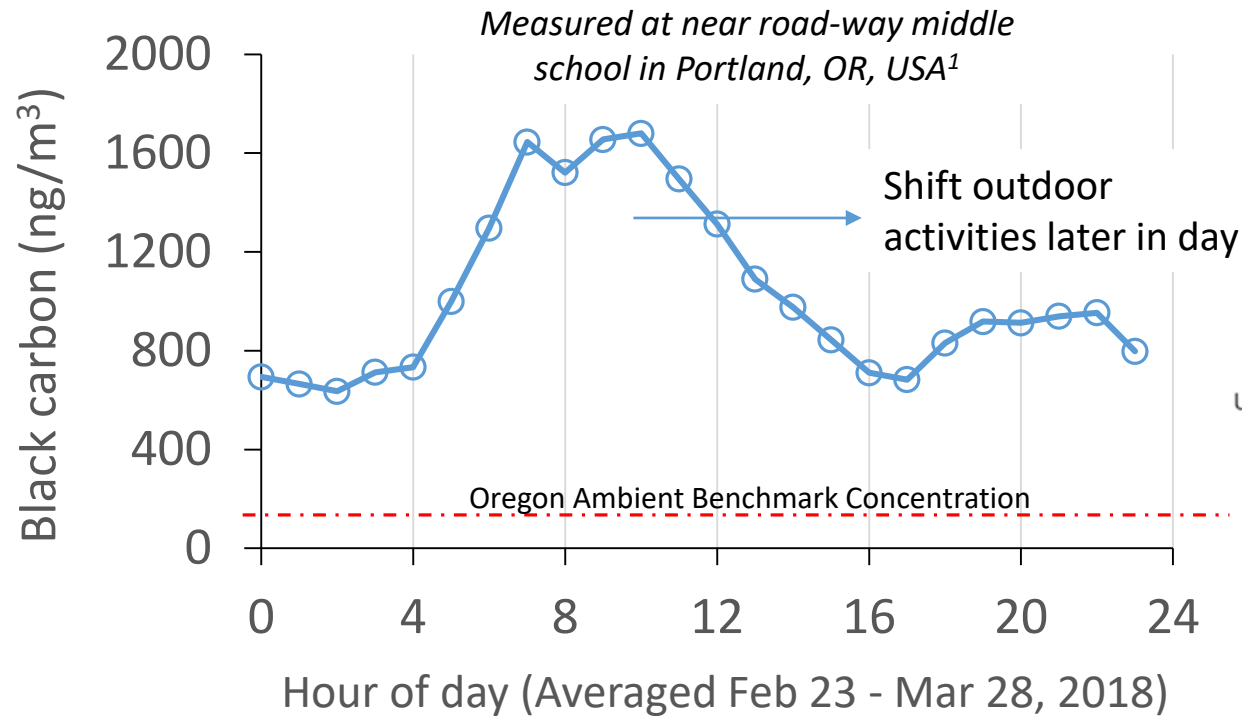
Siting outdoor air intake



- Monitoring of black carbon on two faces of Harriet Tubman Middle School¹
 - At peak periods, ↑ distance is equivalent to a ~MERV8 filter
 - Benefit is realized without energy input (due to mixing and dilution of TRAP)

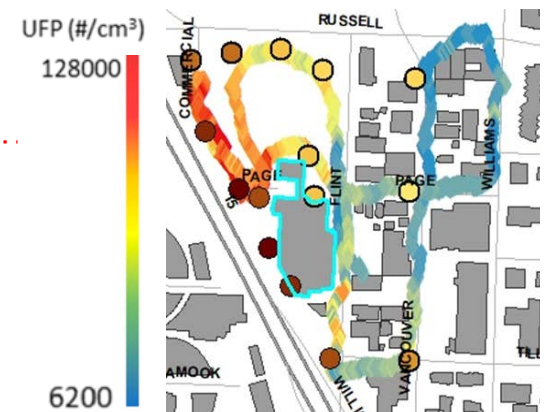
¹Laguerre et al., 2020, ES&T 54(19):11798-11808

Leverage diurnal trends in TRAP



TRAP generally peak between 7 and 9 AM due to rush hour traffic²

And shift location:



Alter timing of ventilation system³

- Test of four schools in Ottawa, ON
 1. Outdoor air ventilation from 5:30 – 6:30 AM
 2. Recirculation only until school starts at 8 or 9 AM
 → Significant reductions in UFP, VOCs for schools starting at 9 AM

¹Laguerre et al., 2020, ES&T 54(19):11798-11808; ²Touma et al. J Air Waste Manage, 56, 1716–1725.

³MacNeill et al. 2015, Indoor Air 26(5): 687-701

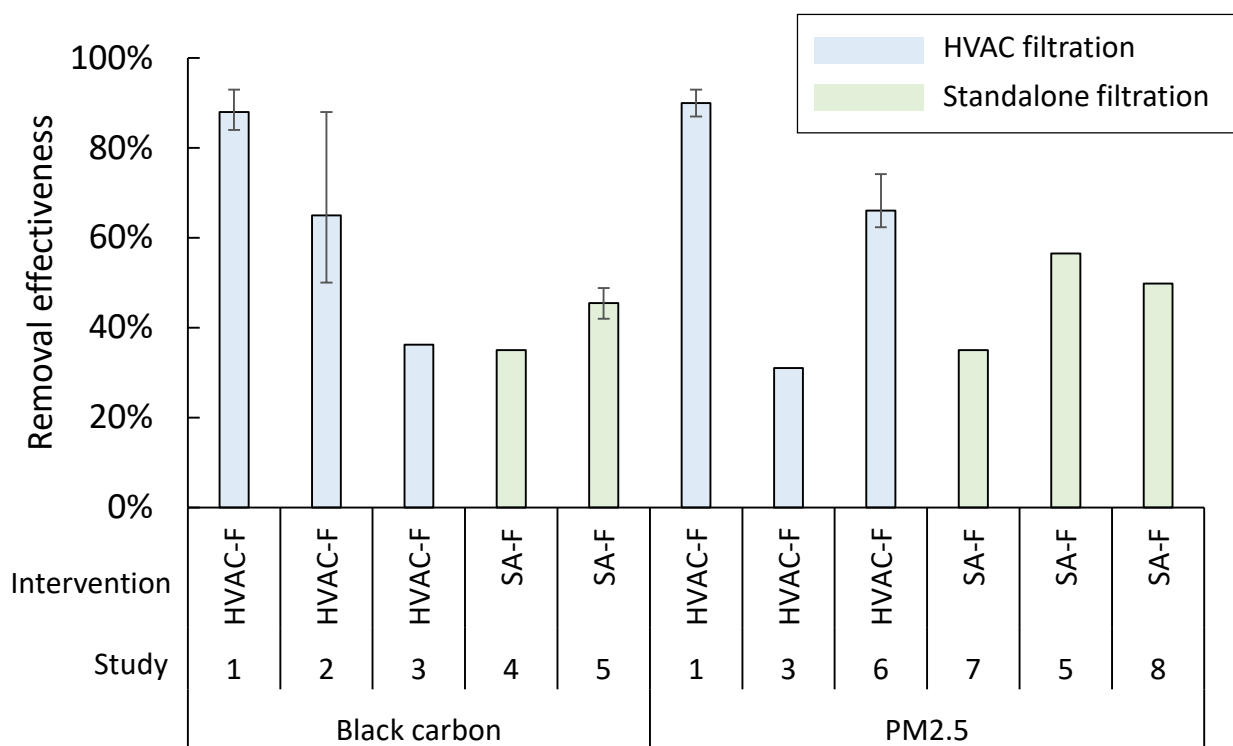
Active approaches: air-cleaning

Air cleaning effectiveness for PM in **occupied** schools

- Filtration *can be* effective, but results are variable
- Eight studies of occupied schools w/ air cleaning intervention¹⁻⁸

$$\text{Removal effectiveness} = 1 - \frac{C_{\text{controlled}}}{C_{\text{uncontrolled}}}$$

○ Most common: Filtration in HVAC system (HVAC-F), standalone (SA-F) filtration



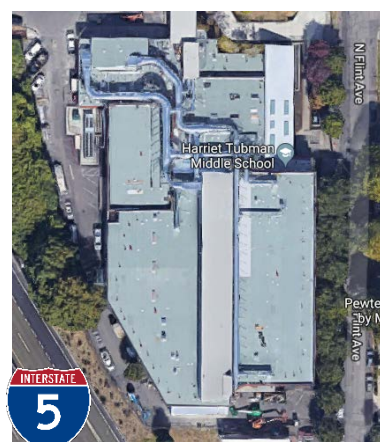
¹Polidori et al. 2013, Indoor Air 23(3): 185-195; ²McCarthy et al. 2012 Indoor Air 23:196-207; ³van der Zee et al. 2017 Indoor Air 27(2): 291-302; ⁴Scheepers et al. 2015 Environ Sci Processes Impacts 17:316-325 ; ⁵ Jhun et al. 2017 J Allergy Clin. Immun. 5(1): 159-159e.3; ⁶ Gao et al. 2019 Env. Res 197 Part A, 108749; ⁷Park et al. 2020 Build. Environ 167:106437; ⁸Smythe, A. 2018 Master's Thesis, Harvard University

At near-roadway school in Portland, OR

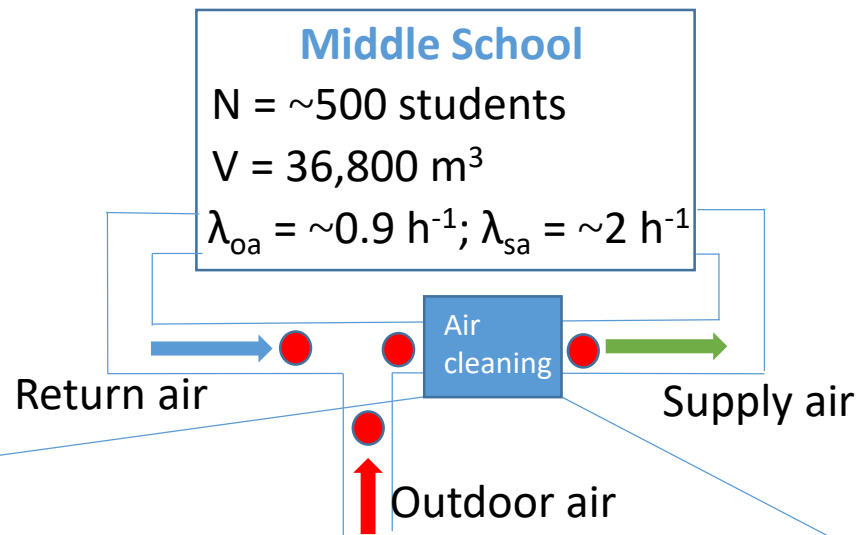
Pre-renovation



Post-renovation



● Air monitoring locations¹



MERV8 pre-filter

protect lifespan of downstream high efficiency filter + AC downstream



MERV16 filter

effective across broad range of particles, $<10 \text{ nm} - 10+ \mu\text{m}$

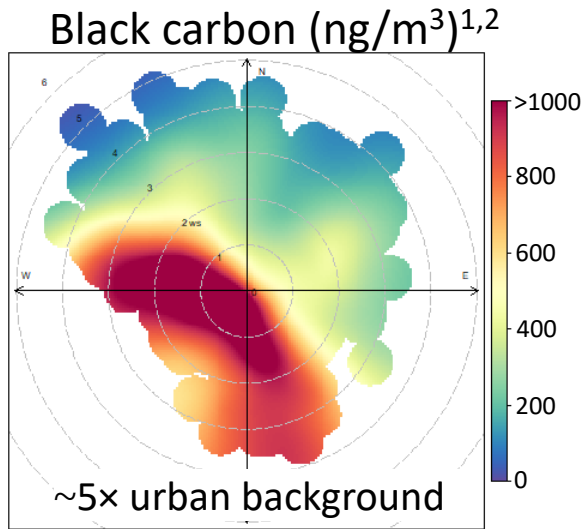


Functionalized carbon
 VOC and NO_2 removal, large mass required, $\uparrow$ cost

¹See Laguerre et al., 2020, ES&T 54(19):11798-11808 for description of monitoring activities

Address TRAP in ventilation air

- Minimum goal for near-roadway school:
 - o TRAP in ventilation air = to urban background w/ standard filtration



η = black carbon removal efficiency (-)
 λ_{OA} = outdoor air exchange rate (h^{-1}), = 0.85 h^{-1}
 C = black carbon outdoor air concentration ($\frac{\text{ng}}{\text{m}^3}$)

Set black carbon source in school ventilation air so that:
 If near-roadway (NR) = If in urban background (BG)

$$[(1 - \eta)\lambda_{OA}C]_{NR} = [(1 - \eta)\lambda_{OA}C]_{BG}$$

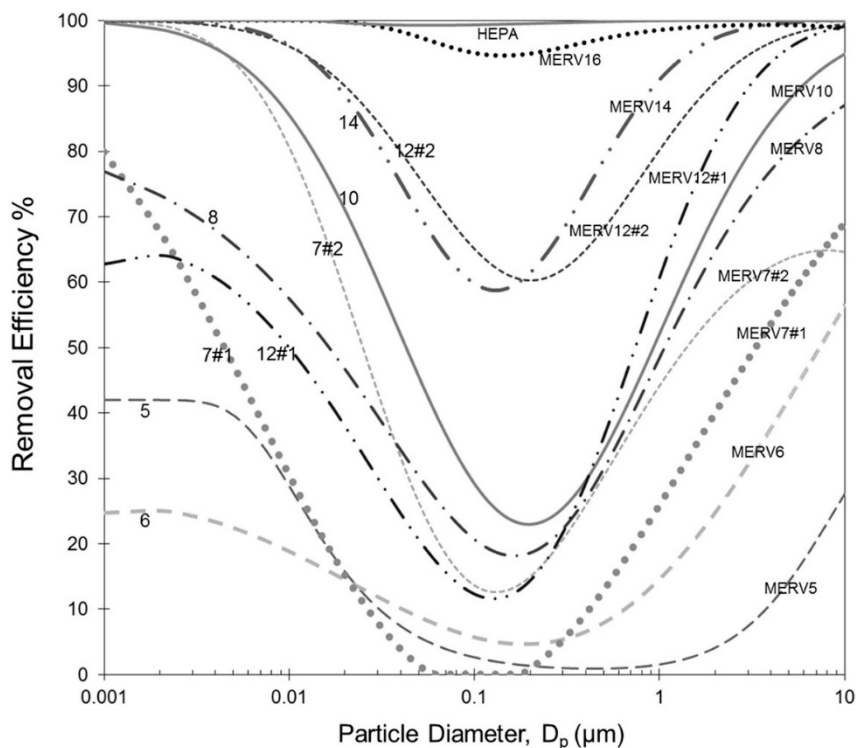
$$\eta_{NR} = 1 - \frac{C_{BG}}{C_{NR}} (1 - \eta_{BG})$$

$$\eta_{NR} = 1 - \frac{1}{5} (1 - 0.2)$$

$\eta \geq 84\%$ required to make BC outdoor source similar to that of urban background w/ typical filter (at near-roadway school in Portland, OR USA^{1,2})

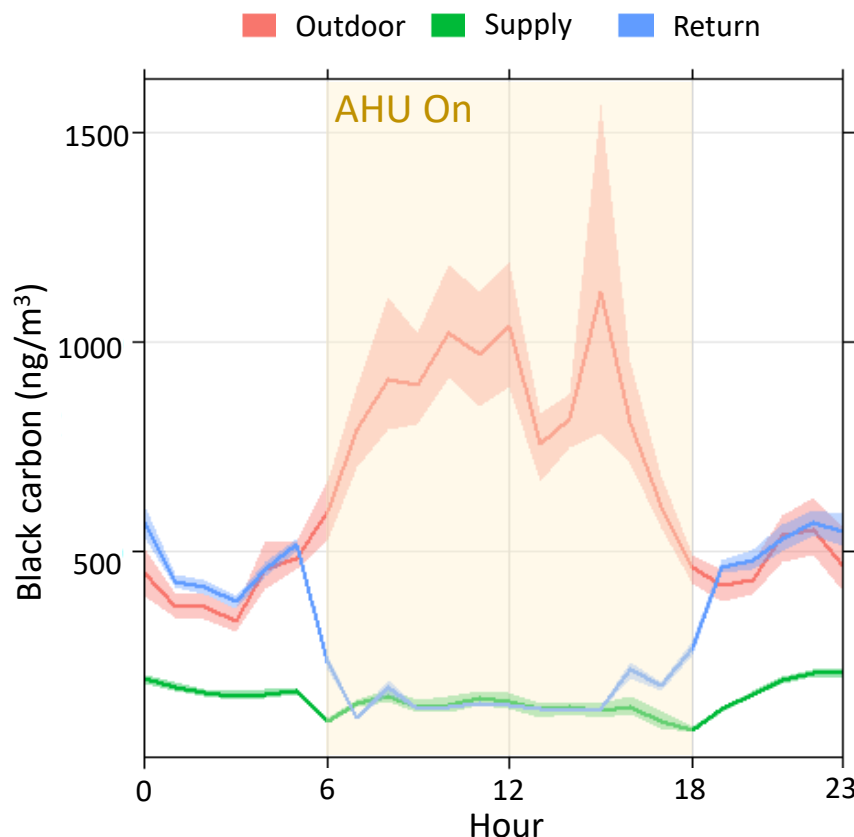
Efficacy of high-efficiency air cleaning

Predictions of size-resolved removal efficiency vs. MERV rating^{1,2}



>84% removal efficiency of fine particulate matter achievable, $f(\text{MERV}, d_p)$

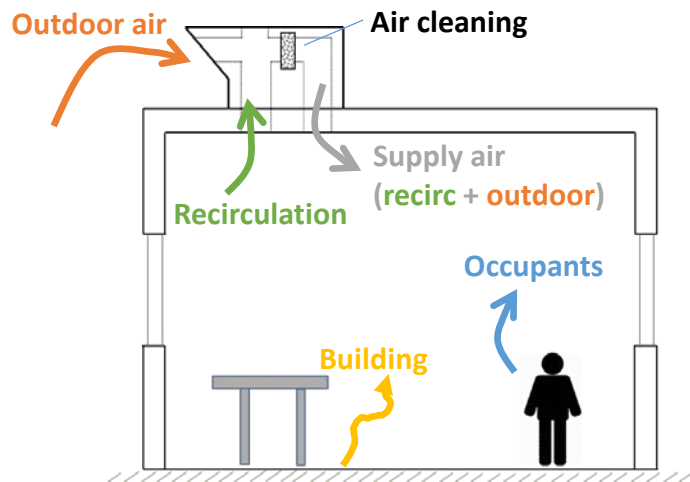
1-week at near-roadway middle school³, w/ MERV 8 + 16



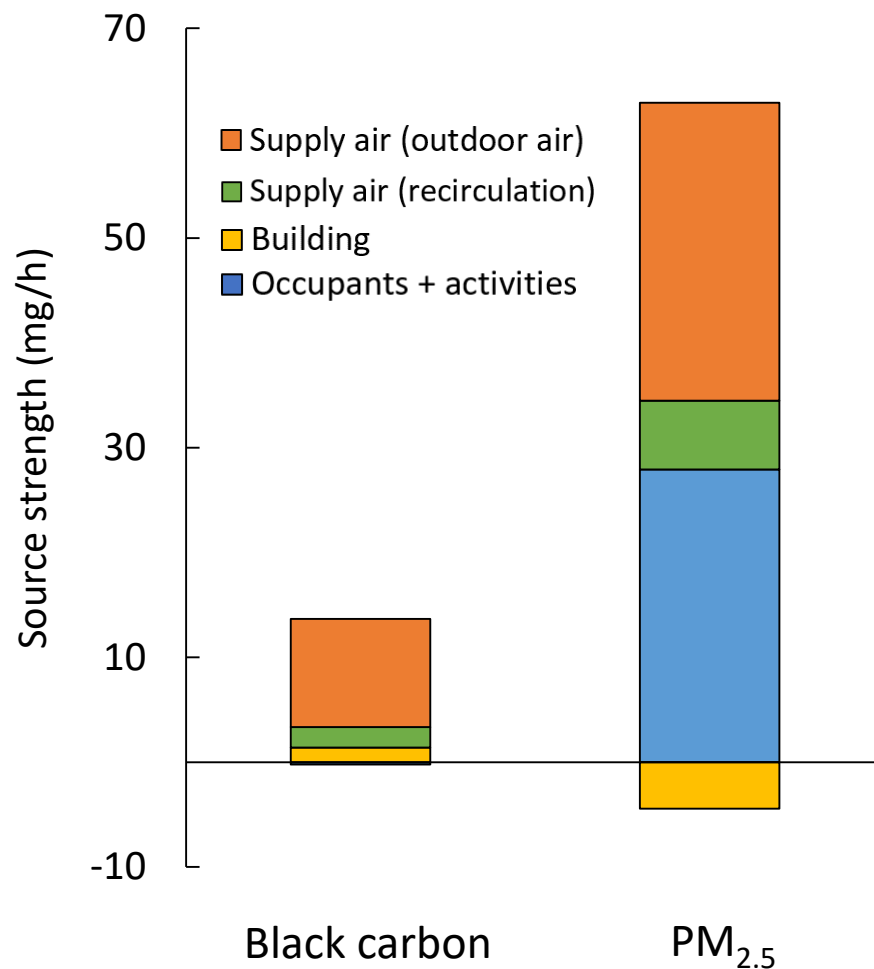
Blue line: Indoor BC levels avg. $150 \text{ ng}/\text{m}^3$ w/ air cleaning; **effectiveness of ~85%**

Indoor sources of fine PM in schools

Occupants and their activities generate fine particulate matter



Preliminary source apportionment of fine particulate matter at HTMS¹:



- Indoor sources important for PM_{2.5}
 - PM_{2.5} indoor emission at HTMS:
 $\sim 70 \frac{\mu g}{h \text{ person}}$ (preliminary calculation)
- For black carbon:
 - No source from occupants + activities
 - Net source from building implies some BC penetrates envelope and persists

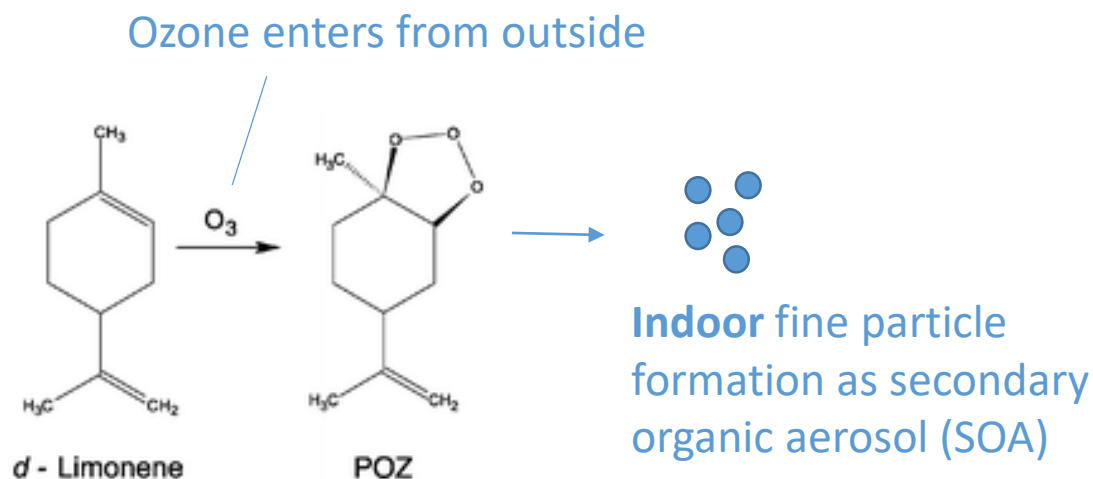
¹method based on as shown in Tang et al., 2016, Environmental Science & Technology 50(23): 12686-12694

Indoor VOC chemistry → particles

- Some volatile organic compounds of indoor origin are reactive
 - o Limonene is the dominant indoor terpene¹
 - o Limonene has high aerosol formation potential²
 - o Ozone – monoterpene chemistry occurs on time-scale relevant to indoor air³
 - o Contributes to ultrafine and accumulation mode PM; possible health impact⁴

Monoterpenes:
compounds with molecular
formula $C_{10}H_{16}$.¹
Over 1000 different
compounds

Limonene, a monoterpene,
with indoor sources



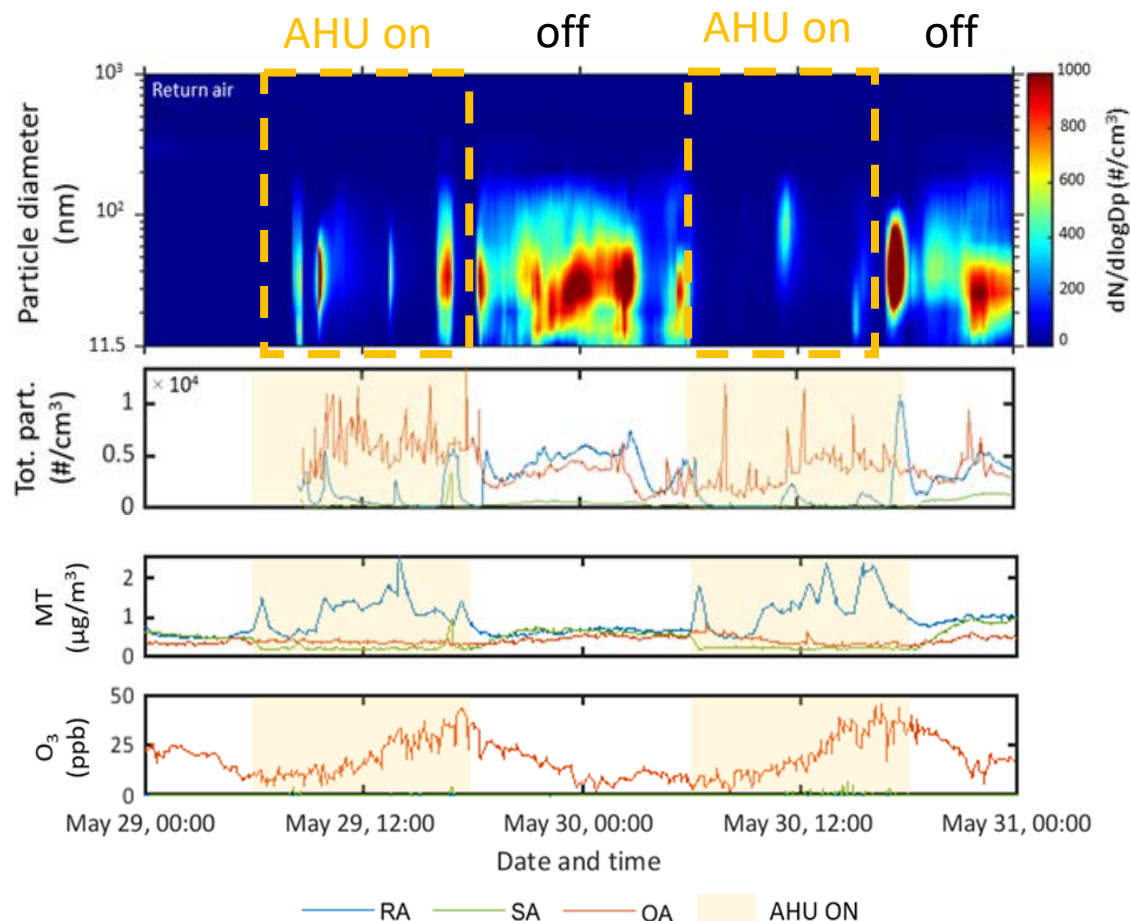
Air-cleaning alters indoor chemistry

Carbon scrubber appears to reduce indoor secondary organic aerosol formation¹

AHU on → air cleaning → low indoor O₃ and lower monoterpene → **lower SOA formation**

AHU off → no air cleaning → higher O₃ and higher monoterpene → **higher SOA formation**

81% reduction in
SOA source strength
 $\Delta = 3\text{-}5 \text{ mg/h}$



¹Laguerre et al., 2020, ES&T 54(19):11798-11808

Fine PM in near-roadway schools

Opportunities to reduce fine particle exposures in near-roadway schools:

1. ↑ distance from outdoor sources
2. Alter timing of activities
3. Air-cleaning
4. Address indoor sources
5. Quench indoor chemistry via air-cleaning and/or source reduction

Research and resource needs for schools:

1. Data on efficacy of installed interventions
2. Lower energy + maintenance methods for ventilation, air-cleaning
3. Research on PM source strengths in schools
4. Health impact of exposures to PM of indoor origin

People:

- Aurelie Laguerre, MS, Portland State University
- Dr. Linda George, Portland State University
- Brett Stinson, Portland State University

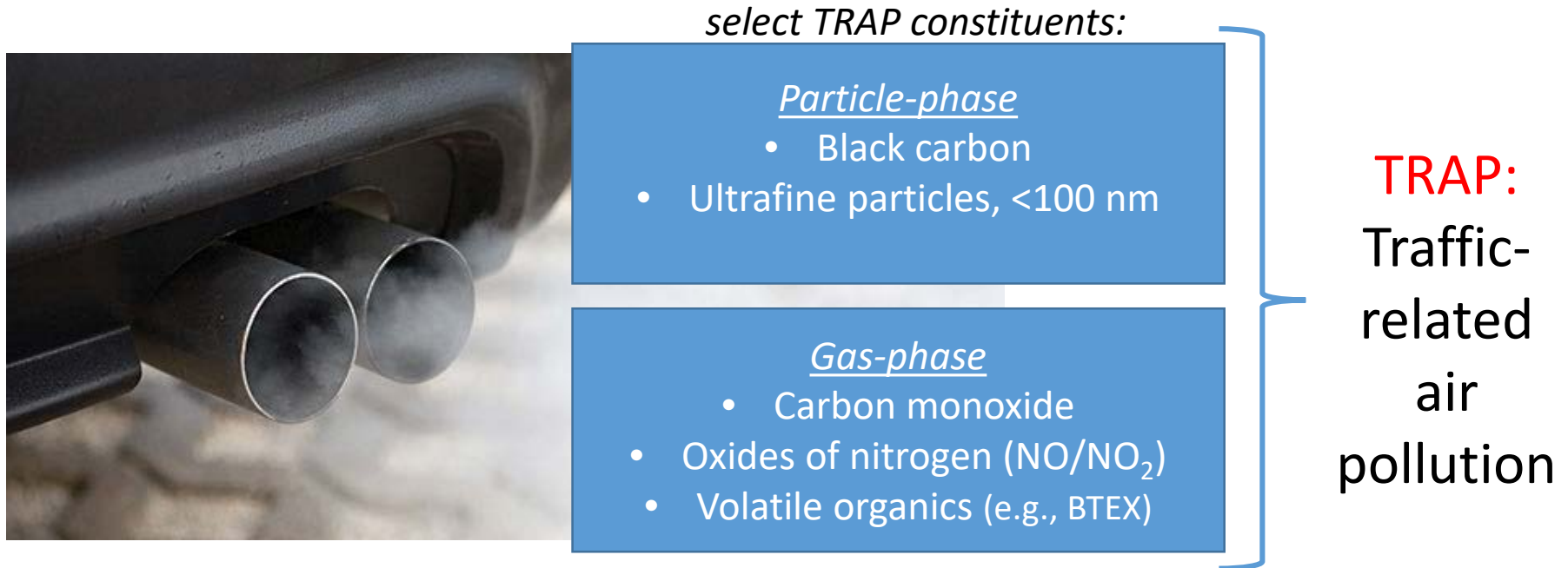
Funding:

- Portland Public School District
- This material is based upon work supported by the National Science Foundation under grant DUE #2037582

Supporting slides

Traffic related air pollution

Vehicles emit products of incomplete combustion:

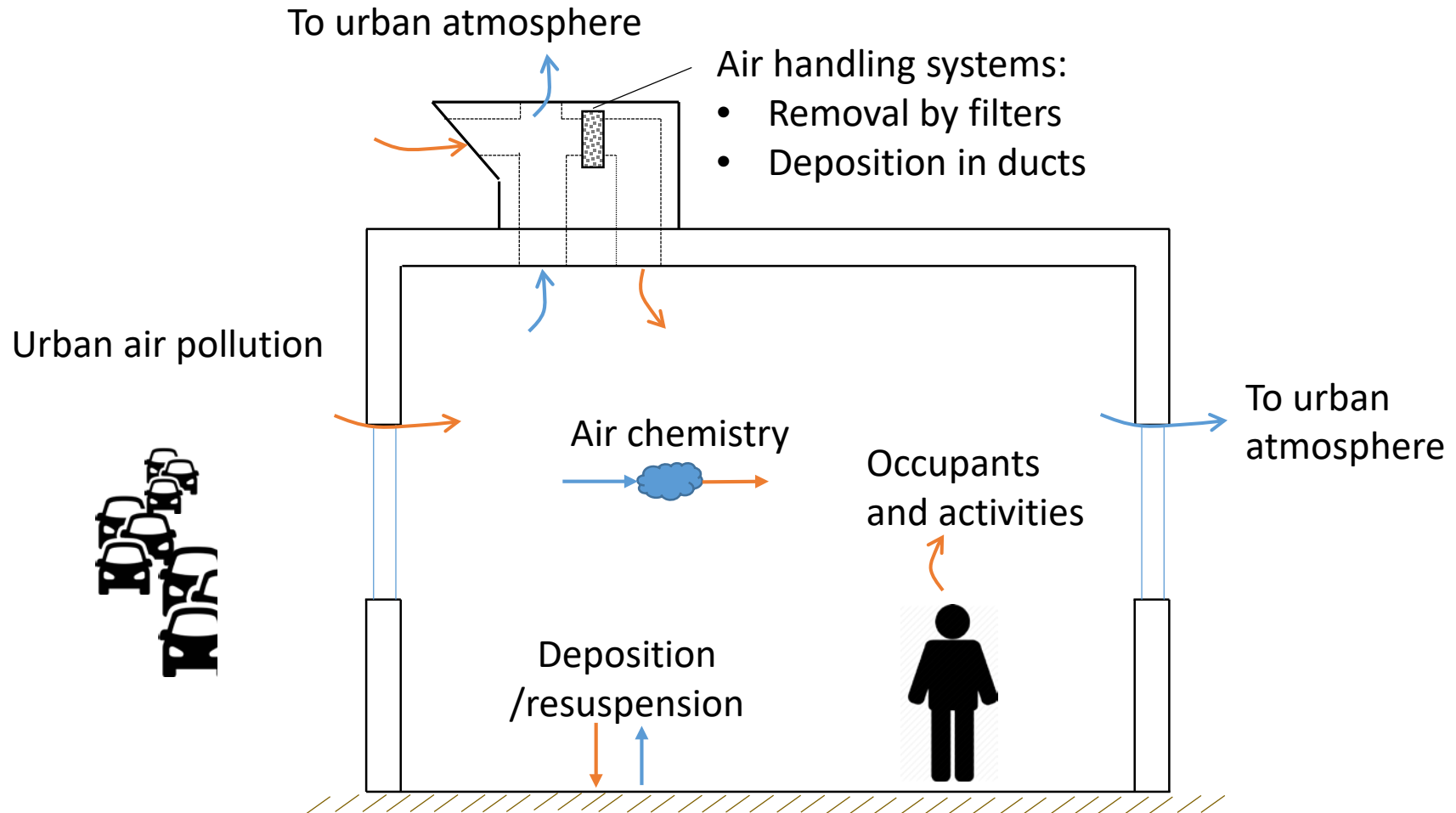


Black carbon (BC): carbonaceous product of incomplete combustion¹

Black carbon is a substantial fraction of:

- i) PM_{2.5} in traffic environments² ii) vehicle PM emission factor³

Indoor particle dynamics



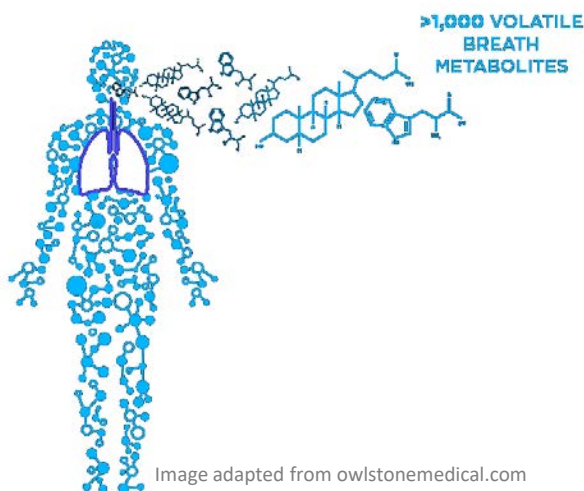
$$\frac{dC_{fp}}{dt} = S - LC_{fp}$$

To apply mass conservation principles, we require parameterization of sources (S , $\mu\text{g m}^{-3} \text{h}^{-1}$) and losses (L , h^{-1}) to indoor control volume

Humans are major source of VOCs

"...building's occupants, simply by being present, significantly change the air chemistry inside..."

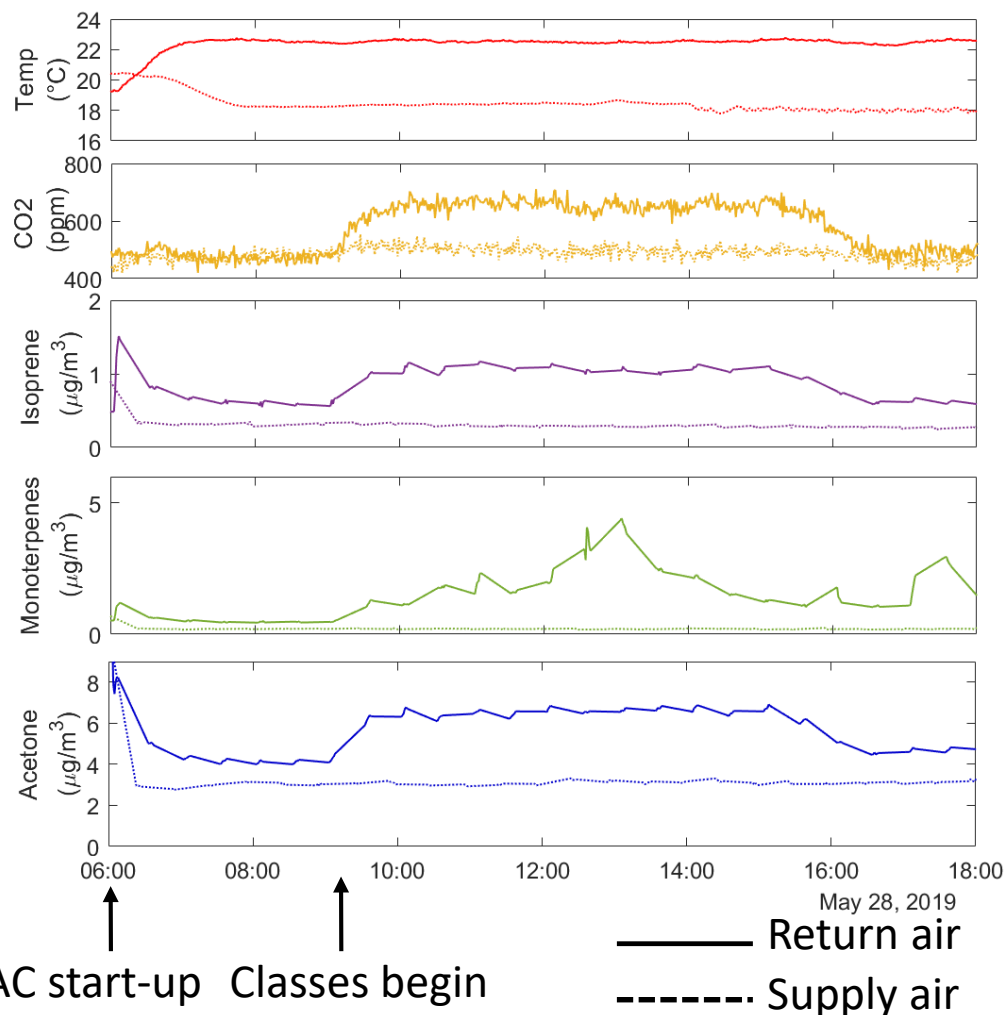
<https://www.pnas.org/content/117/37/22619>



Humans emit VOCs that:

- degrade perceptions of indoor air
- engage in chemistry and become harmful to health

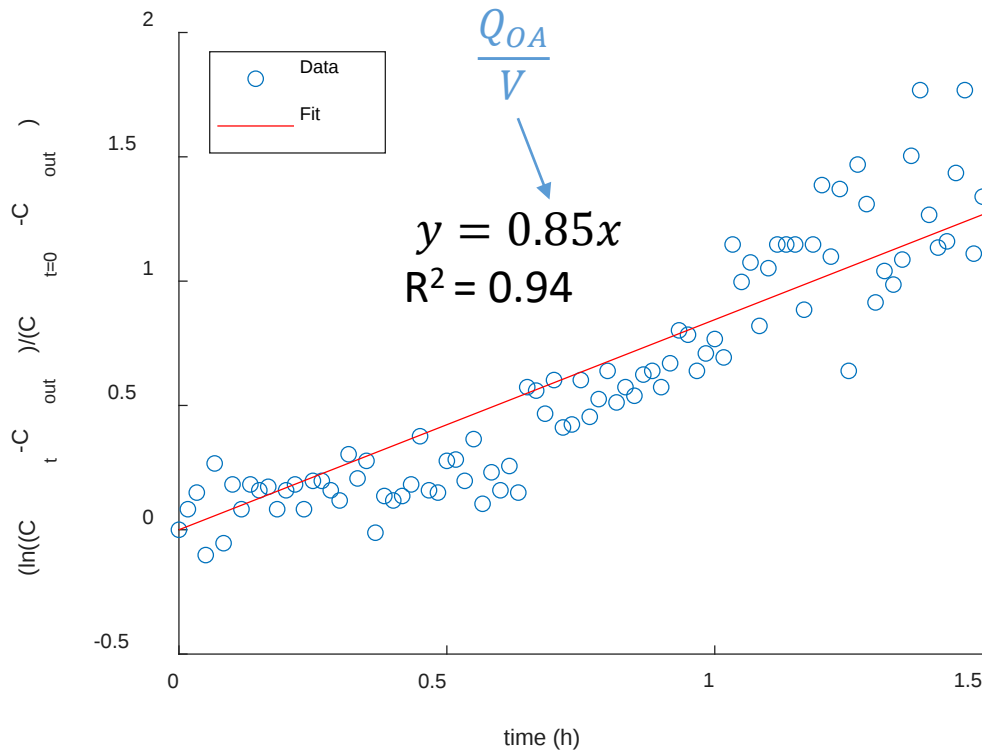
CO₂ and VOCs in Harriet Tubman Middle School



Ventilation and occupancy

1

Analyze CO₂ decay occurring at end of day post occupancy, air handler operating



2

Use steady-state period to estimate # of students*

CO₂ above ambient is due to humans:

$$N = \frac{(C_{CO2,RA} - C_{CO2,OA})Q_{OA}}{EF_{CO2}}$$

$$N = 447 \text{ people}$$

Compares well with school district records: 515 people

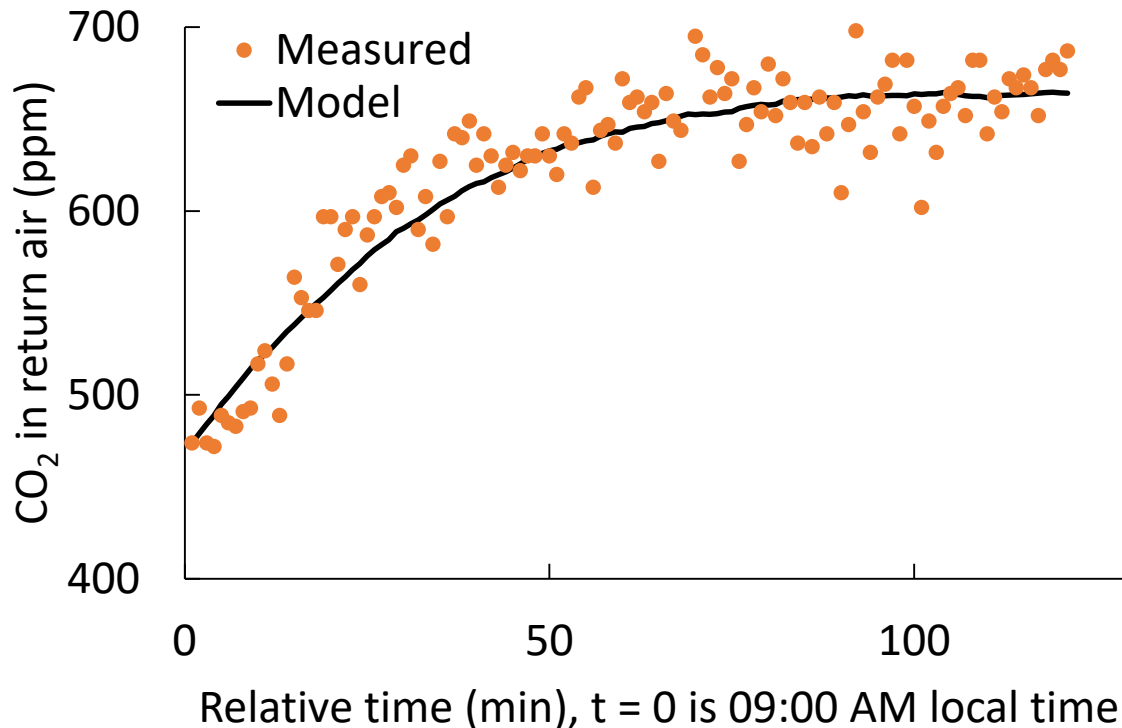
*Assumes Q_{OA} from end of day applies to stable occupancy period

*CO₂ emission factor is weighted average of students aged 11-16 (weighted equally for males, females) and staff from Persily et al.¹

Estimating school airflows

3 Analyze accumulation period determine λ

$$\frac{dC}{dt} = \frac{Q_{SA}}{V} C_{CO2,SA}(t) - \frac{Q_{SA}}{V} C_{CO2,RA}(t) + \frac{EF \times N}{V}$$



1. Discretize
2. Solve for best fit Q_{SA} by minimizing residuals (SSE)

$$Q_{SA} = 43,000 \text{ ft}^3/\text{min}$$

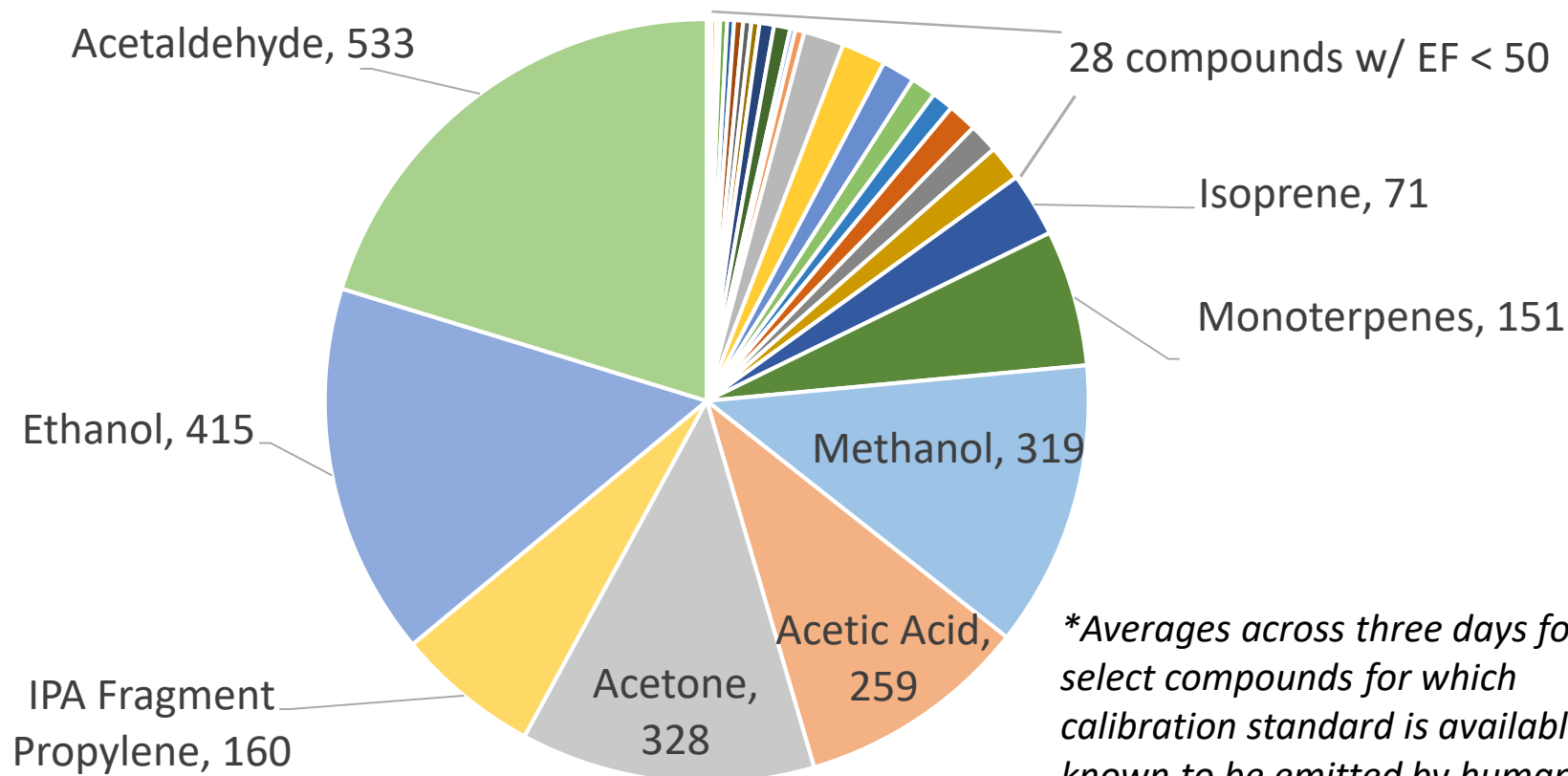
Consistent w/ mechanical contractor's estimate 40,000 – 65,000 ft³/min and Q_{SA} from steady-state flow balance on AHU

Method may be useful for buildings that are occupied, access to air handler only

$$EF = \frac{\left(\frac{M_{emitted}}{\Delta t}\right)_{occupied} - \left(\frac{M_{emitted}}{\Delta t}\right)_{vacant}}{\# \text{ of people}}$$

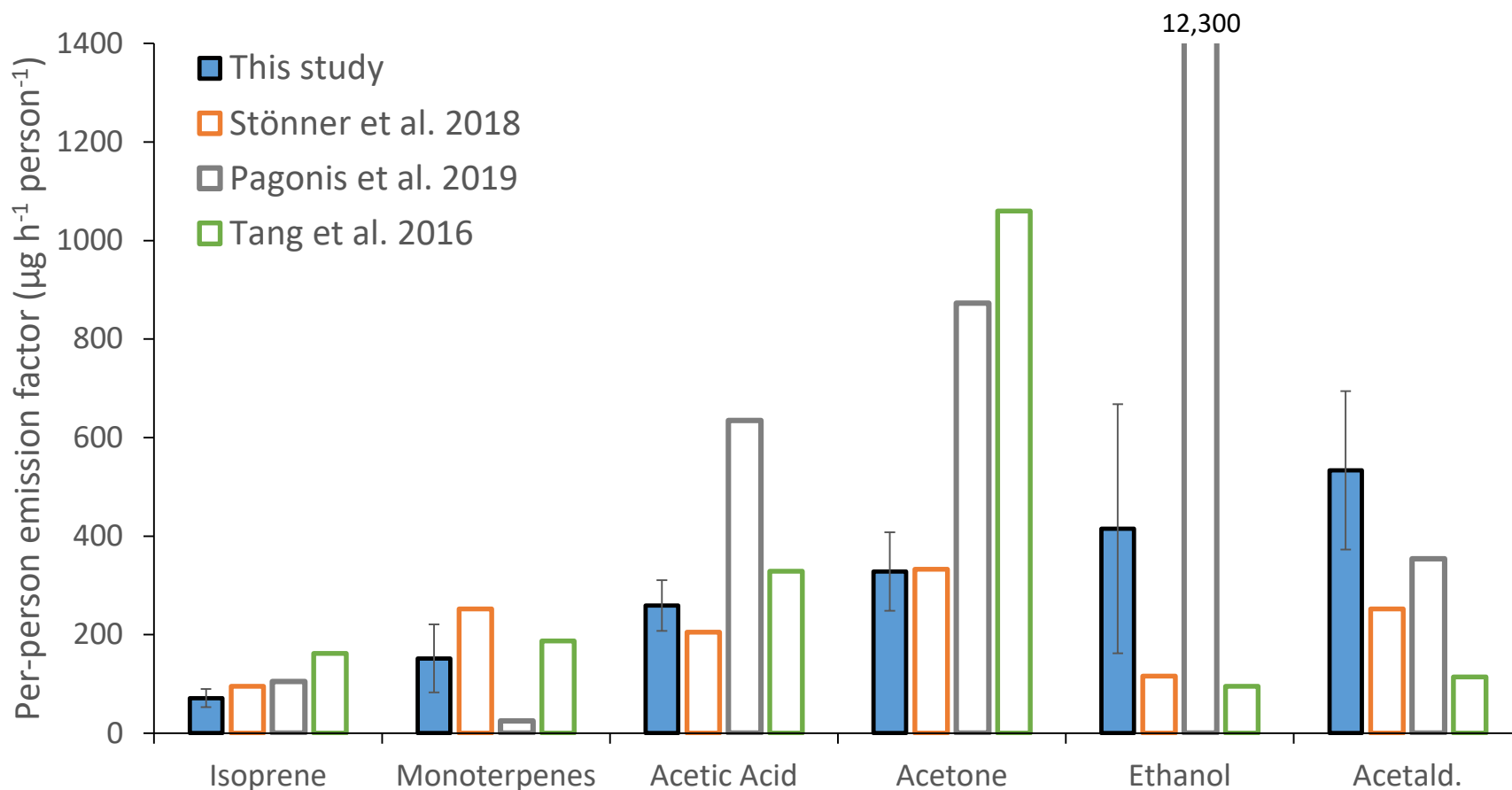
- Monday: Federal holiday as control
- Tues, Wed, Friday: all met criteria for analysis of flows, occupancy calcs

Per-person emission factors* ($\mu\text{g h}^{-1} \text{ person}^{-1}$)



**Averages across three days for select compounds for which calibration standard is available or known to be emitted by humans*

EFs are scarce, esp. for K-12 schools



Fate of indoor emissions

- Chemistry & partitioning¹ to surfaces
- Emissions outdoors?
 - o Monoterpenes are well studied in outdoor air
 - o Limonene emission factor (area): school vs. plants



Area emission factor

$$= \frac{92,000 \mu g h^{-1}}{5800 m^2}$$

= 16 $\mu g m^{-2} h^{-1}$
for a middle school campus

5800 m² area

Needle leaf²: 99 $\frac{\mu g}{m^2 h}$



Broadleaf²: 41 $\frac{\mu g}{m^2 h}$



Delosperma (sedum)³: 5 $\frac{\mu g}{m^2 h}$



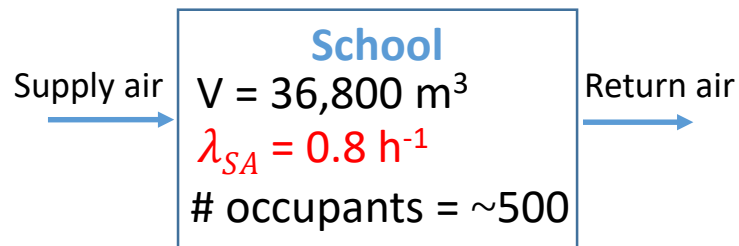
In cities, large building footprints.

- Indoor emissions can be released outdoors and affect urban + regional air quality!

¹Wang et al. 2020 Science Advances 6(8):eaay8973 ²Sakulyanontvittaya et al. 2008, ES&T 42(5):1623-1629; ³Research in progress at PSU

Characterizing indoor emission rates

Mass, M , of compound emitted into school¹

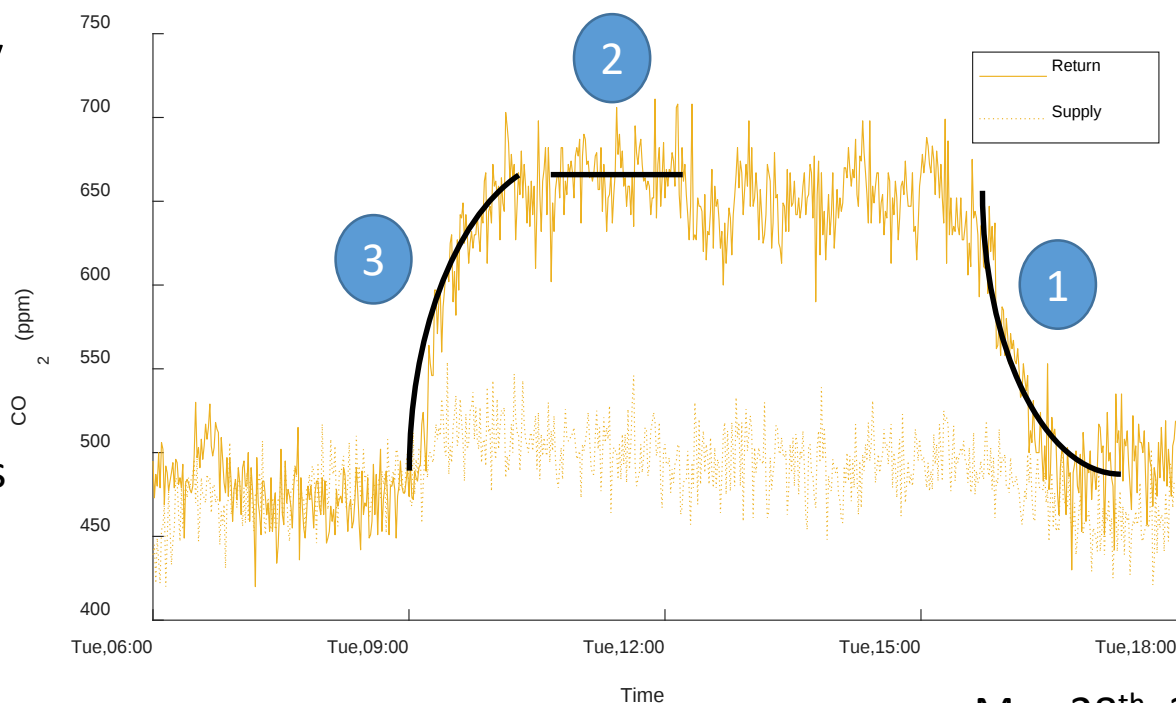


$$M = \lambda_{SA} V \int_{t_0}^{t_1} (C_{return} - C_{supply}) dt + V \int_{t_0}^{t_1} dC_{return}$$

Flowrate to building (SA = supply air) is unknown

- Could not perform intentional tracer decay
- Developing method for λ_{SA} from analysis of:

- 1 λ_{OA} (outdoor air) via “natural” decay
- 2 Use λ_{OA} w/ steady-state to estimate # of students
- 3 Use # students and the CO_2 emission rate to determine λ_{SA}



May 28th, 2019

¹method based on as shown in Tang et al., 2016, Environmental Science & Technology 50(23): 12686-12694

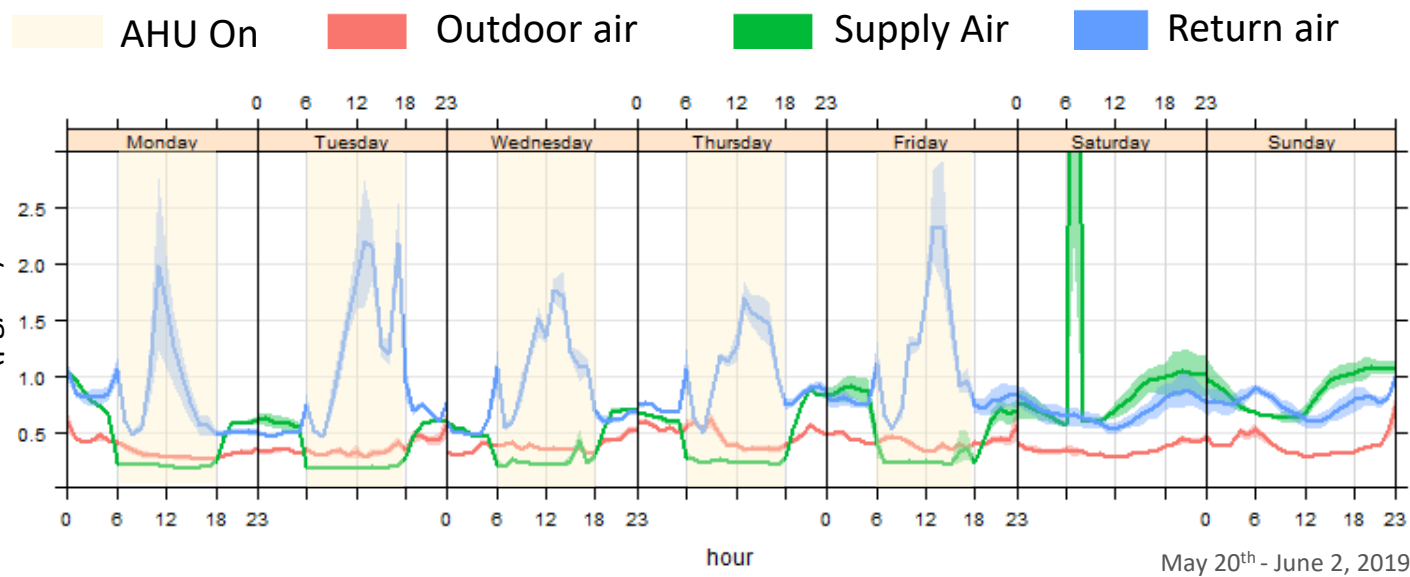
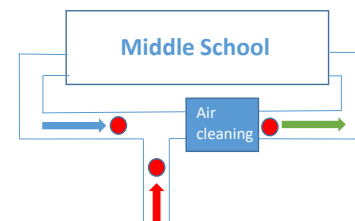
Indoor VOC dynamics at HTMS

VOC monitoring at HTMS shows:

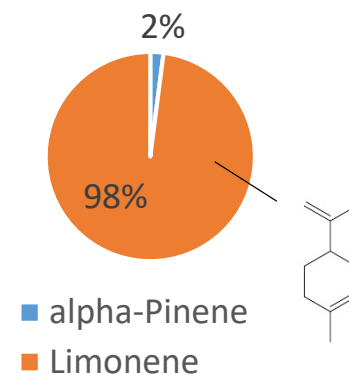
- Indoor VOC concentrations are dynamic
- Humans are an important source of monoterpenes to the space
- Most of the indoor monoterpene signal likely limonene



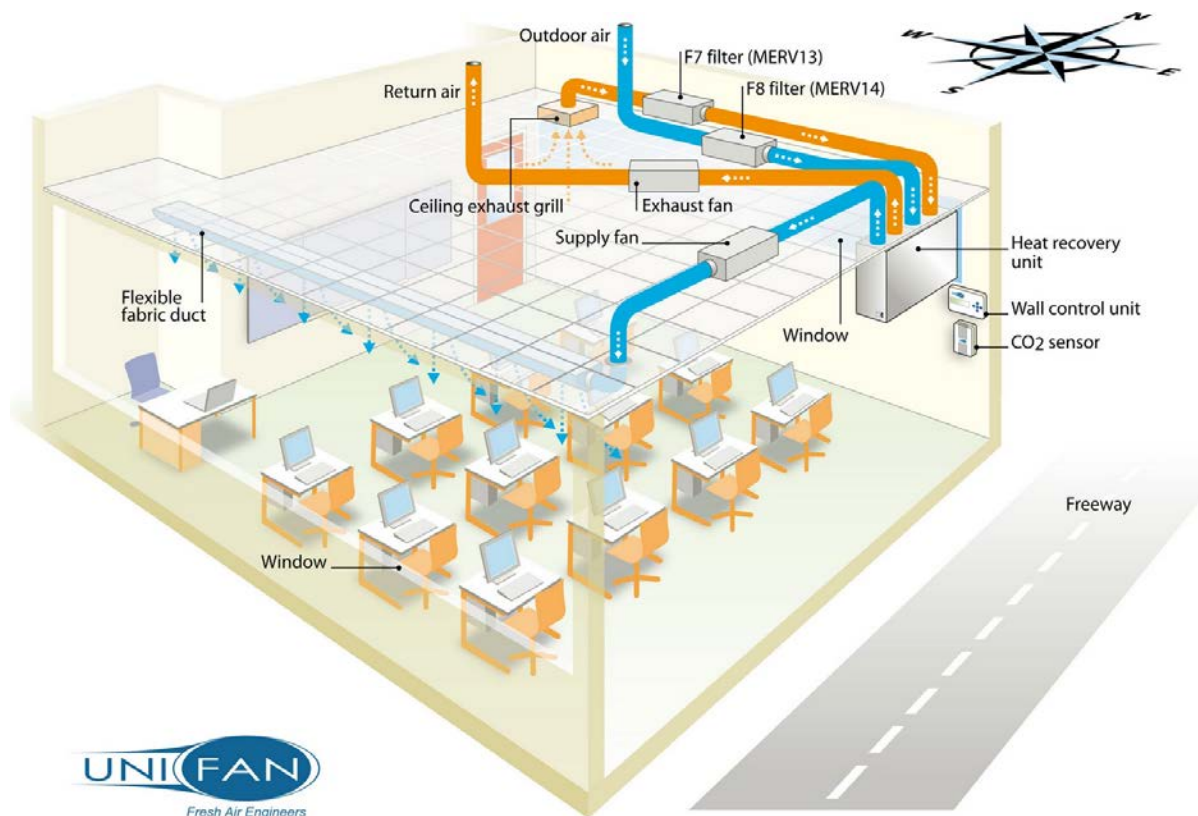
PTR-MS, sampling manifold to three locations in AHU



From TD-GC-MS analysis of return air on May 29th, 2019, of 1.9 µg/m³ measured:



Air cleaning efficacy in occupied schools



Air-cleaning and ventilation (MERV13/14) system implemented in near-roadway school in Amsterdam¹

- Reduced I/O ratio of BC by, on average, 36%
- Authors suggest high infiltration, recommend locating schools far from freeways

Bipolar Ionization/Corona Discharge/ Needlepoint Ionization and Other Ion or Reactive Oxygen Air Cleaners

ASHRAE: Convincing scientifically-rigorous, peer-reviewed studies do not currently exist on this emerging technology; manufacturer data should be carefully considered.¹

May range from “ineffective” to “very effective”²

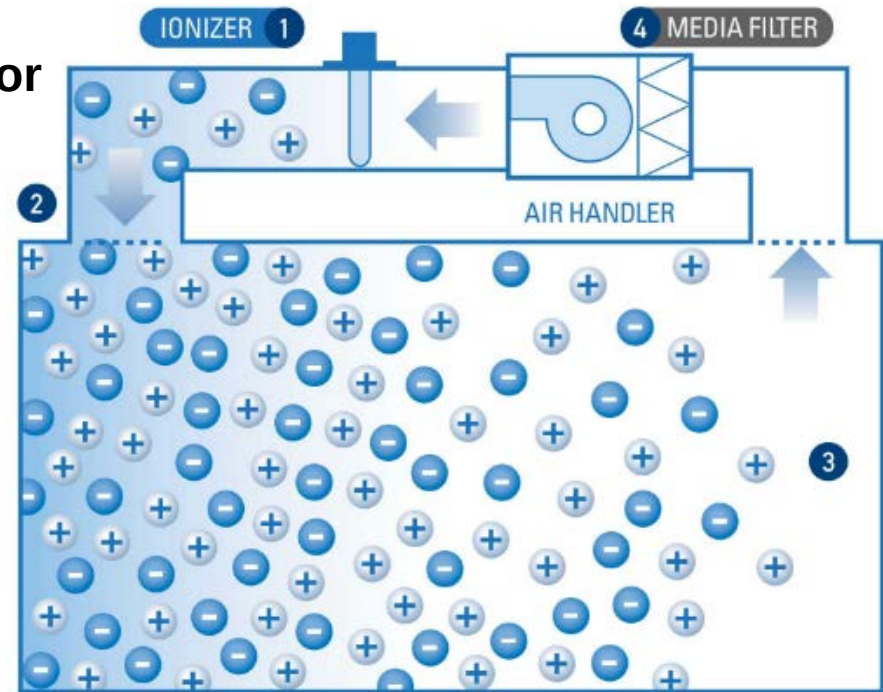


Image: ASHRAE

Recent peer-reviewed studies:

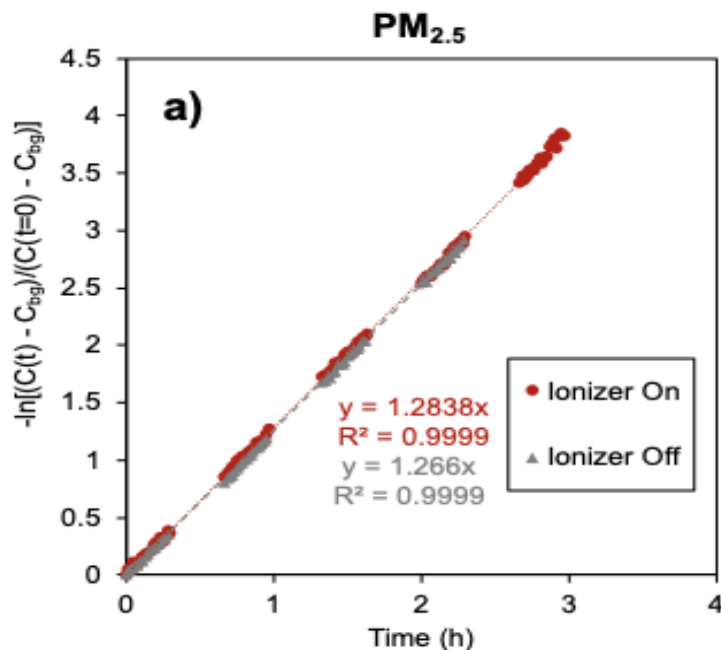
- Negative ions ($\sim 60,000 / \text{cm}^3$) decreased $\text{PM}_{2.5}$, did not reduce markers of oxidative stress. “...downsides do not support the use of [negative ions] as a health-based mitigation strategy ...”³
- Ionizer use in Beijing classrooms ($\sim 13,000 / \text{cm}^3$) may have positively impacted respiratory health at the expense of negative effects on cardiac health.⁴

¹<https://www.ashrae.org/technical-resources/filtration-disinfection>; ²ASHRAE, 2018. ASHRAE Position Document on Filtration and Air Cleaning; ³Liu et al. 2020 Indoor Air, <https://doi.org/10.1111/ina.12728> ⁴Dong et al. 2019, Env. Poll. 254: 113054

Effects in the space

Particles: little to no effect on loss rates

VOCs: some removed, some formed



Loss rates (1/h):

	Ionizer off	Ionizer on
PM2.5	1.27	1.28
SMPS (10-150 nm)	1.31	1.45
OPS (0.3 – 10 μm)	1.16	1.13

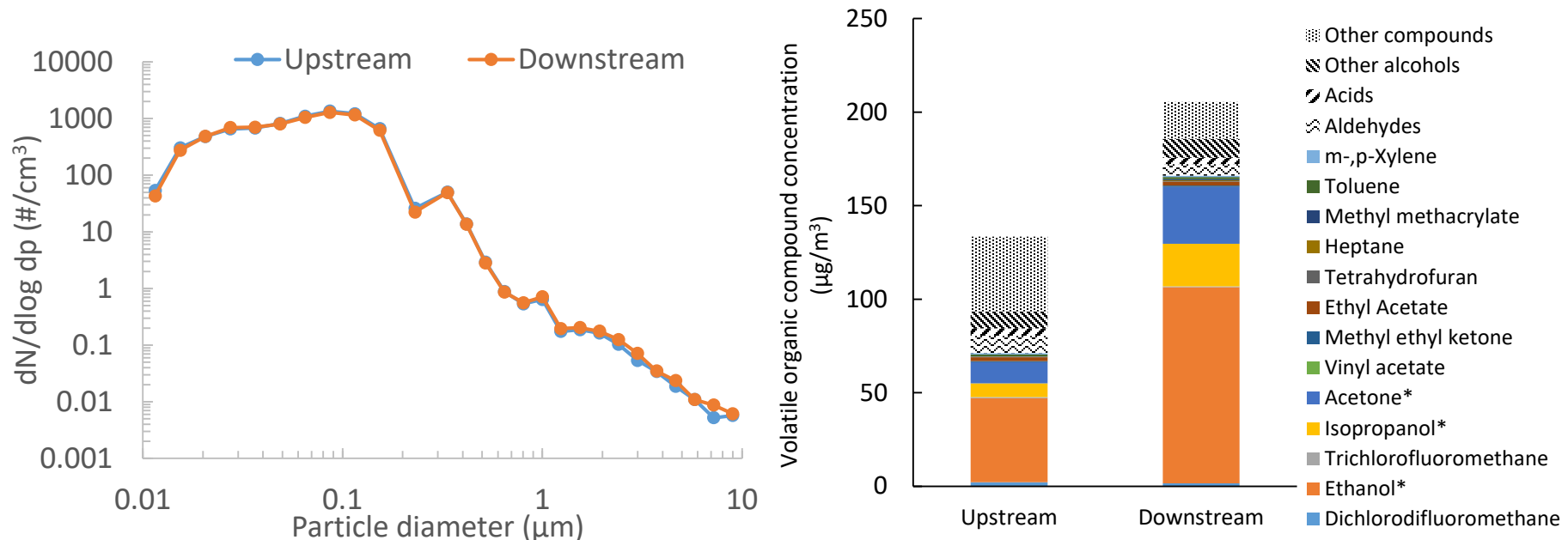
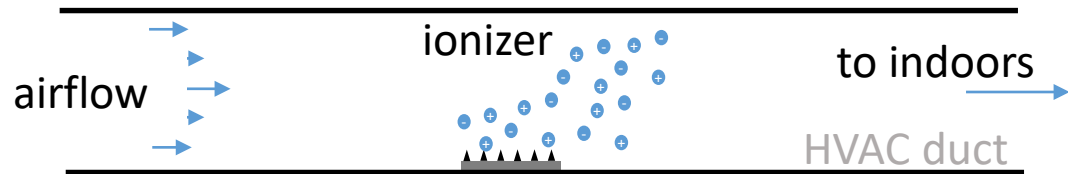
Analyte	% Change in I/O Ratio ¹
Formaldehyde	+2%
Acetaldehyde	+13%
Acetone	+73%
Butyraldehyde	+28%
Toluene	+15%
1,2-Dichloroethane	< -42%
Ethylbenzene	< -64%
m,p-Xylene	< -78%
Dichlorodifluoromethane	< -17%
Summed TOC ²	-19%

¹ values with < indicate measured value inside the chamber was below MDL

Experiments and results by IIT, led by Dr. Brent Stephens w/ his team

Possible claims:

- ↑ removal by filtration if ionizer upstream filter
- ↑ particle deposition in duct



**extrapolated beyond calibration curve*

Field measurement results:

- Measurements in office building with NPBI operating, averaged over 1 h
- Ozone levels were similar upstream and downstream
- No evidence of agglomeration, VOC byproducts generated